

Switchable Materials- The Route to Next Generation Multifunctional Analytical Platforms

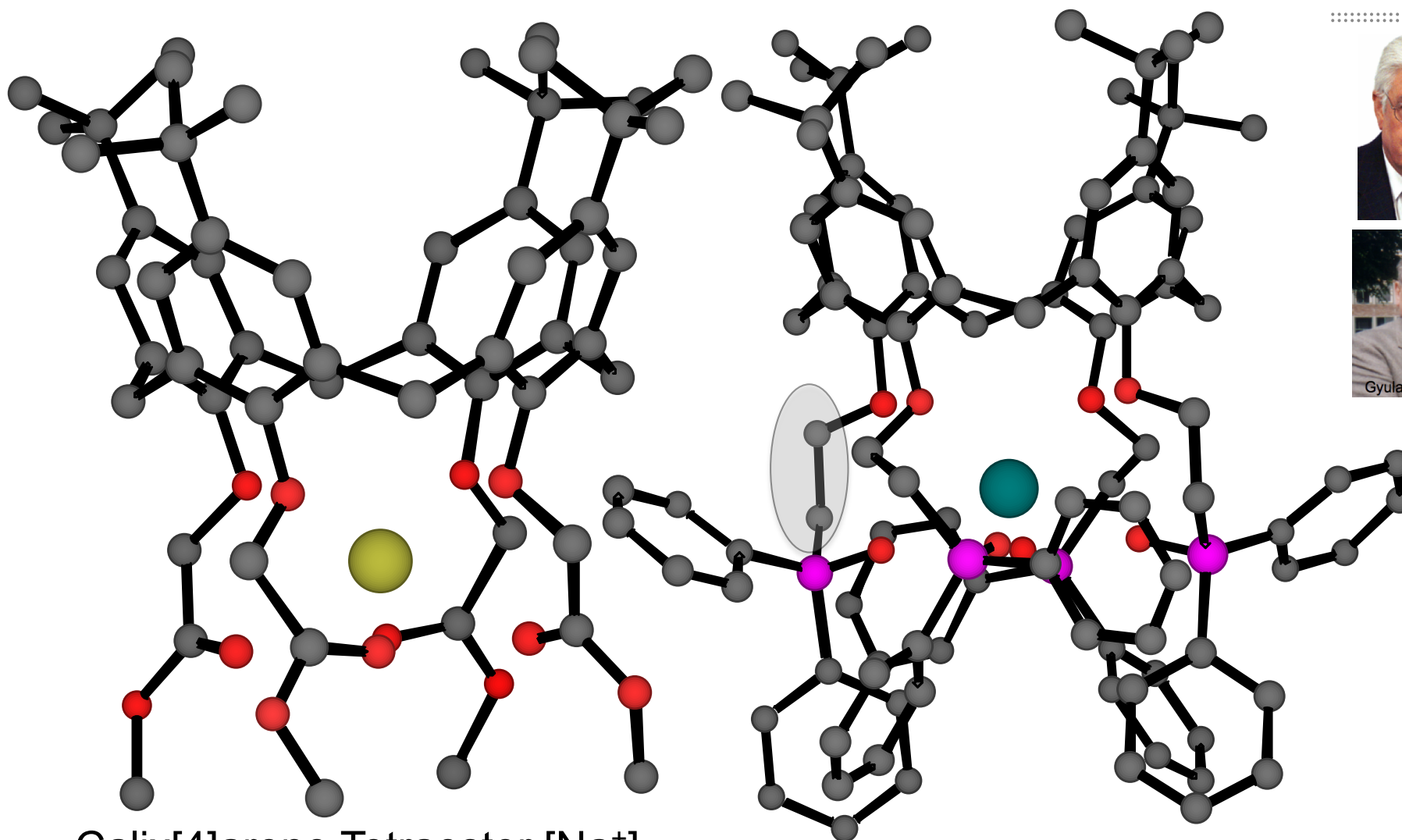
Dermot Diamond
CLARITY Centre for Sensor Web Technologies
National Centre for Sensor Research
Dublin City University

presented at

21st International Ion Chromatography Symposium (IICS 2009)
Malahide, Co. Dublin
September 10-11, 2009



Tetraester & Tetrphosphine Oxide

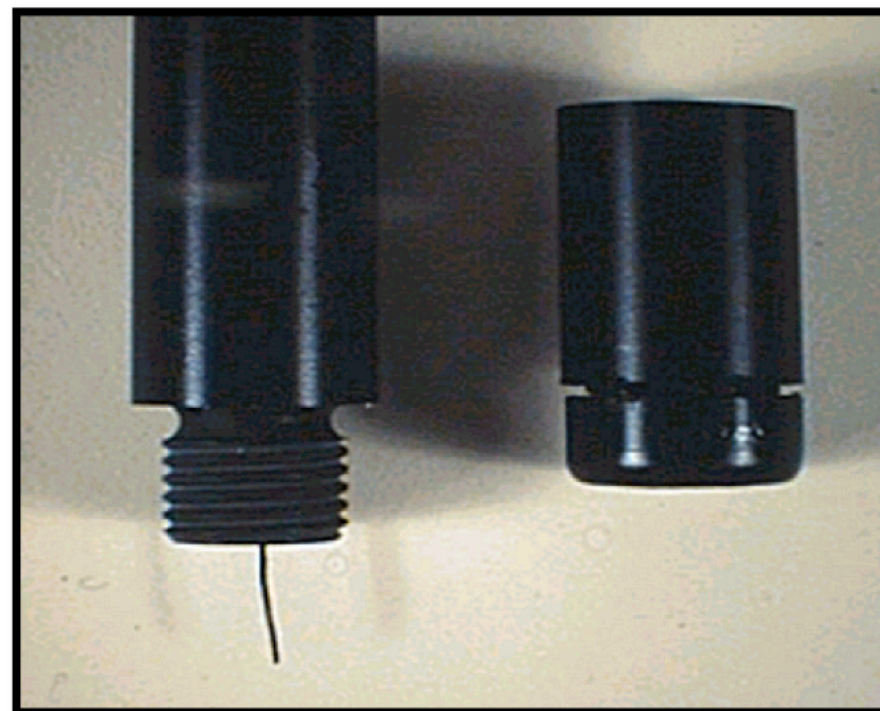
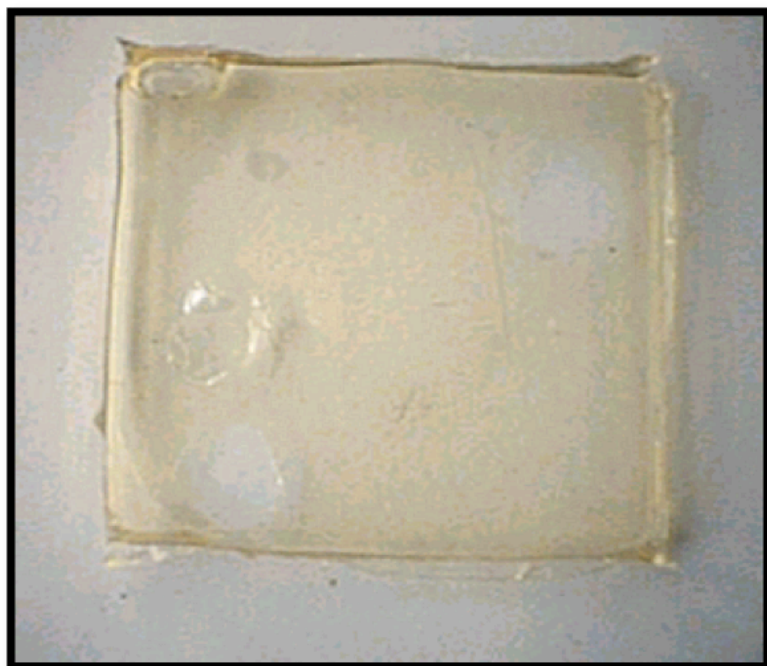


Calix[4]arene Tetraester [Na^+]

Calix[4]arene Tetrphosphine oxide [Ca^{2+}]

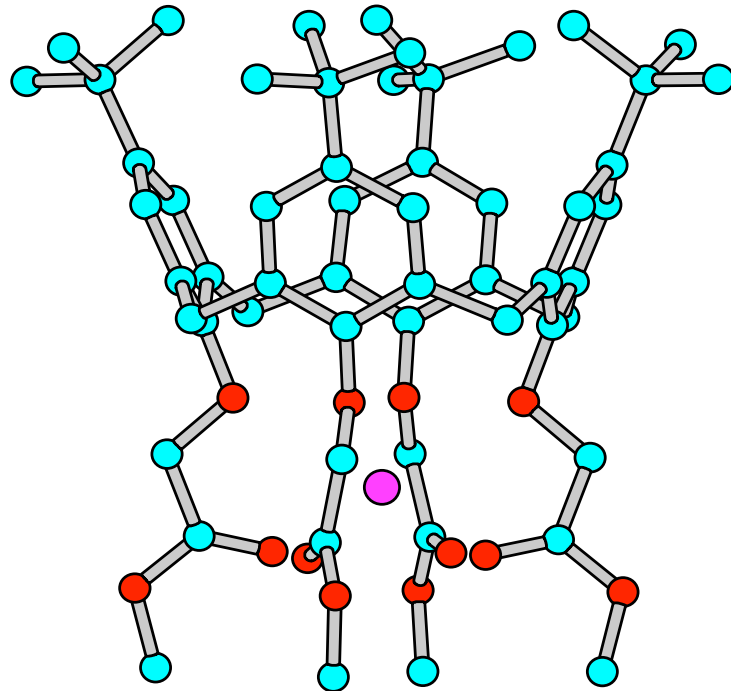


PVC - Membrane ISEs



Typical membrane cocktail (%w/w); PVC:33%, NPOE (plasticiser): 66%; ionophore/exchanger: 1% (ratio at least 2:1 by mole); dissolve in a volatile solvent e.g. THF and cast membrane from this solution

Molecular Functionality



TME:Na⁺ Side-on View

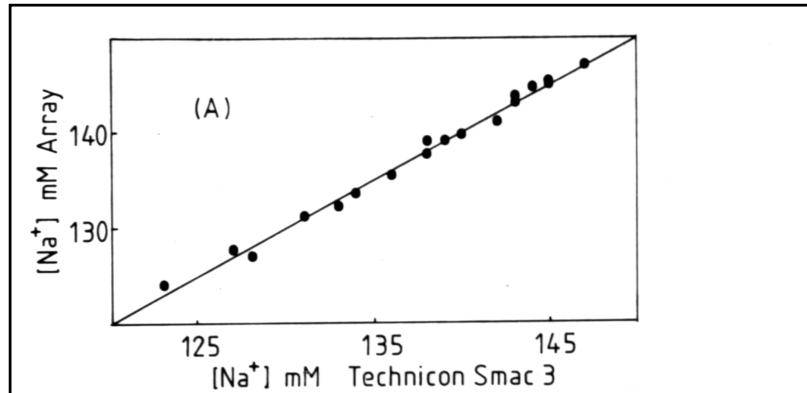
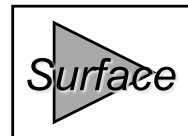
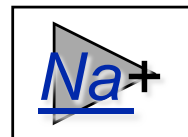
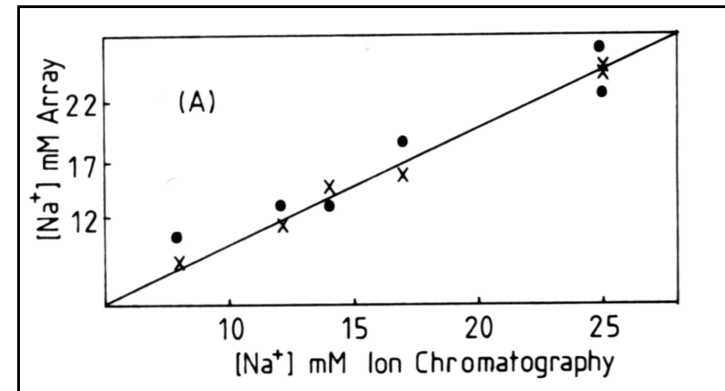


