

Measurement of Substrate Thermal Resistance using DNA Denaturation Temperature

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Abstract

Heat Transfer and Thermal Management have become important aspects of the developing field of μTAS systems particularly in the application of the the μTAS philosophy to thermally driven analysis techniques such as PCR. Due to the development of flowing PCR thermocyclers in the field of μTAS , the authors have previously developed a melting curve analysis technique that is compatible with these flowing PCR thermocyclers. In this approach a linear temperature gradient is induced along a sample carrying microchannel. Any flow passing through the microchannel is subject to linear heating. Fluorescent monitoring of DNA in the flow results in the generation of DNA melting curve plots. This works presents an experimental technique where DNA melting curve analysis is used to measure the thermal resistance of microchannel substrates. DNA in solution is tested at a number of different ramp rates and the different apparent denaturation temperatures measured are used to infer the thermal resistance of the microchannel substrates. The apparent variation in denaturation temperature is found to be linearly proportional to flow ramp rate. Providing knowledge of the microchannel diameter and a non-varying cross-section in the direction of heat flux the thermal resistance measurement technique is independent of knowledge of substrate dimensions, contact surface quality and substrate composition/material properties. In this approach to microchannel DNA melting curve analysis the difference between the measured and actual denaturation temperatures is proportional to the substrate thermal resistance and the ramp-rate seen by the sample. Therefore quantitative knowledge of the substrate thermal resistance is required when using this technique to measure accurately DNA denaturation temperature .

Key words: PCR, Fluorescent Melting Curve Analysis, Denaturation of DNA, Genetic Analysis, Thermal Resistance, Microchannel, Substrate

1 Introduction

Fluorescent Melting Curve Analysis (FMCA) is a technique through which non-specific fluorescent DNA dyes can be used to recognise an individual DNA product (1). The thermodynamic stability of double helix DNA (dsDNA) is contingent on both its length and its base pair composition (2; 3); samples of differing length and compositions may be recognised by their differing denaturation temperatures, t_m . FMCA uses the characteristic variations in sample fluorescence with temperature to identify the denaturation temperature of samples. With increasing temperature the fluorescence of dsDNA in the presence of a dsDNA dye will decrease linearly. However at higher temperatures on the PCR thermocycle — typically between $85^\circ C$ and $95^\circ C$ — the dsDNA denatures and a non-linear decrease in fluorescence will be observed. The measurement of the temperature at which this non-linear change in fluorescence occurs permits measurement of the denaturation temperature of PCR product.

FMCA has become an almost ubiquitous feature of quantitative PCR thermocyclers (4). It is most commonly used as a closed tube technique to ensure the specificity and quality of the PCR sample amplified. The melting curve method can also be used to detect single nucleotide polymorphisms (SNPs) (5) and mutations associated with genetic diseases such as cancer (6). When combined with appropriate primer/probe selection melting curve analysis becomes an important tool in multiplex PCR (7).

Since the Micro Total Analysis System (μTAS) concept was first proposed (8) a number of technologies have undergone miniaturisation from desktop to ‘lab on a chip’. Among these technologies have been the thermocyclers associated with the PCR process. Research on well-based PCR thermocyclers has focused on reducing reaction size (9). Space-domain (flowing) PCR — where the reaction is moved between zones of fixed temperature (10; 11) — has been of interest to many research groups due to its potential for rapid thermocycling, small reaction volumes and high-throughput processing.

Progress towards developing FMCA for use with flowing thermocyclers is ongoing (12). The application of spatial temperature gradients across micro-fluidic substrates has been previously used for DNA melting analysis in microchannel systems (13; 14; 15; 16; 17). Spatial temperature gradients have

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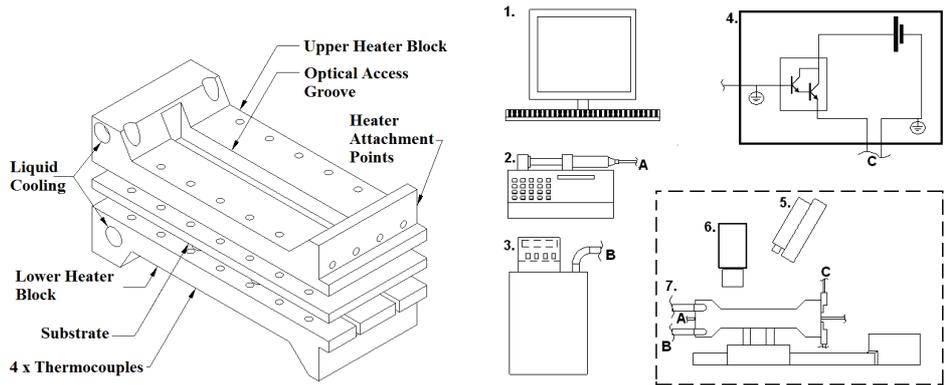
35 also been used to acquire DNA melting curves under non-equilibrium condi-
36 tions (18). Denaturation temperature measurements made from surface bound
37 DNA have previously been used to characterise microfluidic substrates (19).

38 The use of low conductivity materials in the manufacture of microchannel
39 substrates is considered an advantage for many applications (20). For lab-on-
40 chip space-domain PCR fluidic length scales are small which results in rapid
41 fluid temperature transition between thermal zones (21). Thermal gradients
42 are present within the substrates during these transitions but are negligible
43 within the thermal zones themselves. However, in order to integrate melting
44 curve analysis in space-domain thermocyclers the fluid must be subject to a
45 steady heat flux through the substrate walls. Thermal gradients will be present
46 through the substrate at all points of the melting curve analysis resulting in
47 an apparent temperature difference between melting curves measured in a
48 microchannel compared with those measured in a reference platform. In this
49 approach knowledge of substrate thermal resistance is important as the magni-
50 tude of the apparent temperature discrepancy is proportional to the substrate
51 thermal resistance. Much work has been done towards characterising heat
52 transfer in microchannels (22; 23) particularly for applications in electronic
53 cooling although thermal characterisation of microfluidic substrates for appli-
54 cations in biomedical research has become increasingly important (24; 25; 26).

