

‘Water, water, every where, Nor any drop to drink’

(The Rime of the Ancient Mariner, Samuel Taylor Coleridge, 1798)

Dermot Diamond

INSIGHT Centre for Data Analytics

National Centre for Sensor Research

Dublin City University

Invited public lecture presented at

The Royal Society of Chemistry

Burlington House, Piccadilly, London

3rd April 2014



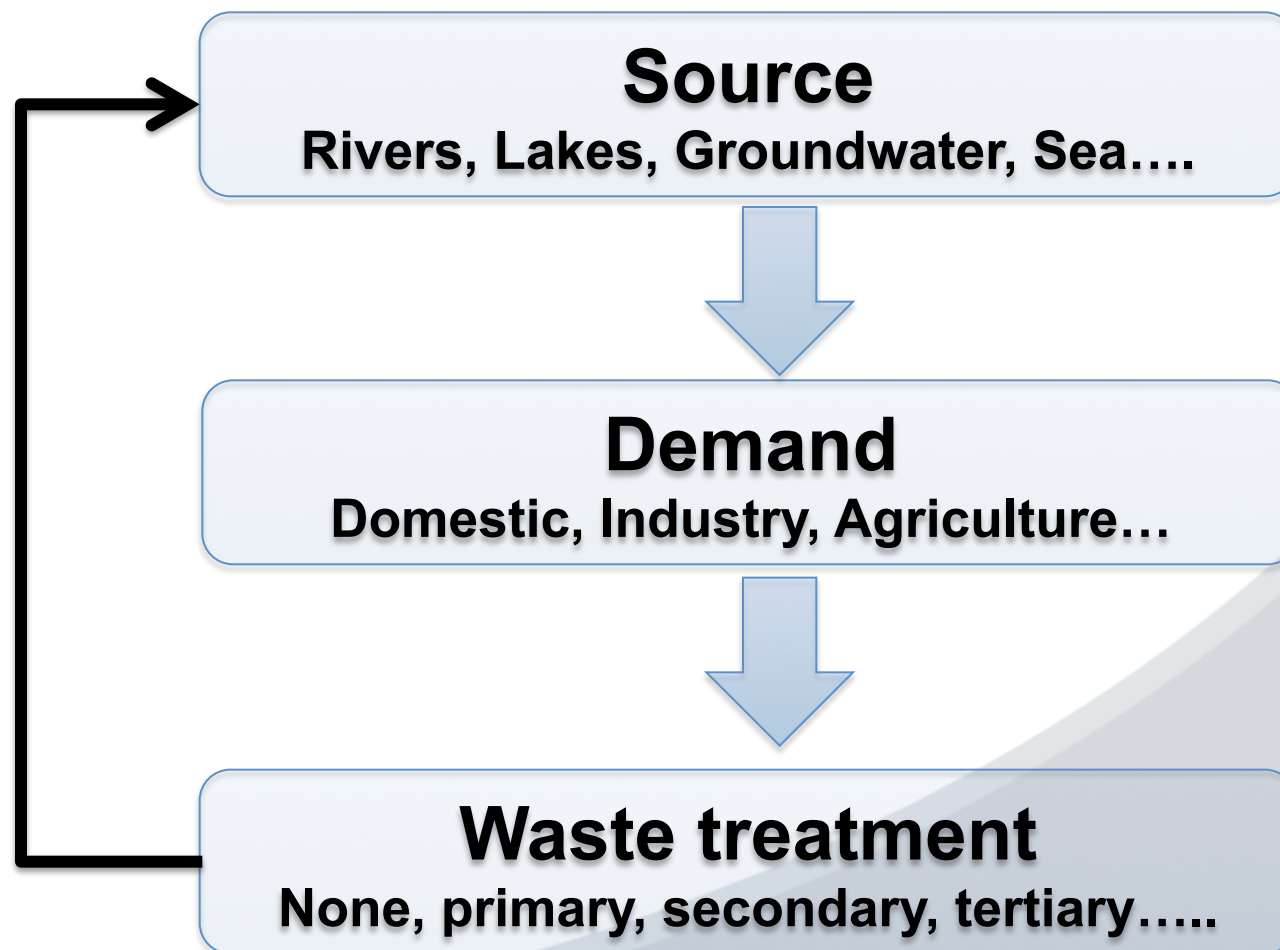
Water and the Romans...