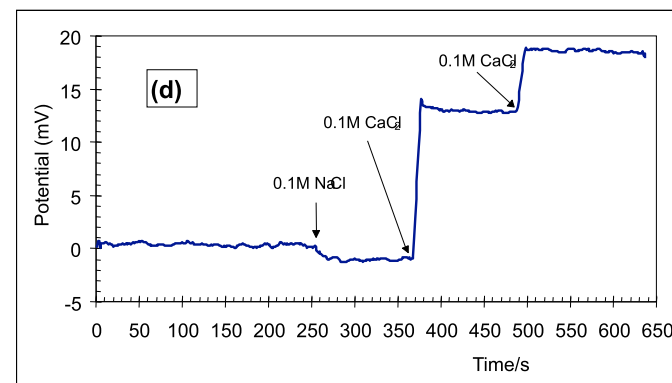
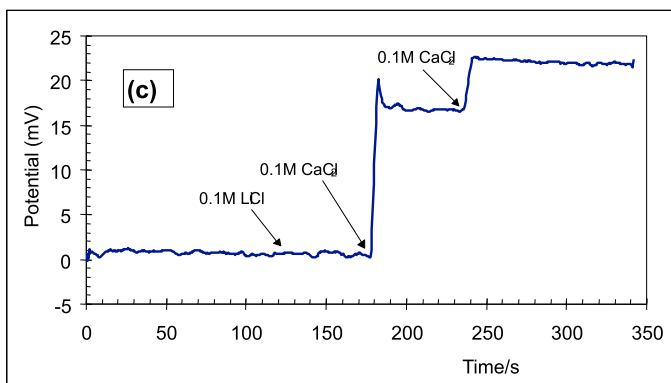
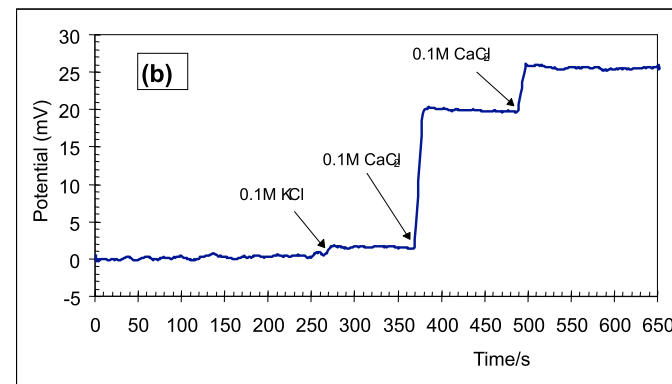
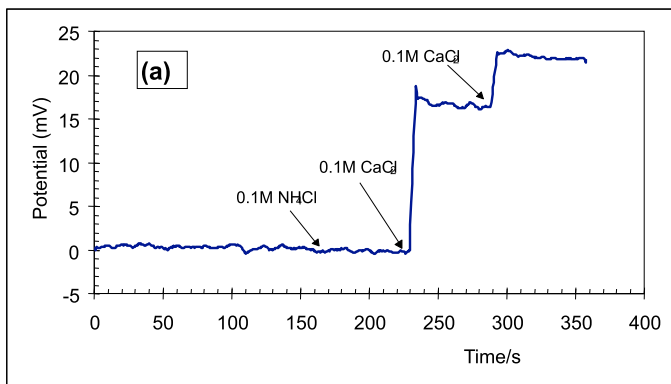
Fig. 3. Comparison of plasma sodium analysis using the array-FIA approach with a SMAC analyser. Good correlation without bias is obtained [5].



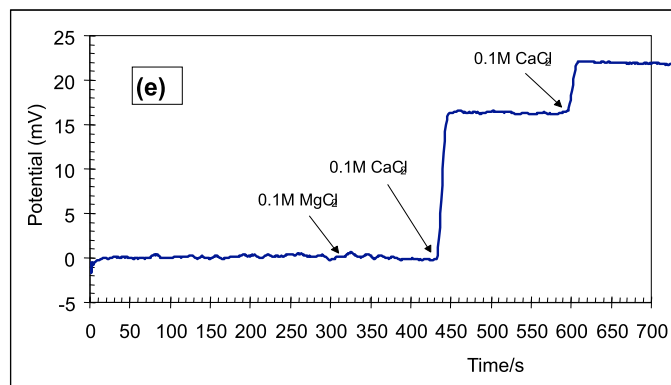
Anal. Chem., 64 (1992) 1721-1728.

Ligand (and variations of) are used in many clinical analysers for blood sodium profiling





response of TPOL electrode



Calcium-selective Electrode based on a Calix[4]arene Tetraphosphine Oxide, Tom McKittrick, Dermot Diamond, Debbie J. Marrs, Paul O'Hagan and M. Anthony McKervey, Talanta 43 (1996) 1145-1148.



Fame at last!!

The screenshot shows the Sigma-Aldrich website interface. At the top, there are logos for Sigma-Aldrich and Fluka. A navigation bar includes links for Log-In, Register, Tech Library, Product Search, MSDS Search, Order Center, Contact Us, Home, and Help. A yellow box on the right encourages registration for local pricing. The main content area displays product details for Sodium Ionophore X (Product Number: 71747). A sidebar on the left provides navigation options like Description/Pricing, Certificate of Analysis, MSDS, Options, and Go To. The product information section lists synonyms, molecular formula ($C_{60}H_{80}O_{12}$), molecular weight (993.29), CAS number (97600-39-0), purity grade (Selectophore®), and BRN (3587002). Literature references are cited, including a paper by D. Diamond et al. in Analyst 114, 1551 (1989). A red box highlights the reference text. The miscellaneous section notes its use as a potentiometric ion-selective electrode.

Sigma-Aldrich **Fluka** **REGISTER for LOCAL PRICING**

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Product Number: **71747**
Product Name: **Sodium Ionophore X**

Register or Login for Pricing
Click Here

Product Information

- ▶ [Description/Pricing](#)
- ▶ [Certificate of Analysis](#)
- ▶ [MSDS](#)

Options

- ▶ [Print Preview](#)
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- ▶ [Ask A Scientist](#)

Go To

- ▶ [Previous Page](#)
- ▶ [Sigma-Aldrich Home](#)

Synonyms: 4-tert-Butylcalix[4]arene-tetraacetic acid tetraethyl ester
Molecular Formula: $C_{60}H_{80}O_{12}$
Molecular Weight: 993.29
CAS: 97600-39-0
Purity Grade: Selectophore®
BRN: 3587002
R&S: S: 22-24/25

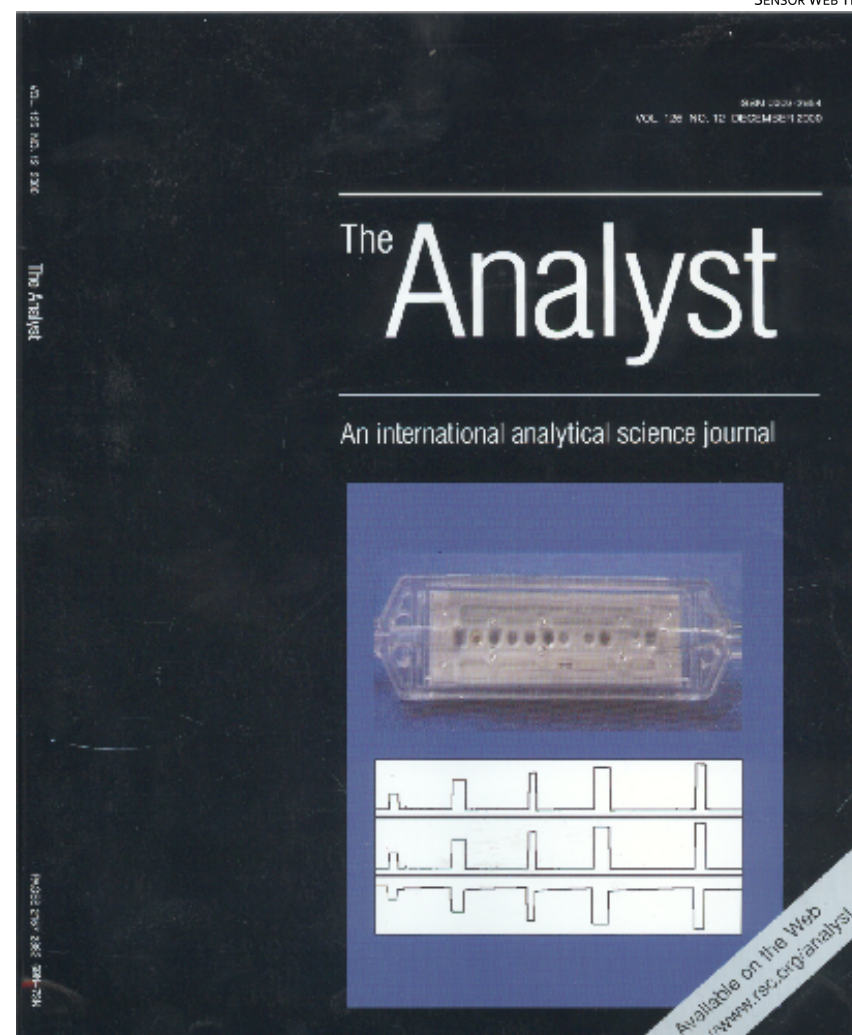
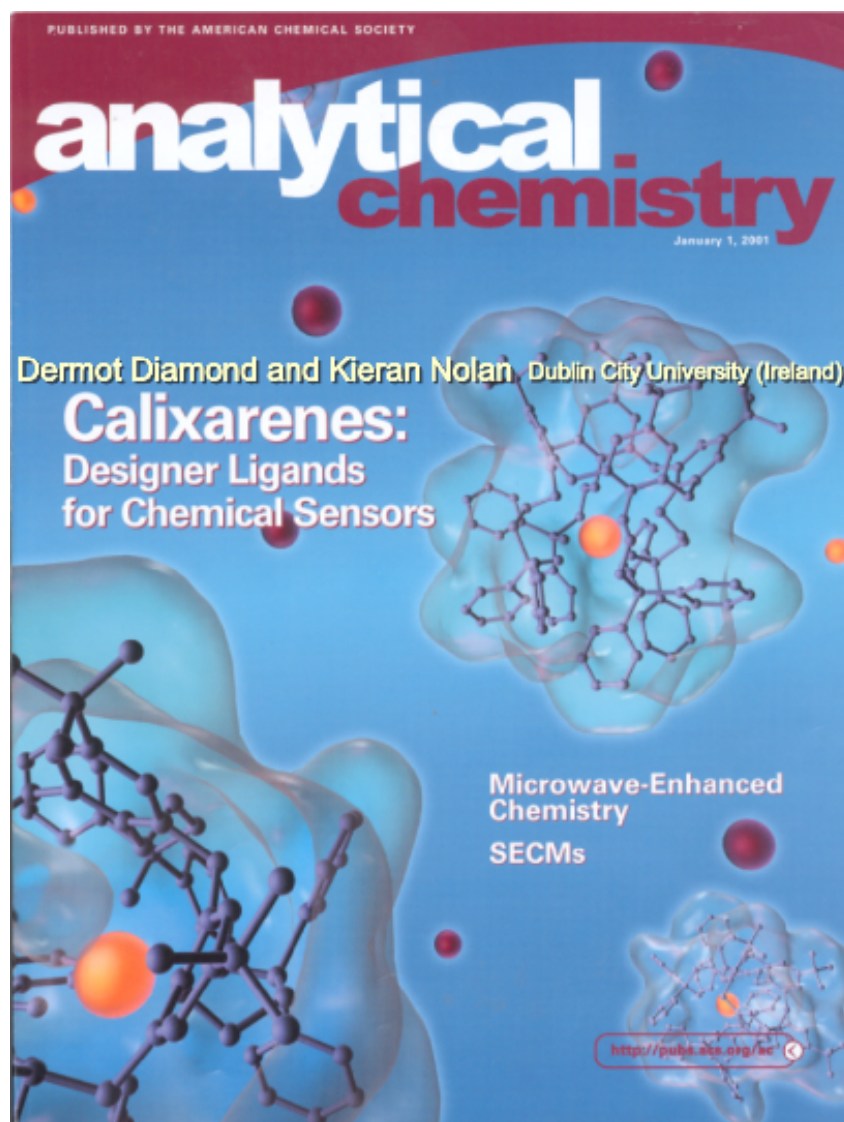
Literature References: Ionophore employed in solid-state and polymeric membrane sodium-selective electrodes: D. Diamond et al., Analyst 114, 1551 (1989); Anal. Chem. 64, 2496 (1992)

Miscellaneous: Ionophores (potentiometric) for ion-selective electrodes

◆ Best ionophore for sodium



Fame at Last!