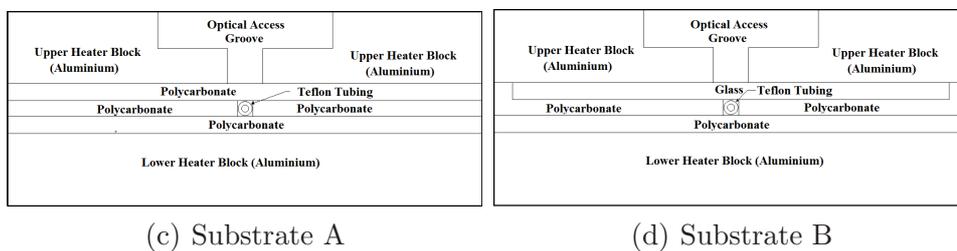
55 In this paper melting curve analysis is performed by pumping DNA in solution
56 along a microfluidic channel. The test platform consists of two thermal blocks
57 which sandwich a microfluidic channel. This channel is transparent Teflon
58 FEP tubing with an internal diameter of $400\mu m$. During experimental testing
59 the channel is aligned within the test platform using two different substrates.
60 Substrate A is composed of polycarbonate while Substrate B is composed of
61 a polycarbonate base with a glass cover-sheet. DNA melting curves are mea-
62 sured in each substrate using a variety of flow velocities and the effects are
63 recorded. These findings are supported using a simple heat transfer analy-
64 sis. Results are then substituted into the theoretical treatment to infer the
65 thermal resistance between the thermal blocks and the fluid flow within the
66 microchannels. With the caveats that the microchannel be of known uniform
67 cross-section and that the composition of the substrate be homogenous in the
68 direction of heat flux; this technique can measure the thermal resistance ir-
69 respective of knowledge of substrate dimensions, contact surface quality or
70 substrate composition/material properties.

71 **2 Experimental Design and Methods**

72 The test device consists of two aluminium blocks of length 120mm, width
73 50mm and minimum thickness 10mm. These thermal blocks are designed to



(a) Exploded view of heater block (b) Schematic of Equipment Setup configuration



(c) Substrate A

(d) Substrate B

Fig. 1. Equipment set up and Substrate compositions. Fig 1(b) shows 1. Computer for control/datalogging. 2. Syringe Pump. 3. Heater bath for cooling. 4. Control circuit for heaters. 5. Excitation Laser. 6. Monochrome Camera. 7. Rig mounted on Positioning Stage. A-A, B-B and C-C indicate connections for wiring and tubing.

74 sandwich a substrate which either has a microfluidic channel directly
 75 into it or — as in the case described here — aligns and secures disposable
 76 tubing within the test rig. Thick film heaters and Darlington transistors are
 77 attached to one end of the blocks for heating. The opposite end of the block
 78 has channels machined to provide for liquid cooling via a precision thermal
 79 bath. As illustrated in Figure 1(a), a slot of dimensions 84mm long and 2mm
 80 wide is machined into the upper block to provide optical access to the sub-
 81 strate. Four thermocouples are embedded in the lower thermal block such that
 82 they align with the optical access groove when the blocks are mated. The sam-
 83 ple is pumped through transparent, disposable Teflon FEP tubing (Upchurch
 84 Scientific). This tubing has an outer diameter of approximately $800\mu\text{m}$ and
 85 inner diameter of approximately $400\mu\text{m}$. Two different substrates are used to
 86 align the Teflon FEP tubing within the device.

87 Substrate A is manufactured from a 1mm thick transparent polycarbonate
 88 sheeting. This is cut to 120mm x 50mm, the footprint of the test rig ther-
 89 mal blocks, and is assembled as shown in Figure 1(c). The two lower ply
 90 are sprayed black and are glued together. The cover-sheet ply is transparent
 91 and is held in place when the blocks are assembled. Substrate B, shown in

92 Figure 1(d), consists of a similar polycarbonate base plate of near identical
93 dimensions but is fitted with a cover-sheet manufactured from borosilicate
94 glass. This cover-plate is composed of two microscope 1mm thick, 26mm wide
95 and 80mm long. One slide is located directly over the point in the optical ac-
96 cess groove where DNA denaturation is expected to occur. The second slide is
97 located upstream. Due to the orientation of the excitation laser, at the point
98 where the two microscope slides meet total internal reflection results in some
99 component of laser illumination being reflected directly at the emission sen-
100 sor, causing noise. Additionally, this total internal reflection also shadows part
101 of the channel from excitation light resulting in a drop in fluorescent signal
102 in this part of the optical access groove. The effects of this may be seen at
103 $\approx 80^\circ C$ upon the melt curves presented in Figure 2, where following moving
104 average smoothing it appears as a wavelike ripple. As this effect occurs away
105 from the denaturation section of the melting curves it has no effect upon the
106 denaturation temperatures measured.