‘The sewers, covered with a vault of tightly fitted stones, have room in some places for hay wagons to drive through them. And the quantity of water brought into the city by aqueducts is so great that rivers, as it were, flow through the city and the sewers; almost every house has water tanks, and service pipes, and plentiful streams of water...’

Strabo, (Greek author ca.60 BC to AD 24, writing about the Romans); see <http://www.britannica.com/EBchecked/topic/567832/Strabo>



Water supply & demand cycle



Importance of Water

WATER UK working on behalf of the water industry towards a sustainable future

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Drinking Water

home page : waterfacts : industry : drinking water

In the UK our drinking water is of the highest standard, at a record level of quality and among the best in the world. We can turn on our taps with the certainty of a safe, clean and refreshing supply.

- Quality standards
- UK drinking water quality reports
- How water quality is assured
- Technical briefings

Quality standards

UK water suppliers place the highest priority on assuring the quality of water provided to their customers. Strict standards for the quality of the public supply are laid down in national regulations derived from the EU Drinking Water Directive (98/83/EC). These standards are based on advice from the World Health Organization (WHO) and are regularly reviewed.

Latest figures for drinking water quality compliance with the strict UK and European standards are:

- England and Wales 99.96%
- Scotland 99.86%
- Northern Ireland 99.78%

Water quality is closely checked and regulated by independent drinking water inspectorates in England and Wales, Scotland and Northern Ireland.

UK drinking water quality reports

Drinking water quality

Publications

- Water supply hygiene: quality reporting
- Technical briefings
- How the quality of drinking water is assured

Access technical briefing papers here

Quality regulators

- Drinking Water Inspectorate England & Wales
- Drinking Water Quality Regulator for Scotland
- Drinking Water Inspectorate (within the Northern Ireland Environment Agency) Northern Ireland

theguardian

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Sanitation rated the greatest medical advance in 150 years

Sarah Boseley, health editor
The Guardian, Friday 19 January 2007

Sanitation is the greatest medical milestone of the last century and a half, according to a poll carried out by the British Medical Journal.

Sanitation was the clear winner among 15 milestones shortlisted by readers of the journal, including the development of vaccines, which has safeguarded many children's lives, and the invention of the contraceptive pill, which was a contributory factor to significant social change.

The winner was chosen by more than 11,000 members of the public around the world, who were invited to read articles championing each of the 15 contenders by prominent scientists, either in the journal or on the BMJ website. The competition was to mark the relaunch of the BMJ and all the innovations had to have taken place since it was first published in 1840.

Sanitation was the undisputed winner, with 1,795 votes, over antibiotics in second place with 1,642 votes, and anaesthesia which took third place with 1,574.



Examples of Legislation (water only)

- [Bathing Water Directive](#),
- [Drinking Water Directive](#)
- [Environmental Impact Assessment Directive](#)
- [Sewage Sludge Directive](#)
- [Urban Waste Water Treatment Directive](#)
- [Nitrate Directive](#)
- [Integrated Pollution Prevention Control Directive](#),
 - cost recovery for water use,
 - measures to promote efficient and sustainable use of water,
 - protection of drinking water sources,
 - authorisation of discharges to groundwater,
 - control of point source discharges and diffuse source pollution,
 - prevention or reduction of accidental pollution



WFD (1)

- **The Directive aims to achieve ‘good water status’ for all waters by 2015.**
 - including inland waters (surface and groundwater, drinking water) and transitional and coastal waters up to one sea mile (and for chemical status also territorial waters which may extend to 12 sea miles) from the territorial baseline of a Member State
- **According to Article 8, ‘monitoring networks should be made operational by 2006’.**
 - These networks are part of the establishment of two primary monitoring programmes: the Surveillance Monitoring (SM) and the Operational Monitoring (OM) networks for surface waters and groundwater.