Cover Article: Point-of-Need- Diagnosis of Cystic Fibrosis using a Potentiometric Ion-Selective Electrode Array

Aogan Lynch and Dermot Diamond, NCSR, Dublin City University, Ireland

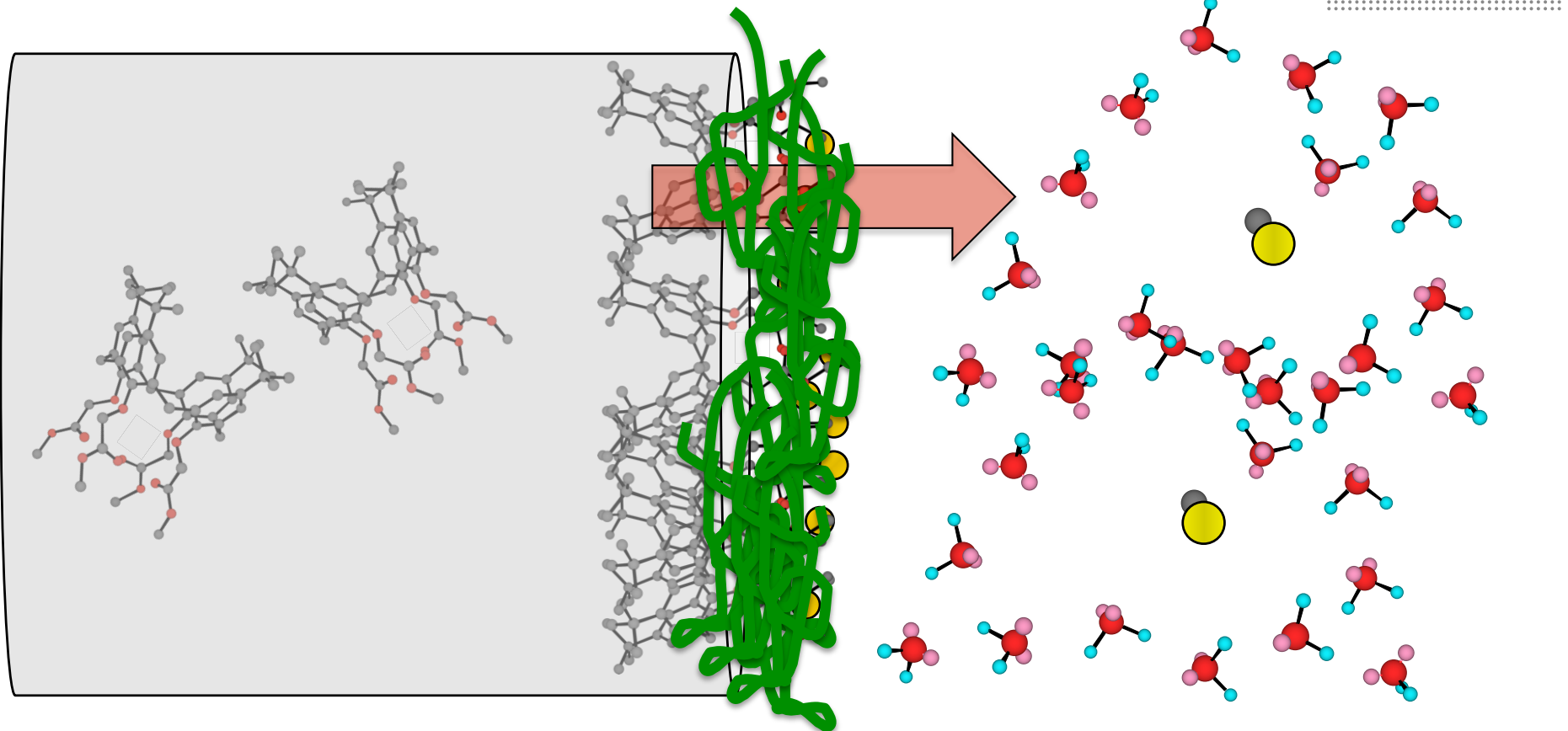


Keynote Article: August 2004, Analytical Chemistry (ACS)



Dermot Diamond, Anal. Chem., 76 (2004) 278A-286A.

Fundamental Problem: Sensor surface will change with time!

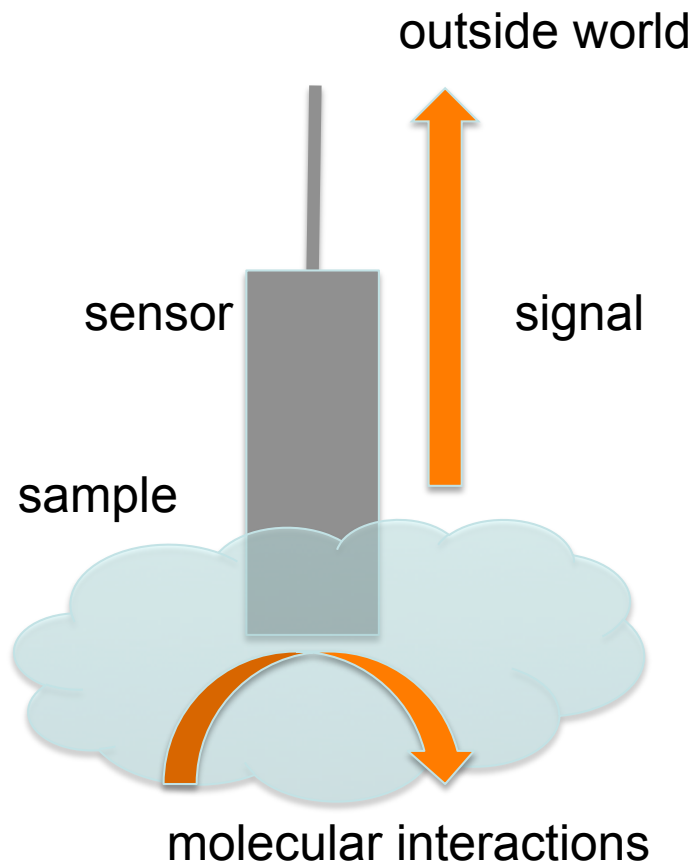


- Surface interactions are critical to signal generation - very susceptible to any process that modifies the surface condition => drift, loss of sensitivity => regular calibration => high cost of ownership
- Leaching, biofouling, physical damage, sample interferences,
- Engineers expect a thermistor, we have platforms closer to a washing machine!