107 Samples are loaded into the Teflon FEP tubing via aspiration and are buffered
108 by an immiscible fluid (M5904 Mineral Oil - Sigma-Aldrich). Samples are slugs
109 approximately 100mm long, which is greater than the interrogation length
110 (length of optical window) of the device. As the recirculation length for the
111 sample (27) is significantly greater than the interrogation length it is assumed
112 that multiphase effects are trivial and the flow approximates single-phase
113 Hagen-Poiseuille flow. The sample is pumped into the device until it is observed
114 entering the interrogation area and a fluorescent signal is acquired.

115 Fluorescent illumination was provided using a 488nm blue laser (BlueSkyRe-
116 search). The laser beam was passed through a concave lens (Edmundoptics) to
117 expand and diffuse the excitation beam. Emissions were filtered via a 515nm
118 long pass filter (Melles-Griot) and the emission sensor was a monochrome CCD
119 camera (The Imaging Source). Fluorescent emissions are proportional to the
120 intensity of excitation light (28). In order to ensure uniform illumination of
121 the interrogation area throughout individual experiments the heating blocks
122 are mounted on a positioning stage. As the sample flows down the microfluidic
123 channel at a given velocity the test device is moved at a velocity of identical
124 magnitude but opposite direction. The sample therefore remains within the
125 interrogation area and stationary relative to the excitation light source and
126 emission sensor. Sample pumping is via a syringe pump (Harvard Apparatus
127 PHD2000). The experimental setup is illustrated in Figure 1(b).

128 On activation of data acquisition the appropriate flow rates and slew rates
129 are calculated automatically (based on the ramp rate selected by the user)
130 for the syringe pump and positioning stage respectively. In some cases the
131 sample was aspirated back through the device and re-tested. The experi-
132 mental conditions used during testing are outlined in Table 1. For convenience
133 the spatial temperature gradient applied across the blocks will be referred to

Table 1
Experimental Test Conditions.

Ramp Rate	Temp. Diff	Thermal Grad	Velocity
$^{\circ}C/s$	$^{\circ}C$	$^{\circ}C/mm$	mm/s
0.000	24	0.286	0.000
0.033	24	0.286	0.116
0.033	8	0.095	0.350
0.100	24	0.286	0.350
0.100	8	0.095	1.050
0.300	24	0.286	1.050
0.300	8	0.095	3.150

134 in terms of the temperature difference between both ends of the optical ac-
 135 cess groove, ΔT_{OAG} . This experimental rig was previously characterised using
 136 substrates of a similar composition but larger internal diameter (29) using in
 137 channel thermocouple measurements. Temperature measurements were made
 138 at 0.1mm intervals using a positioning stage with the position of the thermo-
 139 couple bead being confirmed using a CCD camera. A line of best fit passed
 140 through the temperature field for the optical access groove for the worst per-
 141 forming substrate — manufactured completely from polycarbonate — resulted
 142 in a coefficient of determination of $R^2 = 0.995$. The temperatures measured
 143 by the thermocouples at each end of the optical access groove typically varied
 144 by less than $0.2^{\circ}C$ during the course of an experiment.

145 The denaturation temperature, t_m , for each DNA melting curve was identified
 146 by applying the analysis technique previously described (17).

147 3 Biological Material

148 Two individual PCR plasmid fragments were amplified from E. Coli plasmid
 149 vector pGEM-5Zf(+) (Promega). A 240bp plasmid fragment was amplified
 150 using a forward primer of sequence 5'- AGG GTT TTC CCA GTC ACG ACG
 151 TT-3' and a reverse primer 5'-CAG GAA ACA GCT ATG ACC-3'. A 232bp
 152 plasmid fragment was amplified using forward primer 5'-ATA CCT GTC CGC
 153 CTT TCT CC-3' and reverse primer 5'-CCT CGC TCT GCT AAT CCT
 154 GT-3'. Primers were acquired from MWG Biotech (Ebersberg, Germany).
 155 Fragments were amplified using LightCycler Faststart DNA Master SYBR
 156 Green I reaction mix (Roche). Fragments were thermocycled and a melting
 157 curve analysis performed using the Applied Biosystems AB7900 thermocycler;
 158 fragments were then frozen at $-20^{\circ}C$ prior to use.

159 **4 Theoretical Treatment**

160 During experimental testing a temperature field is applied to the upper and
 161 lower surfaces of the substrates using thermal blocks. Heater elements and
 162 fluidic cooling are used to maintain the temperatures of the end-points of
 163 these blocks. Assuming simple 1D conduction heat transfer this temperature
 164 field is a linear thermal gradient in the direction of heat flux. The microfluidic
 165 channel is on the same axis as heat flux through the thermal blocks. Due to
 166 the thermal resistances of both the working fluid and the tubing/substrate
 167 walls a temperature difference will exist between the mean temperature of the
 168 working fluid and the temperature at the outer surface of the polycarbonate
 169 substrate.

170 As a simplification the experimental platform is treated as cylindrical tube
 171 of uniform cross-section with a linear temperature gradient applied along the
 172 outer wall. In this configuration single-phase fluid flowing through the channel
 173 is subject to a uniform wall heat flux. Therefore the fluid is subject to heating
 174 at a constant ramp-rate.

175 The heat energy required to increase the internal energy of the fluid by the
 176 appropriate ramp rate (30) is:

$$q_w = -c_w \rho_w V_w \frac{dT}{dt} \quad (1)$$

177 where the fluid is subjected to ramp rate $\frac{dT}{dt}$. c refers to the specific heat
 178 capacity of the fluid, ρ its density and V is the volume of the fluid.