Key Analytical Targets under WFD

Rivers

Chemical / Physico-chemical

Thermal conditions: Temperature

Oxygenation conditions: DO

Salinity: Conductivity

Acidification status: pH ANC Alkalinity

Nutrients: TP, TN, SRP*, NO₃ + NO₂ NH₄

*SRP = soluble reactive phosphorus

Biological

Benthic Invertebrates

Macrophytes

Benthic Algae

Fish

Phytoplankton

Chemical / Physico-chemical

Thermal conditions: Thermal profiles

Oxygenation: DO profiles

Salinity: ppt, psu*

Transparency: light penetration

Nutrients: Reactive species and total budgets
(N,P,Si)

*PSU = practical salinity units; 1 PSU = ca. 1000 ppm NaCl

Sea water is ca. 35 PSU

Biological

Benthic Invertebrate fauna

Macroalgae

Angiosperms

Fish fauna

Phytoplankton

Transitional Waters

Chemical / Physico-chemical

Thermal conditions: Thermal profiles

Oxygenation: DO profiles

Salinity: ppt, psu

Transparency: light penetration

Nutrients: NO₃, NO₂, NH₄, PO₄, Si, TN, TP

Biological

Benthic Invertebrate fauna

Macroalgae / Angiosperms

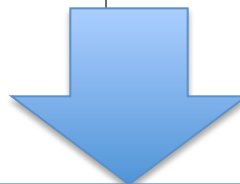
Phytoplankton

Coastal Waters



Examples of Priority Substances

Number	CAS number ⁱ	EU number ⁱⁱ	Name of priority substance ⁱⁱⁱ	Identified as priority hazardous substance	CAS number	Name of other pollutant
(1)	15972-60-8	240-110-8	Alachlor		(6a) 56-23-5	Carbon-tetrachloride(1)
(2)	120-12-7	204-371-1	Anthracene	X	(9b) not applicable	DDT total (1)(2)
(3)	1912-24-9	217-617-8	Atrazine		50-29-3	para-para-DDT (1)
(4)	71-43-2	200-753-7	Benzene		(9a)	Cyclodiene pesticides
(5)	not applicable	not applicable	Brominated diphenyletheriv	X	309-00-2	Aldrin (1)
	32534-81-9	not applicable	Pentabromodiphenylether (congener numbers 28, 47, 99, 100, 153 and 154)		60-57-1	Dieldrin (1)
(6)	7440-43-9	231-152-8	Cadmium and its compounds	X	72-20-8	Endrin (1)
(7)	85535-84-8	287-476-5	Chloroalkanes, C10-13 iv	X	465-73-6	Isodrin (1)
(8)	470-90-6	207-432-0	Chlorfenvinphos		(29a) 127-18-4	Tetrachloro-ethylene (1)
(9)	2921-88-2	220-864-4	Chlorpyrifos (Chlorpyrifos-ethyl)		(29b) 79-01-6	Trichloro-ethylene (1)
(10)	107-06-2	203-458-1	1,2-Dichloroethane			
(11)	75-09-2	200-838-9	Dichloromethane			
(12)	117-81-7	204-211-0	Di(2-ethylhexyl)phthalate (DEHP)			
(13)	330-54-1	206-354-4	Diuron			
(14)	115-29-7	204-079-4	Endosulfan	X		
(15)	206-44-0	205-912-4	Fluoranthenevi			
(16)	118-74-1	204-273-9	Hexachlorobenzene	X		
(17)	87-68-3	201-765-5	Hexachlorobutadiene	X		
(18)	608-73-1	210-158-9	Hexachlorocyclohexane	X		



Only the most powerful analytical techniques can detect these complex mixtures of target molecules at low concentrations; e.g., LC-MS; GC-MS