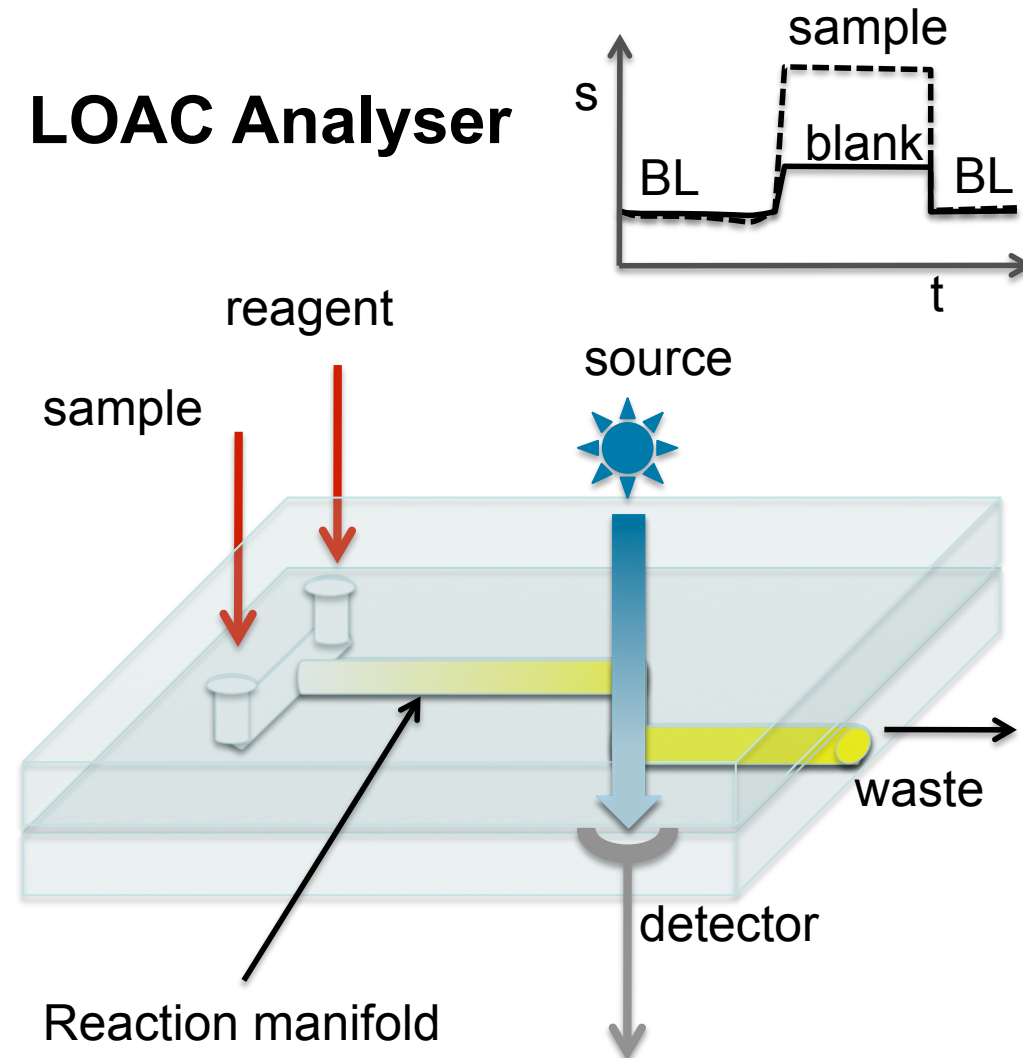


Direct Sensing vs. Reagent Based LOAC

Direct Sensing



LOAC Analyser

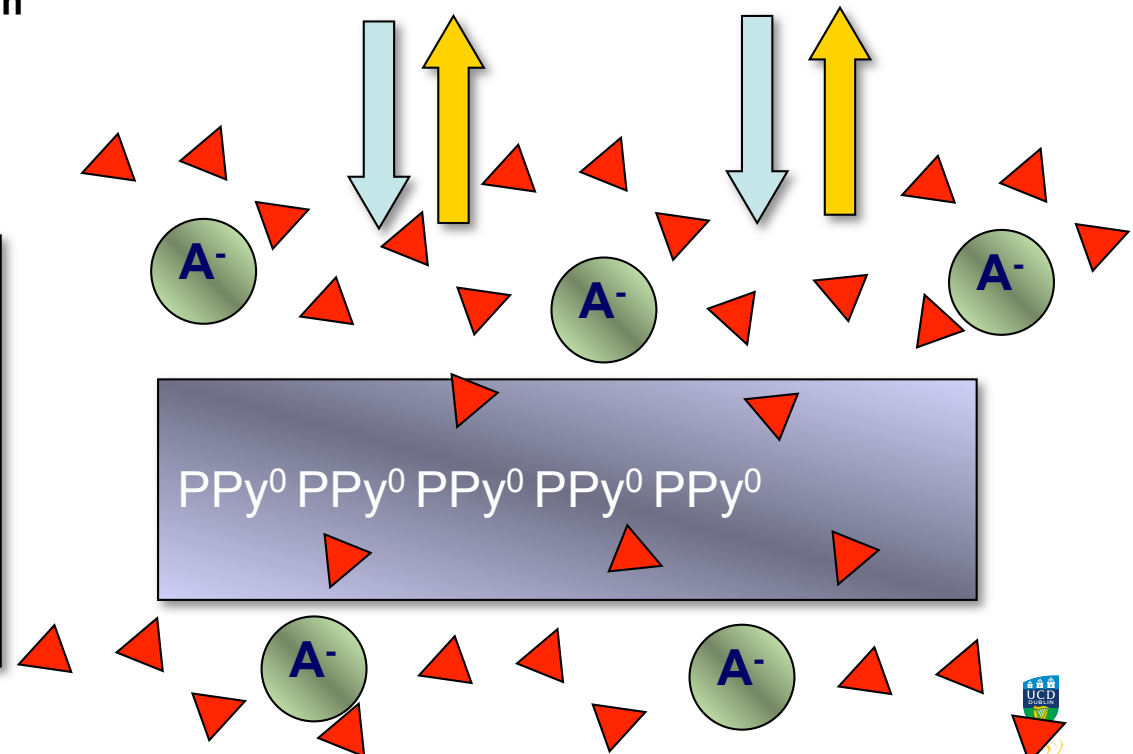
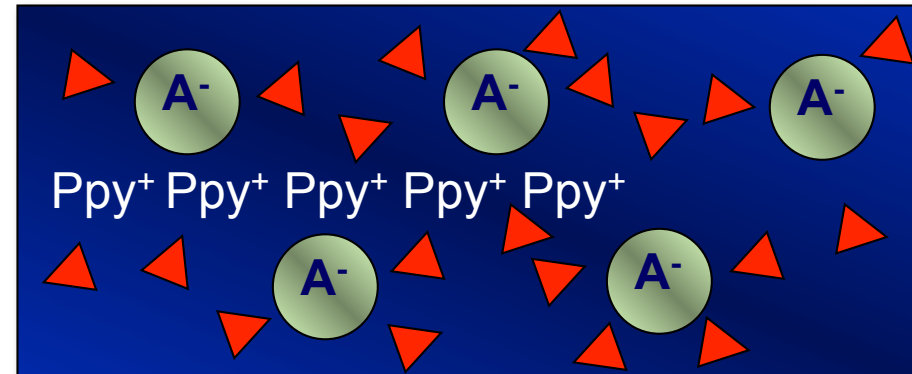
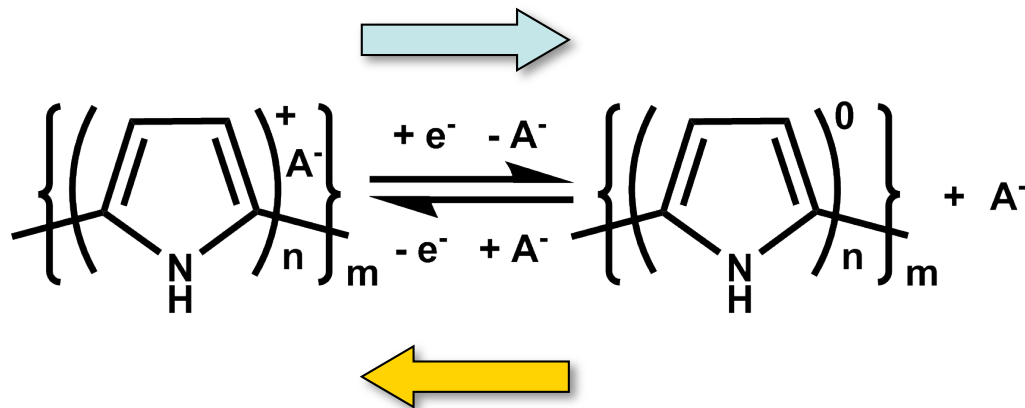


Adaptive (Stimulus-Responsive) Materials

- ◆ Materials that possess ‘multiple personalities’ or characteristics
- ◆ Can switch reversibly between these via a stimulus (chemical, electrochemical, photochemical.....)
- ◆ Properties change dramatically e.g. chemical binding behaviour, surface charge/polarity, porosity, permeability, dimensions, colour.....



Switchable Materials: Soft Polymer Actuators



Principle can be used to make soft polymer (biomimetic, artificial muscle) actuators such as **'benders'** (multi-laminated structures designed so that an outer PPY layer expands as the inner contracts to produce a bending movement)

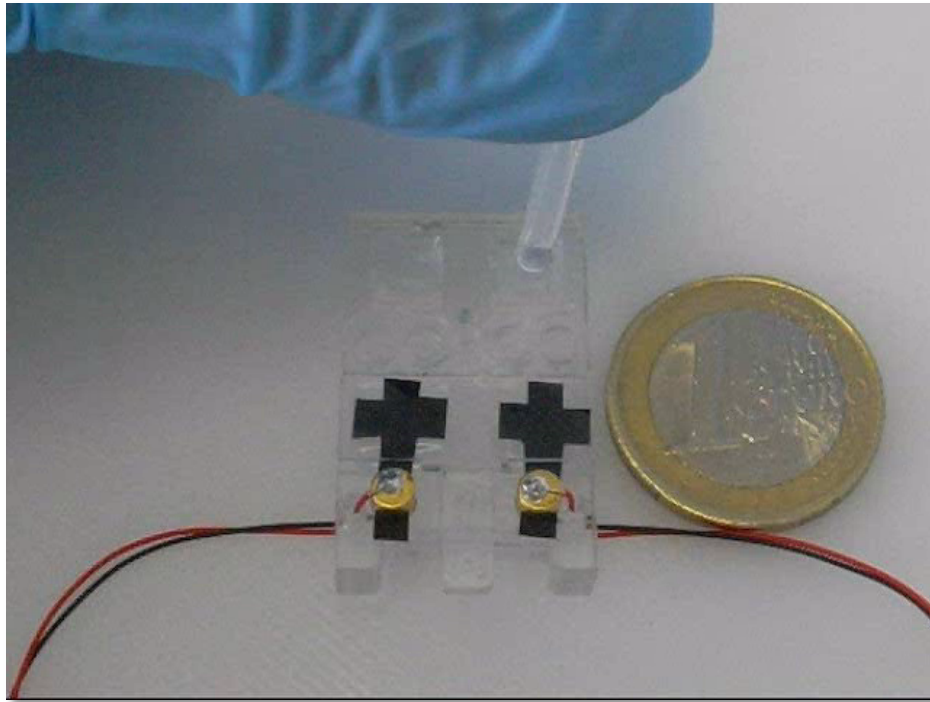
‘Artificial Muscle’ - Mobile Platforms

- Conducting polymer fibre bundles contained within polymer ‘skin’
- Mobile electrolyte solution
- Polarisation of the bundles causes movement of ions and associated water molecules due to charge compensation
- Causes swelling or contraction
- Effect can be translated into bending by laminating two oppositely polarised layers with a flexible porous inactive intermediate layer - soft pumps and valves!



Videos from Moshen Shahinpoor's
website at www.unm.edu/~amri

Polymer Micropumps and Valves



Daniel Kim and Kim Lau

- ◆ Low power, low cost components are vital for realisation of next generation micro-dimensioned analytical platforms
- ◆ Based on polypyrrole CP 'benders'
- ◆ Soft polymer actuators more attractive for integrated ufluidics manifolds
- ◆ 'lego' approach - detector block will slot in

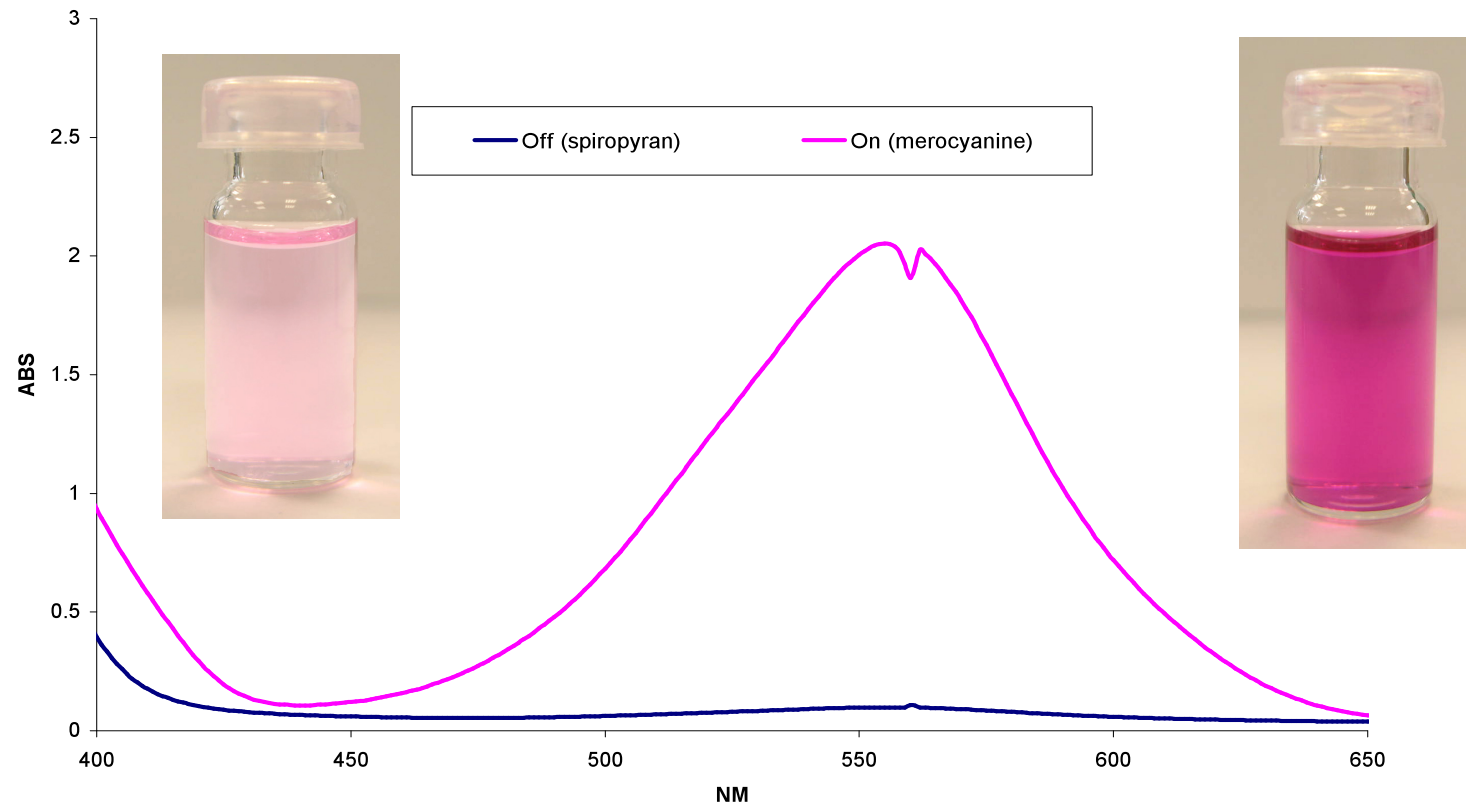
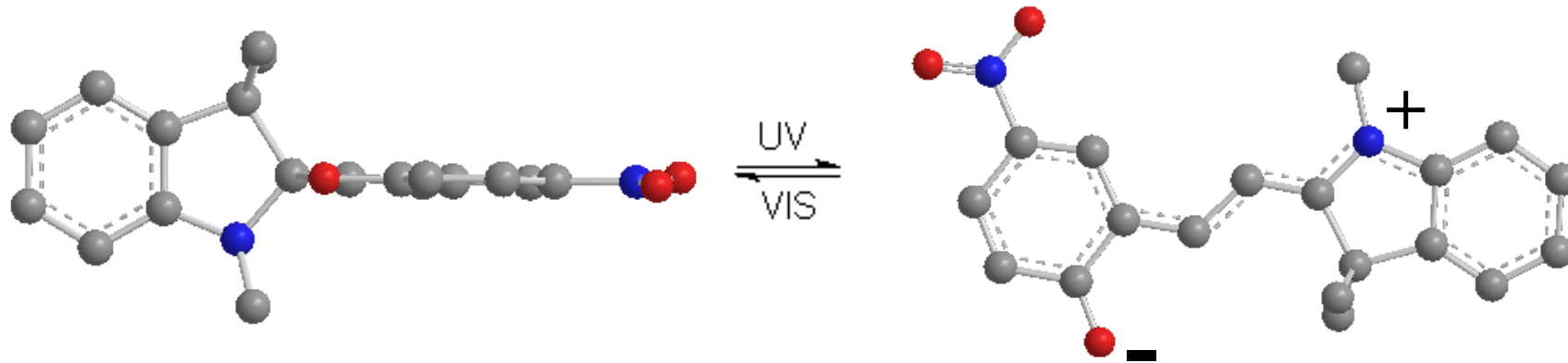