179 The ramp rate seen by the fluid is a function of the spatial thermal gradient
 180 and the velocity of the fluid through the test rig:

$$\frac{dT}{dt} = \frac{dT}{dx} \frac{dx}{dt} \quad (2)$$

181 From Equation 1 the heat energy q_w must enter the fluid element of volume V_w
 182 to increase the temperature of the fluid at a ramp rate of $\frac{dT}{dt}$. In order to enter
 183 the fluid heat energy equal to q_w must pass through the walls of the tubing.
 184 This heat transfer must be driven by a temperature difference between the
 185 inner and outer walls of the tubing. The magnitude of this heat transfer is:

$$q_T = -\frac{k_T A}{x_T} (\Delta T_T) \quad (3)$$

186 where x_T is the thickness of the material through which the heat is passing.
 187 Equating Equation 1 and Equation 3, re-writing in cylindrical co-ordinates of
 188 infinitesimal axial dimension dy , and rearranging for the temperature differ-
 189 ence through the tubing walls ΔT_T :

$$\Delta T_T = -\left(\frac{dT}{dt}\right)\left(\frac{x_T}{k_T(2\pi R\delta y)}\right)\left(c_w\rho_w\pi R^2\delta y\right) \quad (4)$$

190 From (31), Equation 5 presents the temperature variation across the fluid
 191 within the microchannel, $\Delta T_F(r)$, from the wall temperature based upon the
 192 application of a uniform wall heat flux to the walls of the fluid element:

$$\Delta T_F(r) = -\frac{dT}{dt}\left(1 + 0.5\left(1 - \frac{r^2}{R^2}\right)\right)\frac{(R^2 - r^2)}{4\alpha_w} \quad (5)$$

193 where r the position on the radial axis measured from the centre-line, R the ra-
 194 dius of the circular micro-channel and α the thermal diffusivity of the working
 195 fluid $\frac{k_w}{\rho_w c_w}$.

196 Assuming constant material properties, dimensions and ramp rate $\frac{dT}{dt}$, Equa-
 197 tion 4 states a constant temperature difference, ΔT_T , will exist between the
 198 inner and outer walls of the substrate. It is assumed that a negligible thermal
 199 gradient exists across the thermal blocks due to their much greater thermal
 200 conductivity. In addition, applying typical values of water as the working fluid
 201 and applying them to Equation 5 for the maximum ramp-rate used in this
 202 project, $\frac{dT}{dt} = 0.9^\circ C$, results in a temperature variation at the the microchan-
 203 nel centerline, $\Delta T_F(0)$, of less than $0.1K$. Therefore only the temperature
 204 variation through the substrates, ΔT_T , will be considered.

205 From Equation 4, the temperature difference across the microchannel walls,
 206 ΔT_T , is a function of the ramp rate seen by the fluid, $(\frac{dT}{dt})$, the thermal resis-
 207 tance of the microchannel walls, $(\frac{x_T}{k_T(2\pi R\delta y)})$, and the thermal capacitance of
 208 the fluid, $(c_w\rho_w\pi R^2\delta y)$. Assuming that the thermal properties of the fluid, the
 209 cross-section of the microchannel and the thermal resistance of the microchan-
 210 nel walls remain constant; the temperature difference across the tubing walls
 211 , ΔT_T , will be proportional to the ramp rate seen by the fluid, $\frac{dT}{dt}$.

212 By making measurements at different ramp rates, $\frac{dT}{dt}$, it is possible to generate
 213 a linear relationship between temperature difference and ramp rate, ΔT_T vs
 214 $\frac{dT}{dt}$. Rearranging Equation 4 this relationship can be used to measure the
 215 thermal resistance of the tubing:

$$\left(\frac{x_T}{k_T(2\pi R\delta y)}\right) = -\frac{\Delta T_T}{\left(\frac{dT}{dt}\right)(c_w\rho_w\pi R^2\delta y)} = -\frac{\text{slope of line}}{(c_w\rho_w\pi R^2\delta y)} \quad (6)$$

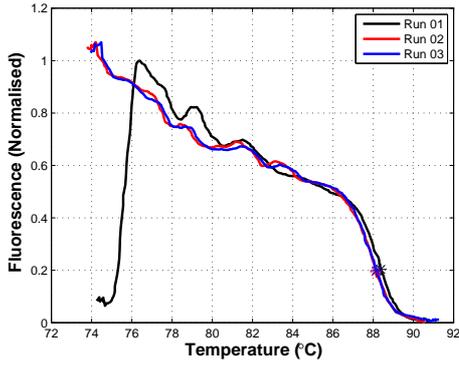
216 This equation is applied to experimental results in order to measure the ther-
 217 mal resistance of a substrate. Assuming the thermal resistance is non-variant
 218 in the direction of fluid flow the equation requires no information regarding the
 219 physical composition of the tubing, its dimensions or its thermal properties.

220 5 Results

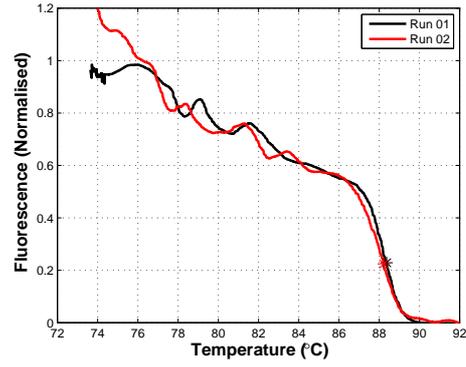
221 The experimental test conditions applied are outlined in Table 1. These were
 222 selected to investigate the effect of varying the flow velocity, varying the
 223 the spatial temperature gradient and varying the test ramp-rates during the
 224 course of experimental testing. With $\Delta T_{OAG} = 24.0^\circ C$, measurements at
 225 $\frac{dT}{dt} = 0.0^\circ C/s$ and $\frac{dT}{dt} = 0.1^\circ C/s$ typically resulted in ≈ 900 fluorescent mea-
 226 surements being made during the traverse of the DNA sample across the opti-
 227 cal access window. For the case of $\frac{dT}{dt} = 0.033^\circ C/s \approx 2700$ measurements are
 228 made; for $\frac{dT}{dt} = 0.3^\circ C/s \approx 300$ measurements are made and for $\frac{dT}{dt} = 0.9^\circ C/s$
 229 ≈ 100 measurements are made. For experiments conducted with a tempera-
 230 ture difference of $\Delta T_{OAG} = 8.0^\circ C$ the number of measurements for a given
 231 ramp-rate are typically reduced by a factor of 3 as these tests occur at higher
 232 velocities.

