

N—H···N hydrogen bonding in 4,6-diphenyl-2-pyrimidinylamine isolated from the plant *Justicia secunda* (Acanthaceae)

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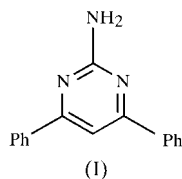
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The title compound, C₁₆H₁₃N₃, isolated from *Justicia secunda* (Acanthaceae), comprises two molecules (which differ slightly in conformation) in the asymmetric unit of space group *P* $\bar{1}$. Intermolecular N_{amino}—H···N_{pyrim} interactions (N_{pyrim} is a pyrimidine ring N atom) involve only one of the two donor amino H atoms and pyrimidine N atoms per molecule, forming dimeric units *via* R₂²(8) rings, with N···N distances of 3.058 (2) and 3.106 (3) Å, and N—H···N angles of 172.7 (18) and 175.8 (17)°. The dimers are linked by C—H··· π (arene) contacts, with an H···centroid distance of 2.77 Å and a C—H···centroid angle of 141°.

Comment

The title compound, 4,6-diphenyl-2-pyrimidinylamine, (I), was isolated as a natural product from the plant *Justicia secunda* (Acanthaceae) collected in Trinidad, West Indies. The parent structure, *viz.* 4,6-diphenyl-1,2-dihydropyrimidine, has been reported (Weis & Vishkautsan, 1985), but the coordinates are not in the Cambridge Structural Database (CSD; Allen, 2002) for comparison with (I) (CSD refcode DEZMEA for parent compound).



Compound (I) crystallizes in space group *P* $\bar{1}$ (No. 2) with two independent molecules, *A* and *B*, in the asymmetric unit, which differ slightly in conformation (Fig. 1). Bond lengths

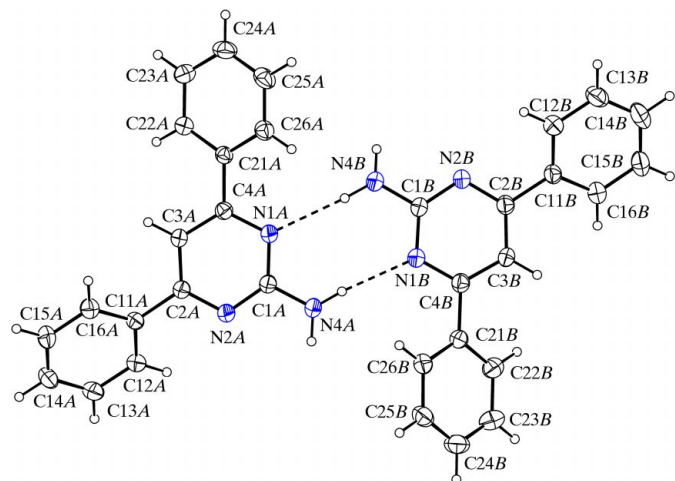
and angles are largely unexceptional and in accord with anticipated values (Orpen *et al.*, 1994); selected dimensions are given in Table 1. The central C—N bond lengths range from 1.340 (2) to 1.346 (2) Å, while the aromatic C—C distances are in the range 1.381 (3)–1.387 (3) Å. Torsion-angle differences are evident from analysis of the N1—C4—C21—C26 angles, which are –29.8 (3) and 36.5 (3)° in molecules *A* and *B*, respectively (Table 1). Even more dramatic is the difference in dihedral angles between the C11—C16 and C21—C26 planes, which are 52.15 (7) and 31.79 (7)° in molecules *A* and *B*, respectively. The weighted (unit weight) r.m.s. fit for the superposition of the non-H atoms in both molecules is 0.65 Å (0.55 Å) (Spek, 2003).

Molecules *A* and *B* associate in pairs through N_{amino}—H···N_{pyrim} intermolecular interactions (N_{pyrim} is a pyrimidine ring N atom), which only involve two of the four amino donor H atoms and N pyrimidine acceptors per *A/B* pair. The N···N distances are 3.058 (2) and 3.106 (3) Å, and the N—H···N angles are 172.7 (18) and 175.8 (17)° (Table 2), thus forming a hydrogen-bonded ring with an R₂²(8) graph set similar to that observed in centrosymmetric carboxylic acid pairs. A C—H··· π (arene) contact from C12A—H12A to the C21A—C26A ring links the dimeric units into a dimer of dimers, which is further augmented *via* a contact from C26A—H26A group to the N1A/N2A/C1A—C4A ring into a one-dimensional stack (Fig. 2 and Table 2). These are the only contacts of note in the structure of (I), apart from the N—H···N hydrogen bonding.

Examination of the structure of (I) for interactions involving N2A/N2B reveals that, along the *b* direction, atom H24A (a potential donor) is 2.99 Å from N2Aⁱ, with a C24A—H24A···N2Aⁱ angle of 163° [symmetry code: (i) *x*, 1 + *y*, *z*]. Likewise, the closest N···H intermolecular distance involving atom N2B is N2B···H24Bⁱ of 3.08 Å, with an N2B···H24Bⁱ—C24Bⁱ angle of 155°. Neither of these interactions constitutes a hydrogen bond.