Soft Polymer Actuators

- ◆ Can be used to provide pumping and valving functions
- ◆ Can be fully integrated into microfluidic manifolds
- ◆ In principle are more reliable in micro-scale than conventional ‘hard’ materials
- ◆ Could drive down the cost and complexity of ‘analyser’ platforms



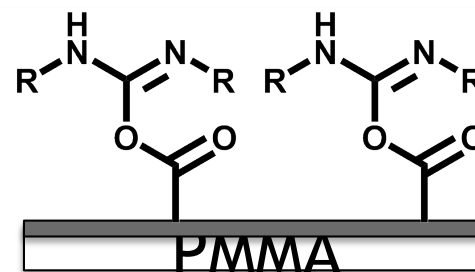
Photoswitchable Materials



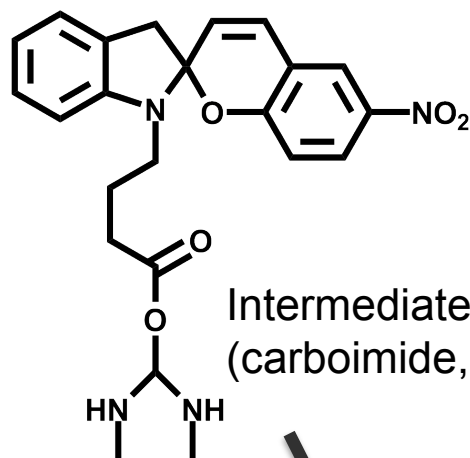
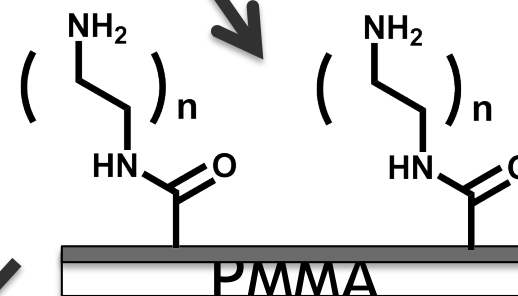
Polymethacrylic acid surface



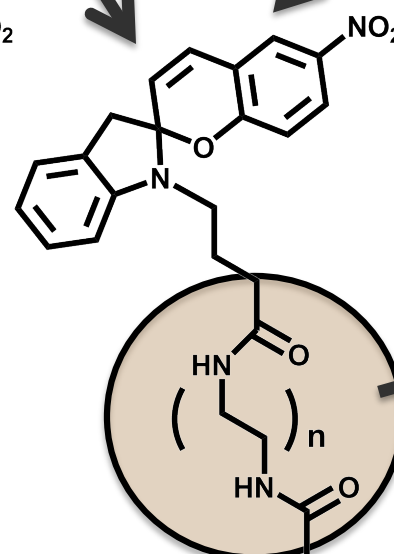
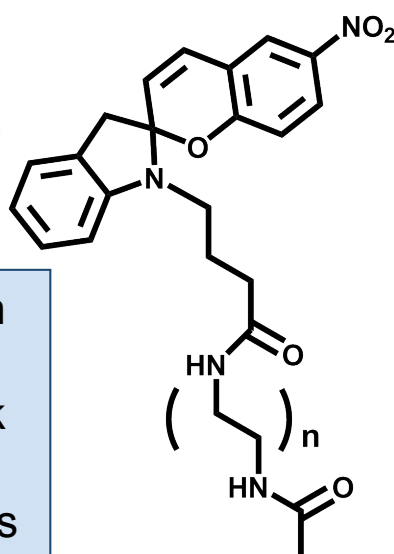
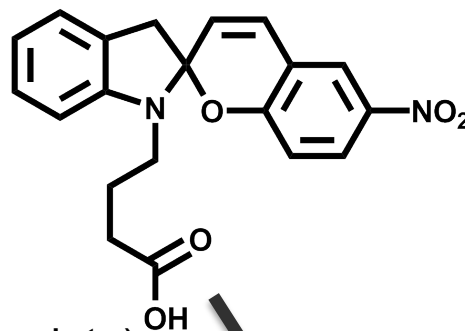
EDC



Various diamino
alkyl linkers



Intermediates
(carboimide, carboxylate)



Each -CH₂- link is ca. 1.5 Å

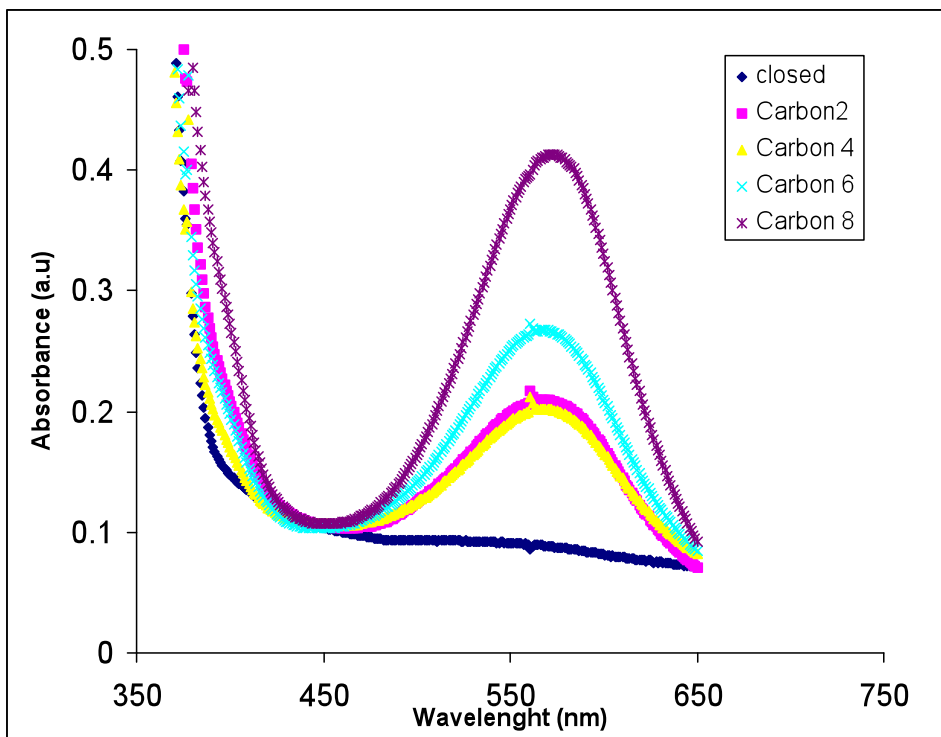
Tether Length (n=)	ID
2	SP-2
4	SP-4
6	SP-6
8	SP-8

Can be immobilised on polymer or silica surfaces, or within bulk materials, e.g. using SP-modified monomers

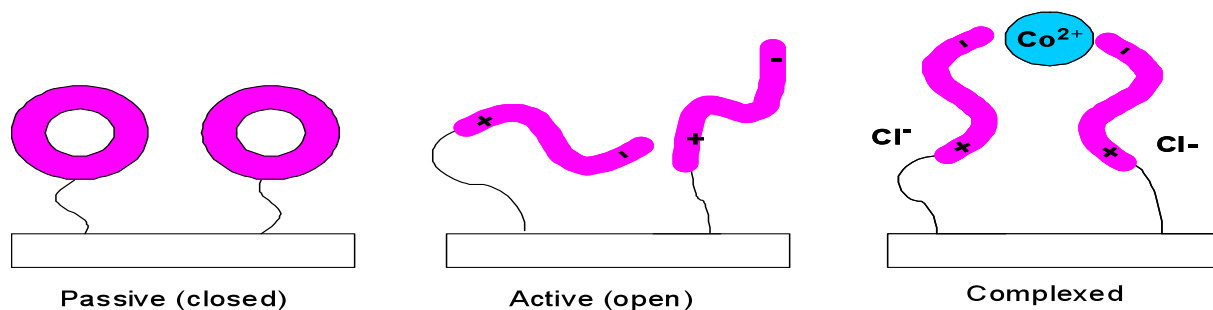
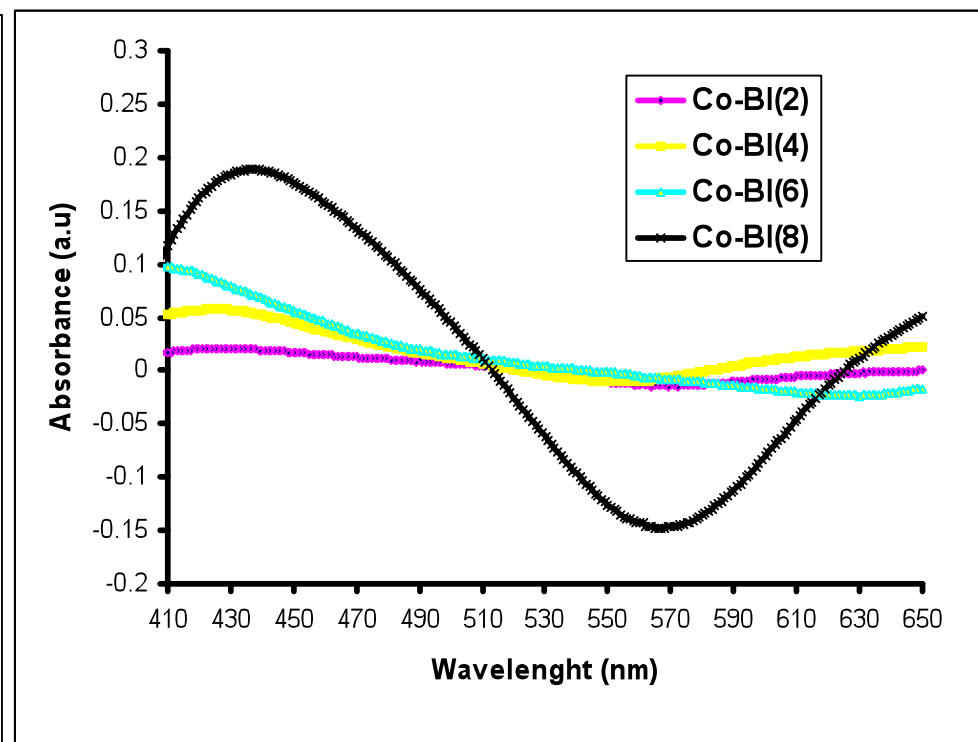