233 Figure 2 shows melting curves acquired from the 232bp plasmid fragment.
 234 These measurements was made by applying a temperature difference $\Delta T_{OAG} =$
 235 $24^\circ C$ to the outer walls of Substrate B. In this case the DNA sample was
 236 recycled through the test rig and retested under the same operating conditions.
 237 The melting profiles measured show good repeatability. A variation of less than
 238 $0.3^\circ C$ in measured denaturation temperature is observed except in the case
 239 of $\frac{dT}{dt} = 0.9^\circ C/s$ ($\Delta T_m = 0.6^\circ C$ (Fig 2(e))). This increased variation is a
 240 function of both the higher velocity at which the sample is tested and the
 241 lower resolution of melting profiles acquired at higher ramp-rates.

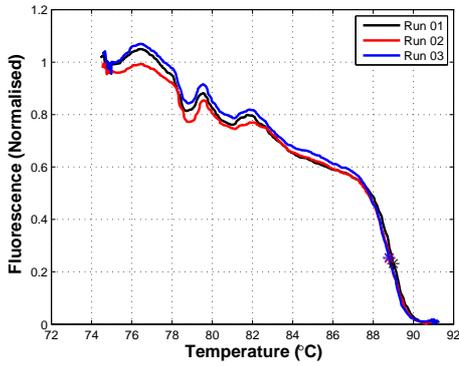
242 For melting profiles acquired at $\frac{dT}{dt} = 0.0^\circ C/s$ the DNA slug was positioned
 243 such that it filled the optical access groove. The fluorescent acquisition sys-
 244 tem then traversed the stationary DNA sample to generate melting curves
 245 (Fig 2(a)). The sharp transition observed at $75^\circ C$ in Figure 2(a) is a result of
 246 the transition from measurement of oil to measurement of DNA sample. As
 247 described previously the ripples which may be seen in the linear portion of
 248 the melting profiles — typically between $78^\circ C$ and $82^\circ C$ in every plot — are a
 249 result of an optical discrepancy in the glass cover-sheet used with Substrate B.



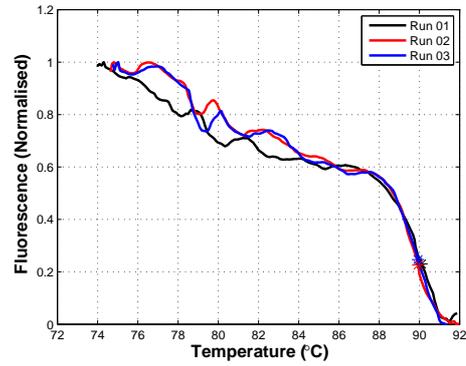
(a) $\frac{dT}{dt} = 0.0^\circ\text{C}/s$



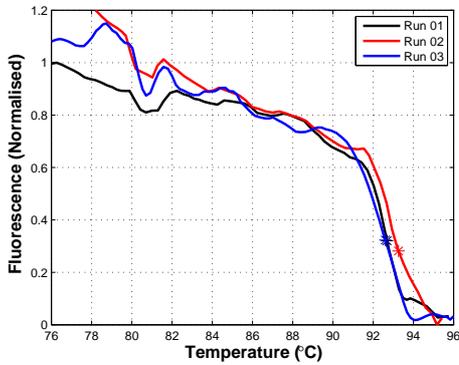
(b) $\frac{dT}{dt} = 0.033^\circ\text{C}/s$



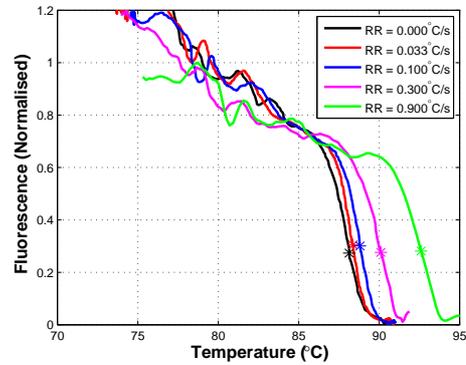
(c) $\frac{dT}{dt} = 0.1^\circ\text{C}/s$



(d) $\frac{dT}{dt} = 0.3^\circ\text{C}/s$



(e) $\frac{dT}{dt} = 0.9^\circ\text{C}/s$



(f) Comparison of Different Ramp Rates

Fig. 2. Plasmid Fragment B (232bp) Melting Curves measured across a temperature difference of $\Delta T = 24.0^\circ\text{C}$ using Substrate B.

250 Figure 2(f) presents a comparison of typical melting profiles acquired at differ-
 251 ent ramp-rates. It is clear that increasing the ramp-rate at which the melting
 252 profiles are acquired has the effect of offsetting apparent denaturation tem-
 253 perature to a higher temperature. As described previously in the theoretical
 254 treatment this effect is a function of both the thermal resistance and the

255 ramp-rate seen by the fluid. Figure 3 presents data acquired from a number
 256 of different experiments at test conditions outlined in Table 1. Each dataset
 257 is normalised by subtracting the mean denaturation temperature measured
 258 at $\frac{dT}{dt} = 0.0^\circ\text{C}/s$ from each measured denaturation temperature. Plotting the
 259 temperature difference measured against the ramp-rate confirms the linear
 260 relationship expected from the theoretical treatment.