There is no alternative to grab sampling and subsequent analysis back in a certified laboratory



Cocaine in Dublin Waste Water

Using environmental analytical data to estimate levels of community consumption of illicit drugs and abused pharmaceuticals†

Jonathan Bones,^a Kevin V. Thomas^b and Brett Paull^{*a}

Received 23rd February 2007, Accepted 1st May 2007

First published as an Advance Article on the web 17th May 2007

DOI: 10.1039/b702799k

A solid phase extraction (SPE) method has been developed and applied in conjunction with a previously reported liquid chromatography tandem mass spectrometry (LC–MS–MS) procedure for the determination of illicit drugs and abused pharmaceuticals in treated wastewater and surface water samples at the ng L⁻¹ level. A full method validation was also performed and determined levels of analytical sensitivity were found to lie in the 1–10 ng L⁻¹ range using river water as a test sample matrix and a sample size of 500 mL. The developed procedure was successfully applied for the determination of the chosen analytes in wastewater treatment plants in Dublin, Ireland and rapidly expanding commuter towns in the surrounding counties. Cocaine was detected in 70% of the collected samples in the range of 25–489 ng L⁻¹, its primary metabolite, benzoylecognine (BZE) was also detected in the range of 22–290 ng L⁻¹. Other substances detected included morphine, Tempazepam and the primary metabolite of methadone.

Highlights in Chemical Science

News from across RSC Publishing.



Something in the water

18 May 2007

How much cocaine is going up Dublin's nose? The answer lies in the output from the city's wastewater treatment plants – at least according to research done at Dublin City University by Brett Paull and colleagues. The researchers have estimated the levels of consumption of substances such as cocaine and morphine by measuring their concentrations, and the concentrations of their metabolites, in effluent and surface water.¹

'There are currently very few ways to non-invasively assess community consumption of illicit drugs,' said Paull. 'Social survey data, although useful, is often subject to sampling bias, whereas environmental forensic approaches such as this, when subject to stringent quality assurance and controls, could provide a more accurate assessment.'

Previously, Paul's team surveyed drug contamination on euro

banknotes in Dublin.² 'However, as an indicator of societal abuse, such samples provide only limited information,' said Paul. 'But water treatment plants accurately record the volume of water flow and serve a known number of people, allowing total consumption to be estimated.'

Using samples taken from the water plant that serves 1.7 million people within the Dublin metropolitan area, the researchers calculated that more than 220 grams of cocaine flowed through the plant per day. Taking into account the way cocaine is metabolised in the body, this equates to the total consumption of more than 2200 grams of cocaine. The researchers estimate that this is equivalent to a consumption level of about 1.4 grams per 1000 people per day.

'The problem of cocaine consumption is not wholly confined to the capital city,' said Paull. 'Similar measurements on wastewater treatment plants that serve small towns just outside Dublin revealed consumption levels at about one fifth that of the metropolitan area.'

The researchers believe that their method could be used for the routine monitoring of cocaine consumption within a community, revealing any usage trends, such as increases during weekends and public holidays.

Colin Batchelor

References

1. J. Bones, K. V. Thomas and B. Paull, *J. Environ. Monit.*, 2007, DOI: 10.1039/b702799k
2. J. Bones, M. Macka and B. Paull, *Analyst*, 2007, DOI: 10.1039/b615669j

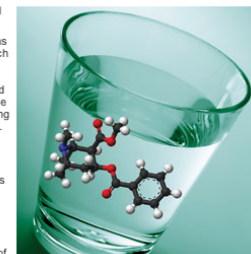
Link to journal article

Using environmental analytical data to estimate levels of community

consumption of illicit drugs and abused pharmaceuticals

Jonathan Bones, Kevin V. Thomas and Brett Paull, *J. Environ. Monit.*, 2007, 9, 701

DOI: 10.1039/b702799k



Cocaine is just one of the illicit drugs flowing through Dublin's water treatment plants

www.rsc.org/Publishing/ChemScience/Volume/2007/06/Dublin_cocaine.asp



Environmental Regulation – What about Enforcement??