Importance of Linker Length

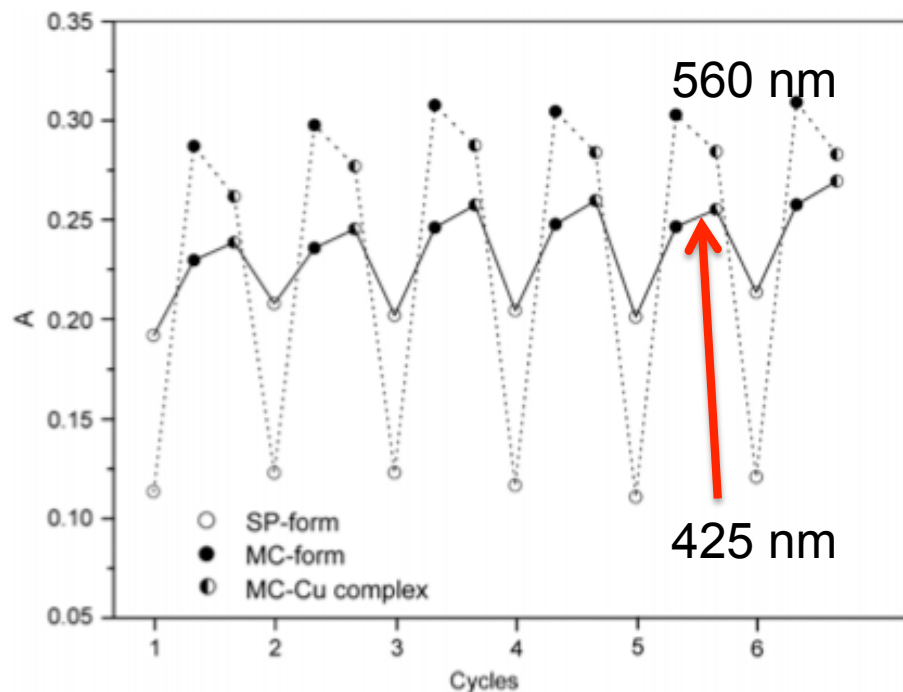
SP-MC Switching



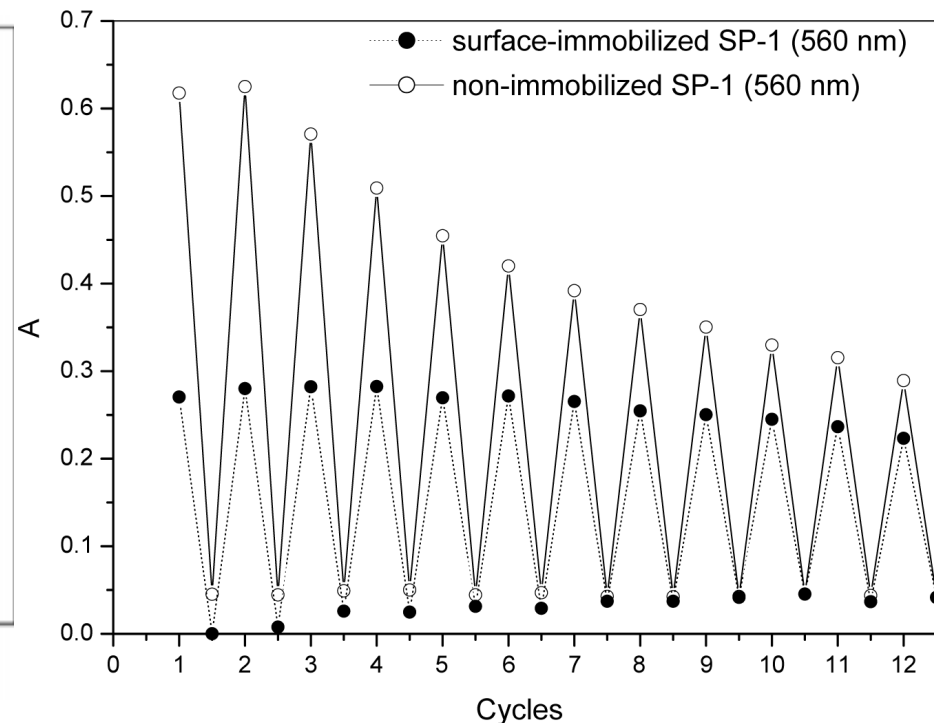
Co²⁺-Complexation



Detection of switching between SP/MC/MC-Cu states using the 'Discophotometer



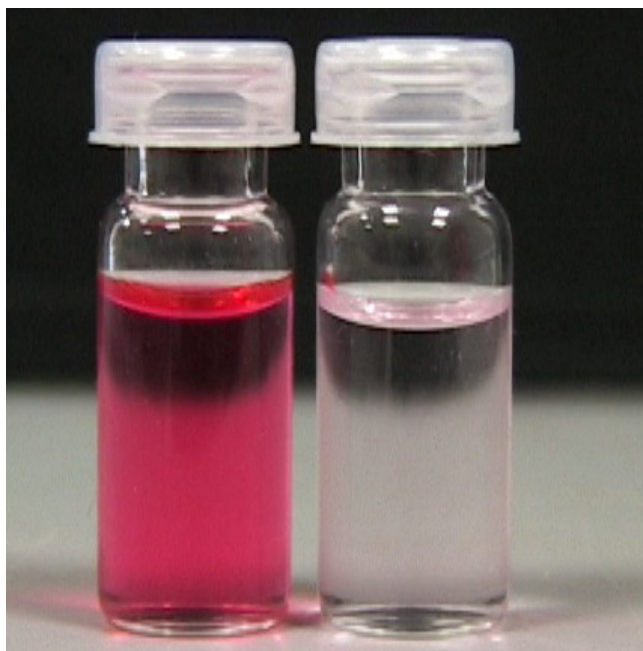
Photonic Modulation of Surface Properties: A Novel Concept in Chemical Sensing, Aleksandar Radu, Silvia Scarmagnani, Robert Byrne, Conor Slater, King Tong Lau and Dermot Diamond, J. Phys.D: Appl. Phys., 40 (2007) 7238-7244.



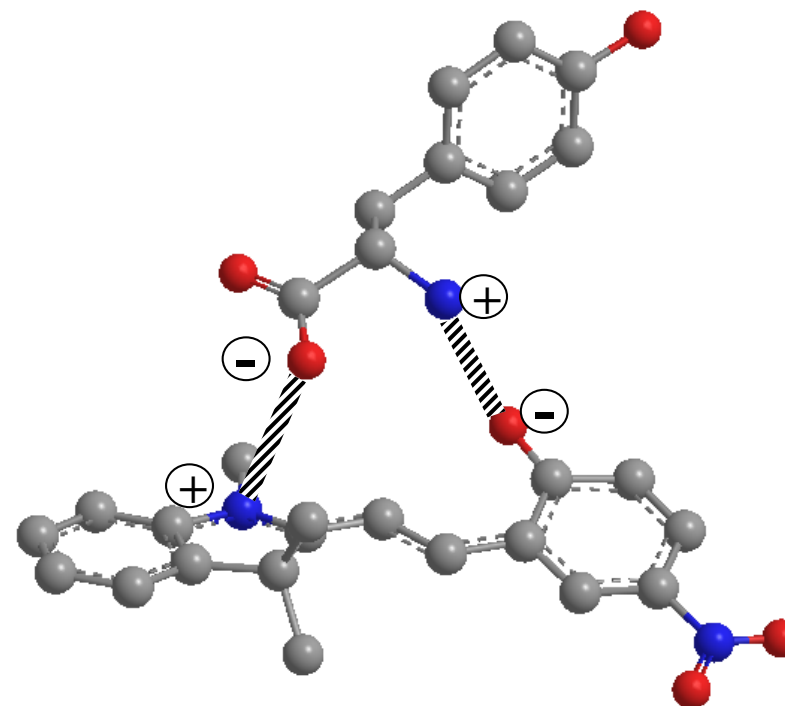
Absorbance measured at 560 nm for cyclical switching between SP and MC when for surface immobilized (full circles) and non-immobilized (open circles) SP-1.

A. Radu et al. Journal of Photochemistry and Photobiology A:Chemistry 206 (2009) 109–115.

Merocyanine Interaction with Amino Acids



A strongly coloured solution of merocyanine and tyrosine in a 4:1 acetonitrile:water mixture. The merocyanine was formed by illuminating spiropyran (1:1 mole ratio to tyrosine) for 1 minute with a UV-LED. The picture was taken after 100 hours storage in the absence of light. (top, right): The control experiment without tyrosine shows almost complete decoloration, i.e. return to spiropyran form.



Energy minimised structures (Chem 3-D Ultra, V. 9.0, Cambridgesoft) suggests complementary binding of tyrosine to the merocyanine zwitterion which stabilises the coloured merocyanine form.

Key: carbon atoms – grey, oxygen atoms – red, nitrogen atoms – blue (hydrogen atoms not shown for clarity)

Chemo/Bio-Sensor Networks, Robert J Byrne and Dermot Diamond, *Nature Mater.*, 5 (2006) 422-424.