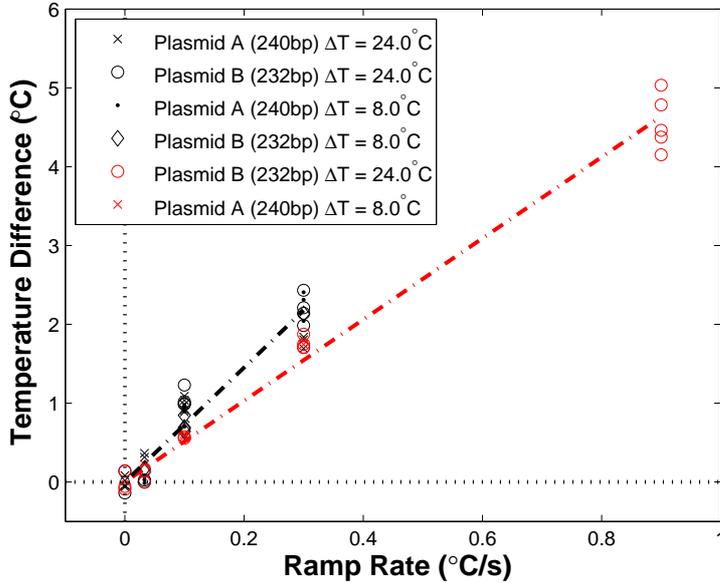


Fig. 3. Comparison of the Change in Apparent Denaturation Temperature, t_m , with respect to ramp rate between measurements made in Substrate A and Substrate B. Measurements from Substrate A are shown in black, while those from Substrate B are shown in red.

261 Figure 3 also shows that this trend is independent of DNA fragment used.
 262 Under the same test conditions, both the 232bp and 240bp plasmid fragments
 263 are subject to near identical changes in apparent denaturation temperature.
 264 In addition, melting profiles measured at identical ramp-rates but different
 265 operating conditions show similar apparent denaturation temperatures. Fi-
 266 nally, it is clear that varying the composition of the alignment substrate re-
 267 sults in a different magnitude of temperature variation. This is a result of the
 268 higher thermal conductivity of borosilicate glass — the cover-sheet used in
 269 Substrate B — compared with polycarbonate used in the cover-sheet of Sub-
 270 strate A. Under the test conditions applied no useable results were acquired
 271 from $\frac{dT}{dt} = 0.9^\circ\text{C}/s$ in Substrate A as the DNA samples were not fully de-
 272 naturated when leaving the optical access window. Some of the melting curves
 273 used to generate the denaturation temperature data shown in Figure 3 for
 274 Substrate A have previously been published (17).

275 Substituting the slopes of the lines of best fit shown in Figure 3 into Equa-
 276 tion 6 permits estimates of the substrate thermal resistances to be made. These

277 results are presented in Table 2.

Table 2

Thermal Resistance Measurements made using Melting Analysis

Method of	Equation	R^2	Thermal Res. (x/kA) KW^{-1}
Substrate A	$y = 7.257x$	0.9270	7.396
Substrate B	$y = 5.151x$	0.9866	4.985

278 6 Discussion and Conclusions

279 Thick walled, low conductivity plastic tubing has been used with success in
280 flowing PCR thermocyclers (32; 33). In the PCR case the flow transitions be-
281 tween 2 or 3 isothermal zones. Past the entry length the heat flux required
282 to maintain fluid temperature is minimal. Hence the temperature distribution
283 through thick walled tubing is minimal; the fluid and tubing walls may be con-
284 sidered isothermal. It has previously been described that a primary advantage
285 of using glass as a working material for $\mu T A S$ applications is its low thermal
286 conductivity (20). However in the case of space-domain melting curve analysis
287 a constant heat flux is required to heat the fluid throughout the analysis. The
288 tubing walls/substrates act as an insulating layer between the heater blocks
289 and the sample, resulting in a temperature disparity between the outer walls
290 of the substrate and the actual fluid temperature.

291 The magnitude of this temperature disparity is proportional to the ramp-rate
292 seen by the fluid and to the thermal resistance between the fluid and the heater
293 blocks. The ramp-rate is a product of the spatial temperature gradient and
294 the velocity of the fluid flow. It is clear from the results presented here that the
295 temperature disparity is observable and will have an effect on denaturation
296 temperature measurements. An equation has been presented which describes
297 the temperature disparity for fluid in a simple tube and measure substrate
298 thermal resistance. This approach can be applied to any micro-channel sub-
299 strate to compensate for temperature disparity effects irrespective of knowl-
300 edge of substrate dimensions or composition; with the caveats that the spatial
301 temperature gradient applied to the walls of the substrate be uniform and
302 that the substrate be non-varying in the direction of heat flux.

303 It has also been shown that denaturation temperature measurements made at
304 different velocities/spatial temperature gradients but at identical ramp-rates
305 are comparable. This approach has a number of advantages. In the case of
306 an integrated $\mu T A S$ system — where flow rates may be dictated by other

307 factors — changing the spatial temperature gradient applied along the block
308 allows the ramp-rate to be selected. Decreasing the velocity of the fluid —
309 and hence the time the sample is in the test rig — can result in a greater
310 number of fluorescent $acq/^\circ C$. The number of fluorescent $acq/^\circ C$ made by
311 a high-resolution melting analysis platform is an important benchmark for
312 its performance, particularly in SNP analysis (4). In the work presented here
313 melting curves measured at $\frac{dT}{dt} = 0.033^\circ C$ across $\Delta T = 24^\circ C$ resulted in
314 approximately $100acq/^\circ C$.