Inspection of Fig. 2 shows the dimeric units in a two-dimensional cross-section through the crystal structure parallel to the [101] plane. In the crystal structure, a more efficient packing could involve dimer formation about inversion centres, which would necessitate rotation of the phenyl rings to align and fulfil symmetry requirements, *e.g.* the phenyl rings have interplanar angles of 52.12 (7) and 31.79 (7)° in molecules *A* and *B*, respectively. The reason why this does not arise is due to the initial hydrogen-bonded dimer formation. Utilization of only half of the N—H···N donor/acceptors per *A/B* hydrogen-bonded dimer prevents the N2A/N2B pair from forming hydrogen bonds, as potential (C)N—H···N donor/acceptor pairs cannot approach one another and because of the concurrent formation of unfavourable H···H contacts. Interactions in (I) are therefore limited to the two N—H···N interactions per *A/B* pair.

Ab initio calculations on (I) gave results similar to those observed in molecules *A* and *B*. Calculations were undertaken at the 6-31+G(*d*) level in the search for a local minimum for comparison with molecules *A* and *B*, using GAUSSIAN03 (Frisch *et al.*, 2003). The N—C—C—C torsion angles are 27.55°, similar to the value of –29.8 (3)° found in molecule *A*.


Figure 1

A view of the two independent molecules in (I) as a hydrogen-bonded dimer, with the atomic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii.

The $\text{H}_2\text{N}-\text{C}$ distance is 1.362 Å and the $\text{N}-\text{C}-\text{N}$ angle is 116.79° [range $116.15(17)$ – $117.05(17)^\circ$ in molecule A]. Overall, the calculated results resemble those of molecule A. However, no attempt was made to find the global minimum in the system.

Crystal structures with more than one molecule present in the asymmetric unit are not uncommon (Steiner, 2000; Gallagher *et al.*, 1998). Recently, Görbitz (2002, 2003) has reported on dipeptide derivatives, *e.g.* two forms of L-valyl-L-phenylalanine trihydrate ($P2_1$, $Z = 16$, $Z' = 8$; $P2_12_12_1$, $Z = 4$, $Z' = 1$) and L-methionyl-L-alanine ($P6_1$, $Z = 42$, $Z' = 7$), which provide rare examples of systems with $Z' = 8$ and 7 molecules in their respective asymmetric units, in contrast with $Z' = 2$, which is not uncommon. Analysis of the CSD (February 2004, version 5.25; Allen, 2002) reveals a total of 65 935 crystal structures in space group $P\bar{1}$ (No. 2) (and

equivalents with no restrictions), 7625 with $Z' > 1$, some 11.6% of the total, and 6775 with $Z' = 2$. No attempt was made to examine incorrect space groups.

Experimental

The plant *Justicia secunda* (Acanthaceae) was collected in Trinidad, West Indies, and dried at 308–313 K. A large quantity (1 kg) of this dried powdered plant (except root and flower/fruit) was placed in a large separating funnel and soaked in acetone overnight. The acetone layer was collected and the process repeated until the acetone layer became faintly green in colour. The combined acetone extracts were evaporated and the dry mass extracted with benzene. The benzene-soluble fraction was dried, dissolved in the minimum amount of dry warm CHCl_3 and spread over activated silica gel in a sintered glass Buchner funnel. It was then eluted with 50 ml portions of benzene and then a CHCl_3 –benzene mixture (1:4) under a slight vacuum. The fraction obtained from CHCl_3 –benzene (1:4) was collected and the solvent removed. It was then subjected to column chromatography with light petroleum (b.p. 313–333 K) and benzene. After concentration, the benzene fraction yielded gummy crystals upon refrigeration. These crystals were further purified by preparative chromatography to afford the pure title compound, (I), as well shaped crystals from CHCl_3 .

Crystal data

$\text{C}_{16}\text{H}_{13}\text{N}_3$
 $M_r = 247.29$
 Triclinic, $P\bar{1}$
 $a = 7.8263(9)$ Å
 $b = 10.8009(9)$ Å
 $c = 15.7878(12)$ Å
 $\alpha = 83.457(5)^\circ$
 $\beta = 77.039(7)^\circ$
 $\gamma = 82.326(7)^\circ$
 $V = 1284.0(2)$ Å³

$Z = 4$
 $D_x = 1.279$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 74 reflections
 $\theta = 5.4$ – 17.9°
 $\mu = 0.08$ mm⁻¹
 $T = 294(1)$ K
 Plate, colourless
 $0.50 \times 0.30 \times 0.08$ mm