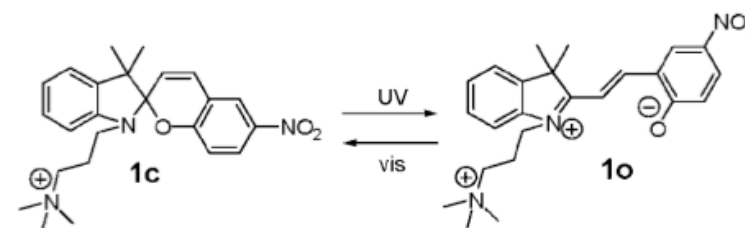
J|A|C|S
COMMUNICATIONS

Photoswitched DNA-Binding of a Photochromic Spiropyran

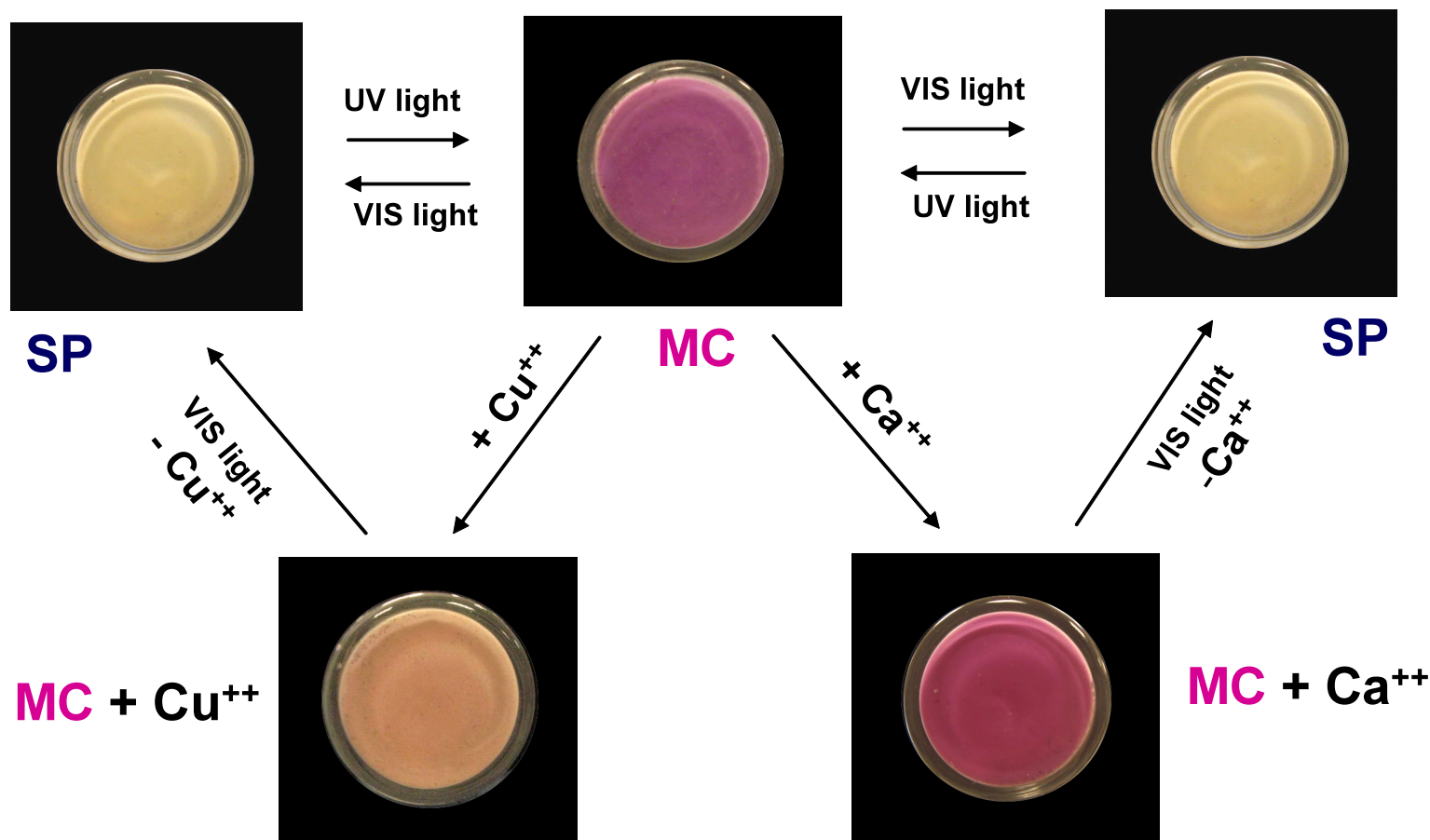
*Department of Chemical and Biological Engineering, Physical Chemistry, Chalmers University of Technology,
SE-412 96 Göteborg, Sweden*

Figure 1. Absorption spectra of **1c** (top panel) and **1o** (lower panel) in the absence (blue lines) and presence (red lines) of calf-thymus DNA. The contribution from DNA to the overall absorption has been subtracted for ease of comparison. Likewise, the contribution from **1c** has been subtracted from the spectra of **1o** shown in the lower panel. The green line corresponds to a sample of 100% **1o** bound to DNA as the contribution from unbound **1o** has been corrected for (see Supporting Information for details). The total concentration of **1** was $\sim 1.5 \times 10^{-5}$ M. The concentration of DNA was 11.6×10^{-5} M, and the NaCl concentration of the solution was 8.6×10^{-3} M.

Chart 1. Structures of Photochromic Spiropyran **1**



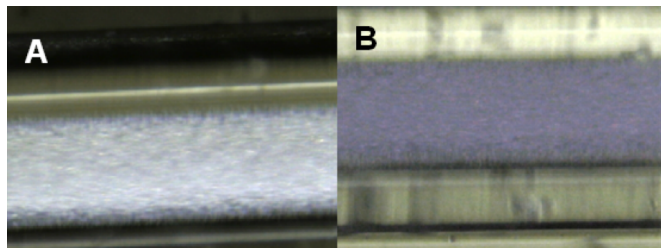
Functionalised Microbeads: optically controlled selective binding of metal ions



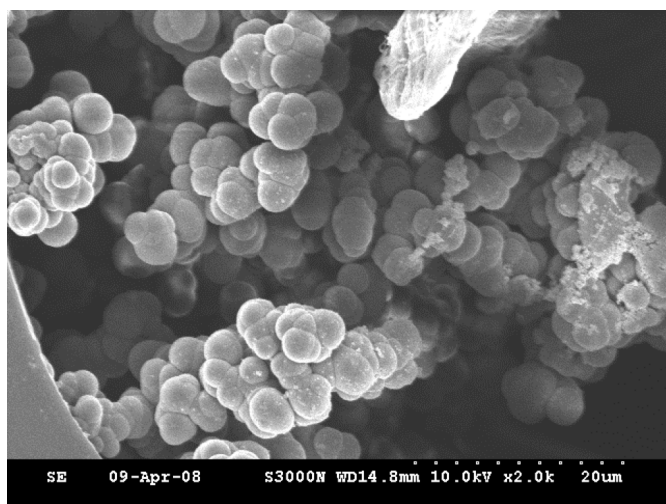
Polystyrene Beads-Based System for Optical Sensing using Spiropyran Photoswitches, Silvia Scarmagnani, Zarah Walsh, Conor Slater, Nameer Alhashimy, Brett Paull, Mirek Macka and Dermot Diamond, J. Mater. Chem., 2008, 18, 5063 – 5071



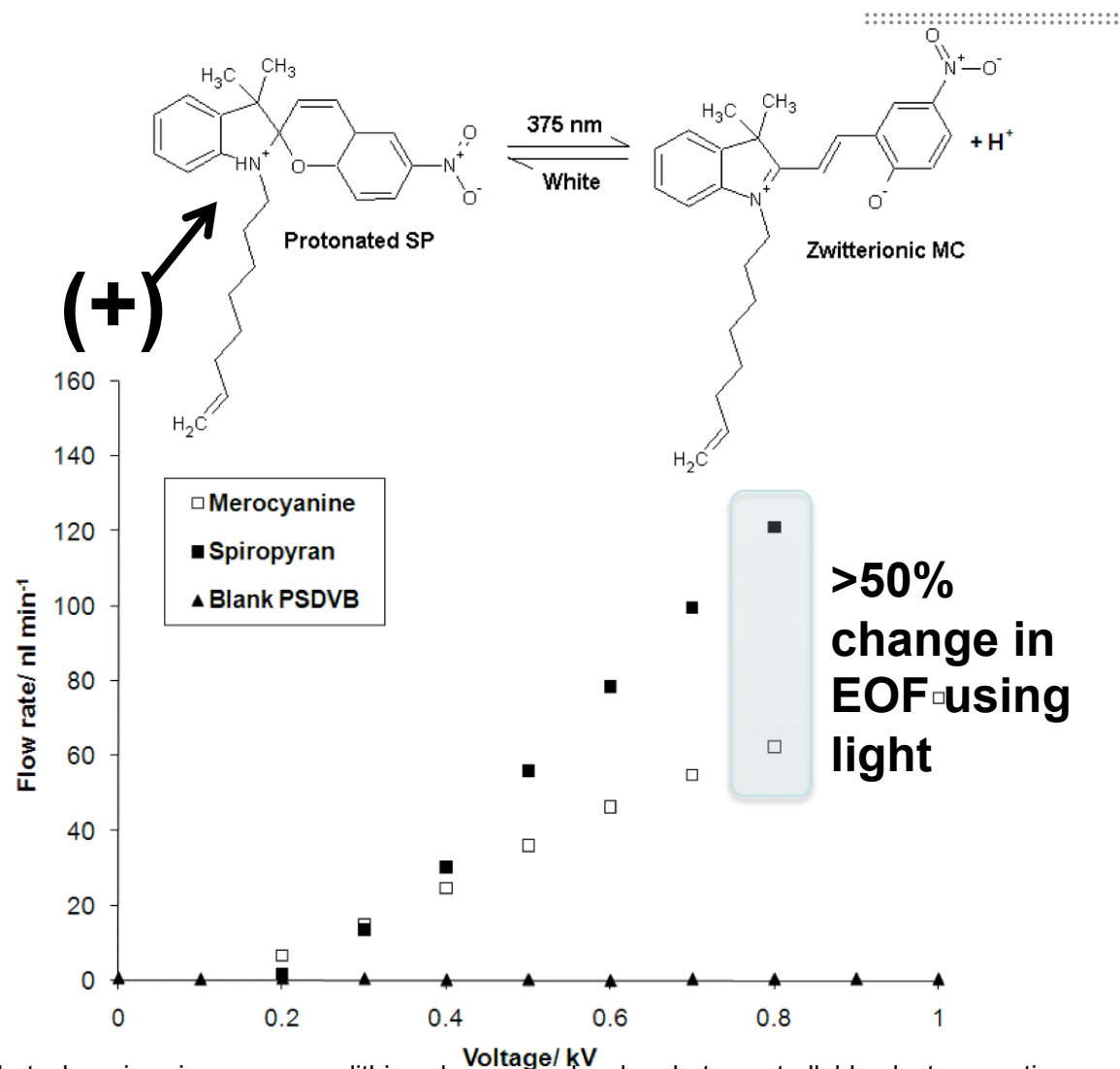
Changing EOF in CE using Light!



Optical microscope images of poly (spiropyran-co-divinylbenzene) monoliths in both the spiropyran (A) and merocyanine (B) forms.



Scanning electron micrograph of the monolithic poly (spiropyran-co-divinylbenzene) within the PTFE coated fused silica capillary; channel dimensions 8mm x 0.4mm x 0.4mm



Photochromic spiropyran monolithic polymers: molecular photo-controllable electroosmotic pumps for micro-fluidic devices, Zarah Walsh, Silvia Scarmagnani, Fernando Benito-Lopez, Silvija Abele, Fu-Qiang Nie, Conor Slater, Robert Byrne, Dermot Diamond, Brett Paull and Mirek Macka, J. Mater. Chem., 2009, submitted for publication