315 Of particular note were measurements made at $\frac{dT}{dt} = 0.0^\circ C$. In this case the
316 velocity of the fluid was zero and the fluorescent acquisition system was tra-
317 versed across the sample. Using this method it is clear that ramp-rate ef-
318 fects are greatly attenuated. Unlike well-based systems (3) this approach will
319 not require any compensation for thermal capacitance effects. In addition the
320 number of fluorescent measurements made will be entirely a function of the
321 viability of the test sample, resulting in the potential for very high resolution
322 melting analysis.

323 It is clear from this study that the effects of thermal resistance must be com-
324 pensated for when trying to make cross-comparison of melting profiles ac-
325 quired using space-domain melting analysis at different ramp-rates; and indeed
326 in different substrates/microchannel profiles. However this study describes a
327 melting curve analysis based technique whereby the thermal resistance of sub-
328 strates can be measured. This approach is non-invasive and can be applied to
329 channel geometries where direct instrumentation — such as thermocouples —
330 may not be possible. dsDNA can be used as an alternative to other on chip
331 temperature sensing methods such as liquid crystal thermometry (34; 23) and
332 fluorescent dyes (35). With denaturation temperatures ranging from $\approx 60^\circ C$
333 to $\approx 93^\circ C$ depending primarily on DNA strand length and composition ds-
334 DNA has previously been used to measure temperatures in both well-based
335 PCR systems (36) and also from substrate surfaces (19).

336 References

- 337 [1] C. Wittwer, N. Kuskawa, Molecular Microbiology: Diagnostic Principles
338 and Practice, ASM Press, Washington DC, 2004, Ch. 6 Real Time PCR.
339 [2] C. Wittwer, M. Herrmann, A. Moss, R. Rasmussen, Continuous fluo-
340 rescence monitoring of rapid cycle DNA amplification, Biotechniques 22
341 (1997) 130–138.
342 [3] K. Ririe, R. Rasmussen, C. Wittwer, Product Differentiation by Analysis
343 of DNA Melting Curves during the Polymerase Chain Reaction, Analyt-
344 ical Biochemistry 245 (1997) 154–160.
345 [4] M. G. Herrmann, J. D. Durtschi, L. K. Bromley, C. T. Wittwer, K. V.
346 Voelkerding, Amplicon DNA Melting Analysis for Mutation Scanning and

- 347 Genotyping: Cross-Platform Comparison of Instruments and Dyes, *Clinical Chemistry* 52 (3) (2006) 494–503.
- 348
- 349 [5] C. Gundry, J. Vandersteen, G. Reed, R. Pryor, J. Chen, C. Wittwer, Amplicon Melting Analysis with Labeled Primers: A Closed-Tube Method for Differentiating Homozygotes and Heterozygotes, *Clinical Chemistry* 49 (3) (2003) 396–406.
- 350
- 351
- 352
- 353 [6] P. Bernard, C. Wittwer, Real Time PCR Technology for Cancer Diagnostics, *Clinical Chemistry* 48 (2002) 1178–1185.
- 354
- 355 [7] M. Hernandez, D. Rodriguez-Lazaro, T. Esteve, S. Prat, M. Pla, Development of melting temperature based SYBR Green I polymerase chain reaction methods for multiplex genetically modified organism detection, *Analytical Biochemistry* 323 (2003) 164–170.
- 356
- 357
- 358
- 359 [8] A. Manz, N. Graber, H. Widmer, Miniaturized total chemical analysis systems: A novel concept for chemical sensing, *Sensors and Actuators B: Chemical* 1 (1990) 244–248.
- 360
- 361
- 362 [9] H. Nagai, Y. Murakami, Y. Morita, K. Yokoyama, E. Tamiya, Development of A Microchamer Array for Picoliter PCR, *Analytical Chemistry* 73 (2001) 1043–1047.
- 363
- 364
- 365 [10] M. Kopp, A. de Mello, A. Manz, Chemical Amplification: Continuous-Flow PCR on a Chip, *Science* 280 (1998) 1046–1048.
- 366
- 367 [11] I. Schneegass, J. Kohler, Flow-through polymerase chain reactions in chip thermocyclers, *Reviews in Molecular Biotechnology* 82 (2001) 101–121.
- 368
- 369 [12] S. O. Sundberg, C. T. Wittwer, J. Greer, R. J. Prior, O. Elintoba-Johnson, B. K. Gale, Solution-phase DNA mutation scanning and SNP genotyping by nanoliter melting analysis, *Biomedical Microdevices* 9 (2) (2007) 159–166.
- 370
- 371
- 372
- 373 [13] H. Mao, M. A. Holden, M. You, P. S. Cremer, Reusable Platforms for High-Throughput On-Chip Temperature Gradient Assays, *Analytical Chemistry* 71 (2002) 5071–5075.
- 374
- 375
- 376 [14] T. M. Dalton, D. J. Kinahan, M. R. Davies, Fluorescent Melting Curve Analysis compatible with a Flowing Polymerase Chain Reactor, in: *Proceedings of 2005 ASME International Mechanical Engineering Congress and Exposition*, ASME, 2005.
- 377
- 378
- 379
- 380 [15] N. Crews, C. Wittwer, J. Montgomery, R. Pryor, B. Gale, Spatial DNA Melting Analysis for Genotyping and Variant Scanning, *Analytical Chemistry* 81 (6) (2009) 2053–2058.
- 381
- 382
- 383 [16] D. J. Kinahan, T. M. Dalton, M. R. Davies, ICNMM2008-62014 Microchannel Fluorescent Melting Curve Analysis, in: *Proceedings of 2008 Sixth International Conference on Nanochannels, Microchannels and Minichannels*, ASME, 2008.
- 384
- 385
- 386
- 387 [17] D. Kinahan, T. Dalton, M. Davies, Effect of substrate thermal resistance on space-domain microchannel fluorescent melting curve analysis, *Biomedical Microdevices* 11 (4) (2009) 747–754.
- 388
- 389
- 390 [18] P. Baaske, S. Duhr, D. Braun, Melting curve analysis in a snapshot, *Applied Physics Letters* 91 (2007) 133901.
- 391