Data collection

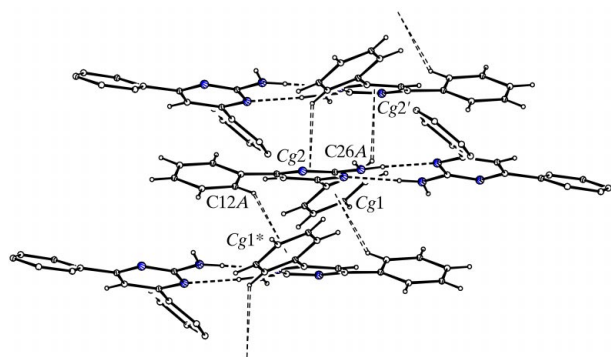
Siemens P4 diffractometer
 $\omega/2\theta$ scans
 5607 measured reflections
 4529 independent reflections
 2987 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.019$
 $\theta_{\text{max}} = 25.0^\circ$

$h = -9 \rightarrow 1$
 $k = -12 \rightarrow 12$
 $l = -18 \rightarrow 18$
 4 standard reflections
 every 296 reflections
 intensity decay: 1%

Table 1

Selected geometric parameters (Å, °).

N1A—C1A	1.345 (2)	N1B—C1B	1.346 (2)
N1A—C4A	1.346 (2)	N1B—C4B	1.343 (2)
N2A—C1A	1.343 (2)	N2B—C1B	1.343 (2)
N2A—C2A	1.340 (2)	N2B—C2B	1.340 (2)
N4A—C1A	1.357 (2)	N4B—C1B	1.347 (2)
C2A—C3A	1.385 (2)	C2B—C3B	1.381 (3)
C2A—C11A	1.488 (2)	C2B—C11B	1.488 (3)
C3A—C4A	1.384 (2)	C3B—C4B	1.387 (3)
C4A—C21A	1.481 (2)	C4B—C21B	1.484 (3)
N1A—C1A—N2A	126.76 (17)	N1B—C1B—N2B	126.72 (17)
N1A—C1A—N4A	117.05 (17)	N1B—C1B—N4B	116.73 (18)
N2A—C1A—N4A	116.15 (17)	N2B—C1B—N4B	116.55 (18)
N1A—C4A—C21A—C26A	−29.8 (3)	N1B—C4B—C21B—C26B	−36.5 (3)
C3A—C4A—C21A—C26A	151.83 (19)	C3B—C4B—C21B—C26B	142.9 (2)
C3A—C2A—C11A—C12A	149.4 (2)	C3B—C2B—C11B—C12B	−152.6 (2)


Figure 2

A view of the dimeric units in (I) linked by $\text{C}-\text{H} \cdots \pi(\text{arene})$ interactions. Atoms are depicted as small spheres of arbitrary sizes. The ten phenyl-ring H atoms in molecule B have been omitted for clarity. The centroid labels are as defined in Table 2, and the centroids labelled with an asterisk (*) or a prime (') are at symmetry positions (i) and (ii), respectively, given in Table 2.

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.047$
 $wR(F^2) = 0.119$
 $S = 1.01$
 4529 reflections
 359 parameters
 H atoms treated by a mixture of
 independent and constrained
 refinement

$$w = 1/[\sigma^2(F_o^2) + (0.055P)^2 + 0.13P]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} = 0.001$$

$$\Delta\rho_{\max} = 0.18 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.17 \text{ e } \text{\AA}^{-3}$$

Table 2

Hydrogen-bonding and contact geometry (\AA , $^\circ$) for compound (I).

Cg1 and Cg2 represent the centroids of the rings C21A–C26A and N1A/N2A/C1A–C4A, respectively.

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N4A–H1A \cdots N1B	0.89 (2)	2.17 (2)	3.058 (2)	172.7 (18)
N4B–H2B \cdots N1A	0.91 (2)	2.20 (2)	3.106 (3)	175.8 (17)
C12A–H12A \cdots Cg1 ⁱ	0.93	2.77	3.538 (2)	141
C26A–H26A \cdots Cg2 ⁱⁱ	0.93	2.78	3.393 (2)	124

Symmetry codes: (i) $-x, 2 - y, 1 - z$; (ii) $1 - x, 2 - y, 1 - z$.

Compound (I) crystallized in the triclinic system, the systematic absences permitting space group $P1$ or $P\bar{1}$; $P\bar{1}$ was assumed and confirmed by the analysis. All H atoms bonded to C atoms were treated as riding atoms, with C–H distances of 0.93 \AA , while the four amino H atoms were refined with isotropic displacement parameters, giving N–H distances in the range 0.89 (2)–0.91 (2) \AA .

Data collection: *XSCANS* (Siemens, 1996); cell refinement: *XSCANS*; data reduction: *XSCANS*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996) and *PLATON* (Spek, 2003);

software used to prepare material for publication: *NRCVAX96* (Gabe *et al.*, 1989), *SHELXL97* and *PREP8* (Ferguson, 1998).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: GD1292). Services for accessing these data are described at the back of the journal.

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