Polymer based photoactuators

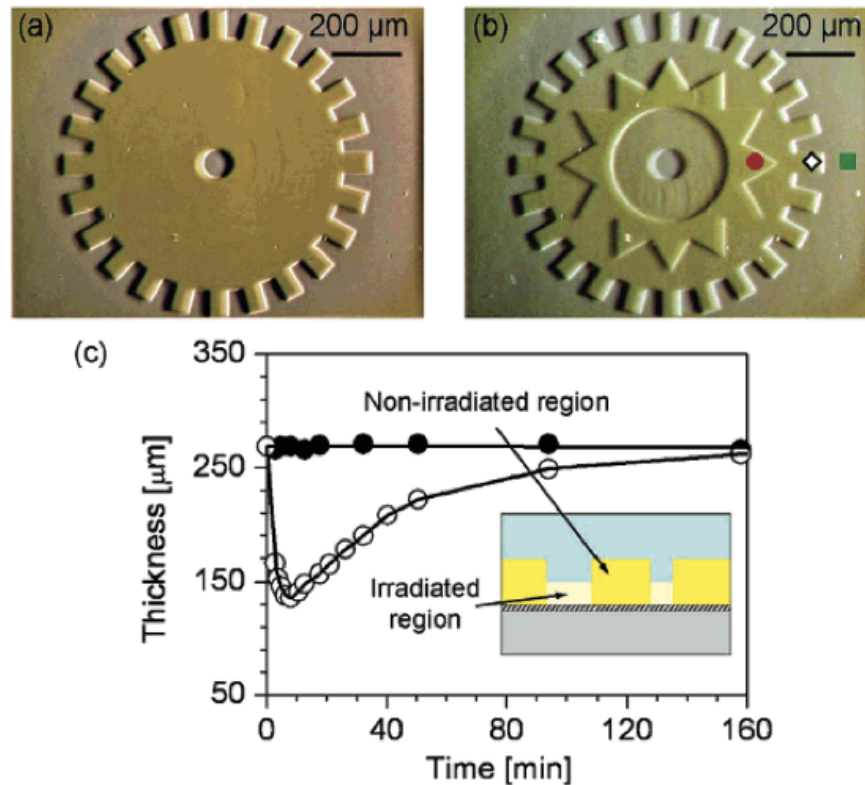
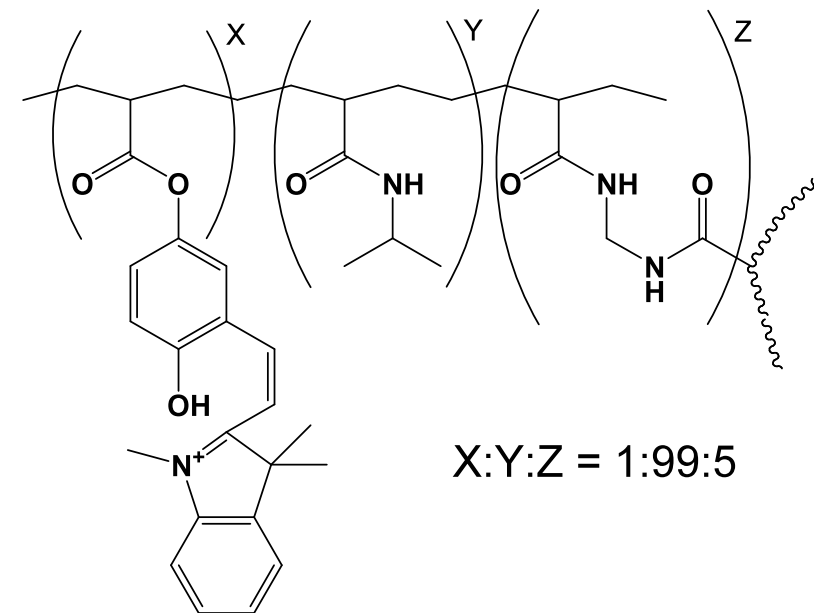


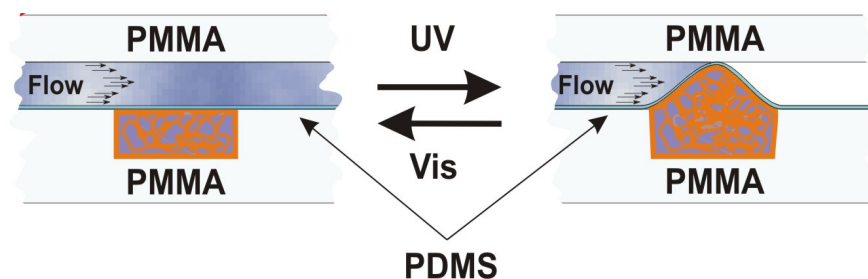
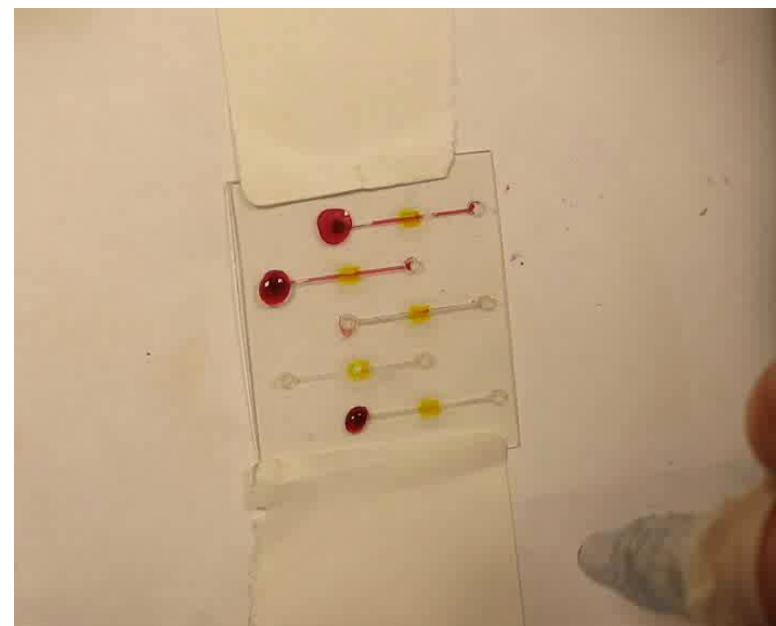
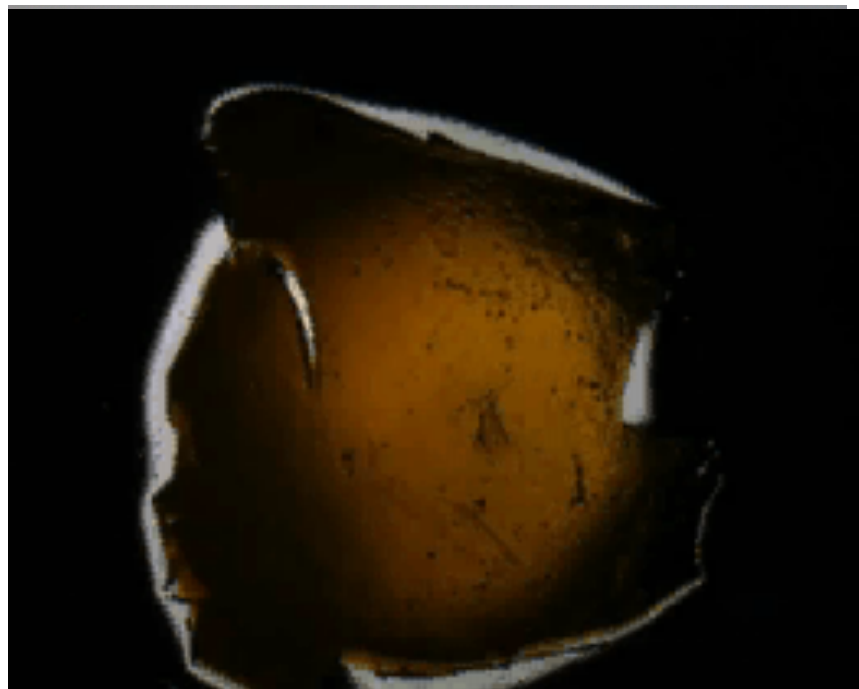
Figure 3. (a, b) Images of the pSPNIPAAm hydrogel layer just after the micropatterned light irradiation. Duration of irradiation was (●, red) 0, (◇) 1, and (■, green) 3 s. (c) Height change of the hydrogel layer in (●) non-irradiated and (○) irradiated region as a function of time after 3 s blue light irradiation.



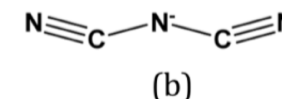
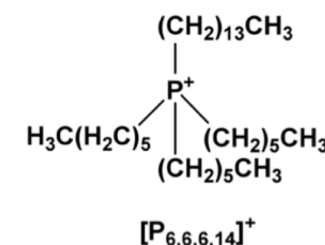
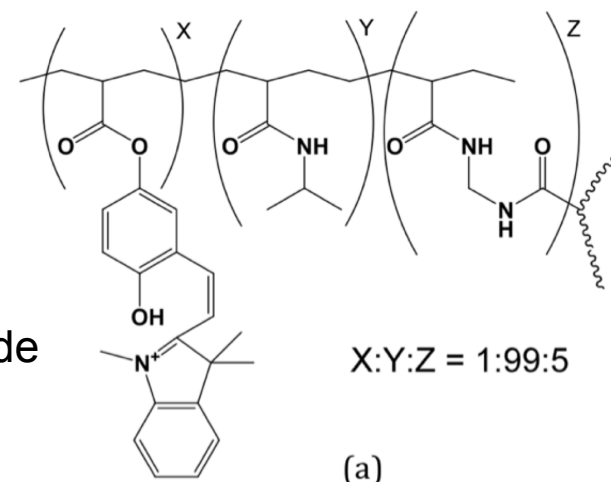
Polymer developed by by Kimio Sumaru et al¹

1) *Chem. Mater.*, **19** (11), 2730 -2732, 2007.

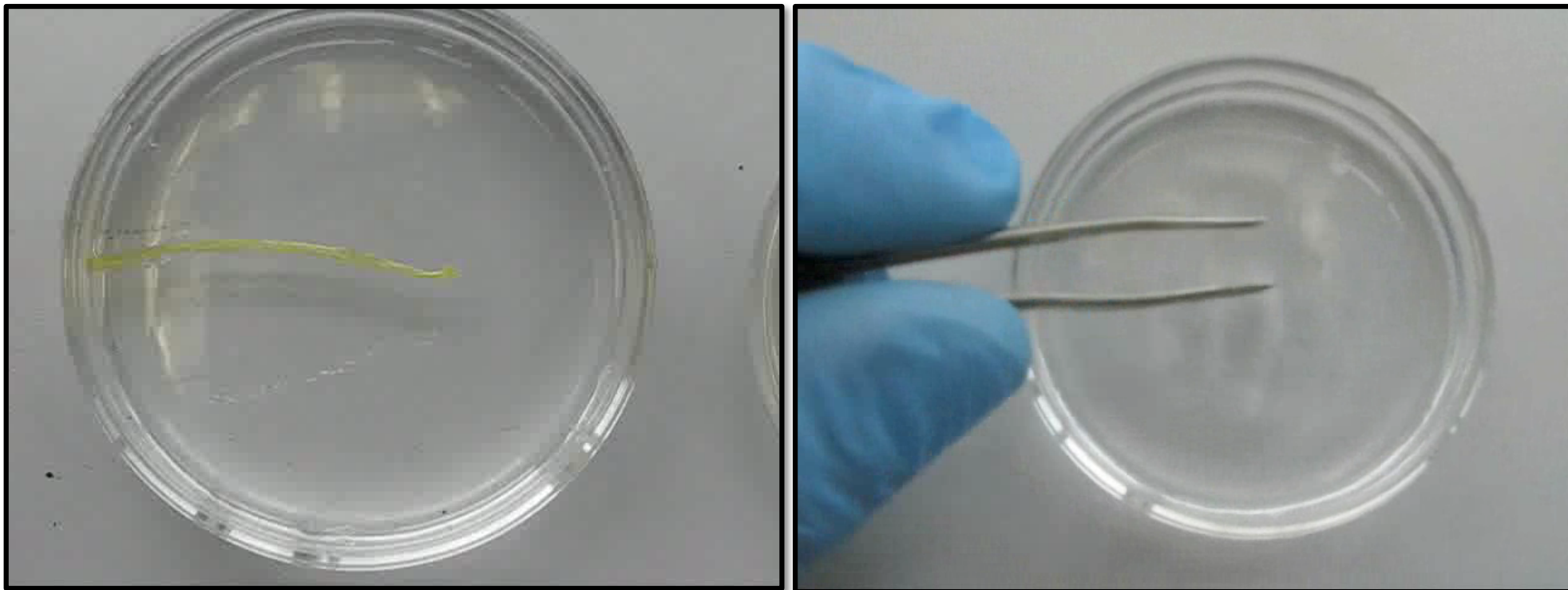
Photo-actuator polymers as microvalves in microfluidic systems



trihexyltetradecylphosphonium dicyanoamide
[P6,6,6,14][dca]



Mobile platforms with chemical actuation: No external power required



Based on solvent exchange within ionogel (water/ethanol)

Robert Byrne and Fernando Lopez



Conclusions

- ◆ Switchable materials open the way to devices with radically different behaviour - much of which can be applied to analytical flow systems
 - Biomimetic polymer pumps and valves
 - Photocontrolled actuation
 - Photocontrolled uptake and release
 - Movement of loaded particles and structures
 - Very low power flow systems
 -

Very exciting possibilities for developing next generation analytical devices including separations targeting many environmental and pHealth applications