- 392 [19] A. Dodge, G. Turcatti, I. Lawrence, N. F. de Rooij, E. Verpoorte, A
393 Microfluidic Platform Using Molecular Beacon-Based Temperature Cali-
394 bration for Thermal Dehybridization of Surface-Bound DNA, *Analytical*
395 *Chemistry* 76 (6) (2004) 1778–1787.
- 396 [20] B. Giordano, E. Copeland, J. Landers, Towards dynamic coating of glass
397 microchip chambers for amplifying DNA via the polymerase chain reac-
398 tion, *Electrophoresis* 22 (2) (2001) 334–340.
- 399 [21] K. Sun, A. Yamaguchi, Y. Ishida, S. Matsuo, H. Misawa, A heater-
400 integrated transparent microchannel chip for continuous-flow PCR, *Sen-*
401 *sors and Actuators: B. Chemical* 84 (2-3) (2002) 283–289.
- 402 [22] H. Zhang, D. Pinjala, T. Wong, K. Toh, Y. Joshi, Single-phase liquid
403 cooled microchannel heat sink for electronic packages, *Applied thermal*
404 *engineering* 25 (10) (2005) 1472–1487.
- 405 [23] R. Muwanga, I. Hassan, Local heat transfer measurements in microchan-
406 nels using liquid crystal thermography: methodology development and
407 validation, *Journal of Heat Transfer* 128 (2006) 617.
- 408 [24] D. Ross, M. Gaitan, L. Locascio, Temperature measurement in microflu-
409 idic systems using a temperature-dependent fluorescent dye, *Anal. Chem*
410 73 (17) (2001) 4117–4123.
- 411 [25] D. Ross, L. Locascio, Microfluidic temperature gradient focusing, *Anal.*
412 *Chem* 74 (11) (2002) 2556–2564.
- 413 [26] R. Fu, B. Xu, D. Li, Study of the temperature field in microchannels of
414 a PDMS chip with embedded local heater using temperature-dependent
415 fluorescent dye, *International Journal of Thermal Sciences* 45 (9) (2006)
416 841–847.
- 417 [27] M. N. Kashid, I. Gerlach, S. Goetz, J. Franzke, J. F. Acker, F. Platte,
418 D. W. Agar, S. Turek, Internal Circulation with the Liquid Slugs of a
419 Liquid-Liquid Slug-Flow Capillary Microreactor, *Ind. Eng. Chem. Res* 44
420 (2005) 5003–5010.
- 421 [28] P. Haugland, *Handbook of Fluorescent Probes and Research Products,*
422 *Molecular Probes, Oregon, 2002.*
- 423 [29] D. J. Kinahan, T. M. Dalton, M. R. Davies, ICNMM2008-62015 Ther-
424 mal Resistance Measurements from a Microchannel Fluorescent Melting
425 Curve Analysis Platform, in: *Proceedings of the 2008 6th International*
426 *Conference on Nanochannels, Microchannels and Minichannels, ASME,*
427 2008.
- 428 [30] J. P. Holman, *Heat Transfer, 8th Edition, McGraw Hill International*
429 *Editions, 1997.*
- 430 [31] A. Bejan, *Heat Transfer, John Wiley and Sons, 1993.*
- 431 [32] M. B. Sayers, T. M. Dalton, M. R. Davies, Real-Time Fluorescence Mon-
432 itoring of the Polymerase Chain Reaction in a Novel Continuous Flow
433 Reactor for Accurate DNA Quantification, in: *Proceedings of 4th Inter-*
434 *national Conference on Nanochannels, Microchannels and Minichannels,*
435 *ASME, 2006.*
- 436 [33] M. Chabert, K. D. Dorfman, P. de Cremoux, J. Roeraade, J.-L. Viovy,

- 437 Automated Microdroplet Platform for Sample Manipulation and Poly-
438 merase Chain Reaction, *Anal Chem* 78 (2006) 7722–7728.
- 439 [34] A. Chaudhari, T. Woudenberg, M. Albin, K. Goodson, Transient liq-
440 uid crystal thermometry of microfabricated PCR vesselarrays, *Microelec-
441 tromechanical Systems, Journal of* 7 (4) (1998) 345–355.
- 442 [35] J. Zhou, H. Yan, Y. Zheng, H. Wu, Highly Fluorescent Poly (dimethyl-
443 siloxane) for On-Chip Temperature Measurements, *Advanced Functional
444 Materials* 19 (2).
- 445 [36] C. Nellaker, U. Wallgren, H. Karlsson, Molecular beacon-based temper-
446 ature control and automated analyses for improved resolution of melting
447 temperature analysis using SYBR I green chemistry, *Clinical Chemistry*
448 53 (1) (2007) 98.