- **Challenges**

- A detailed regulatory framework is in place
- Regulations cannot be enforced without measurements
- Current norm is manual grab sampling 3 or 4 times a year
- We do not measure the status of our environment often enough in enough locations

- **Solution**

- Simple: Measure **more often** in **more locations**
- **Why is this not happening?**

Conventional Analytical Equipment used for environmental monitoring is too expensive to buy and maintain:

Transporting samples to centralised laboratories is not a scalable model



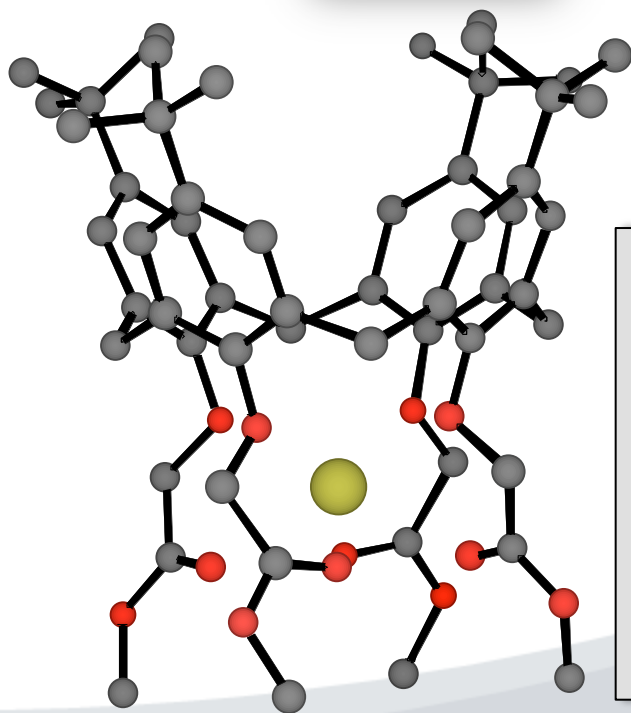
Budapest - 1956



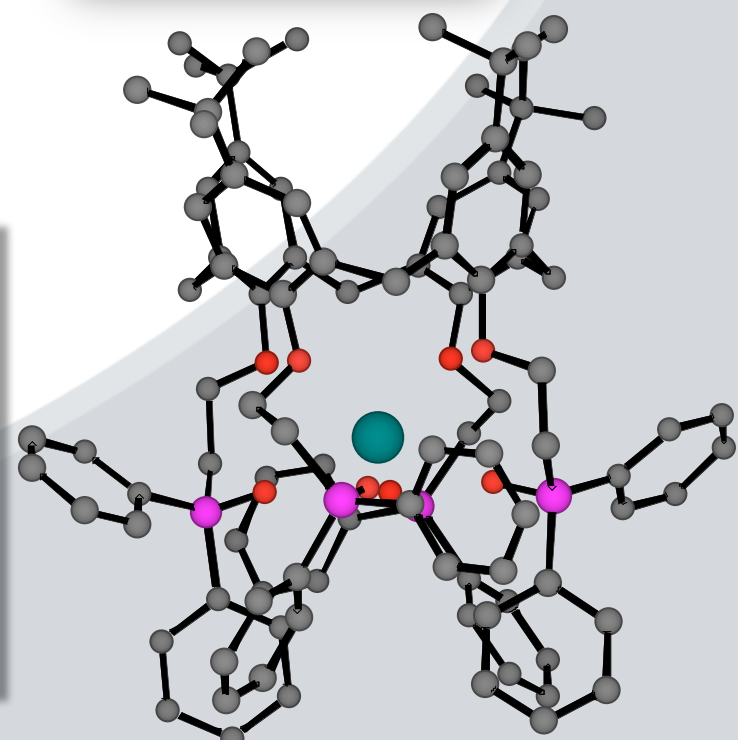
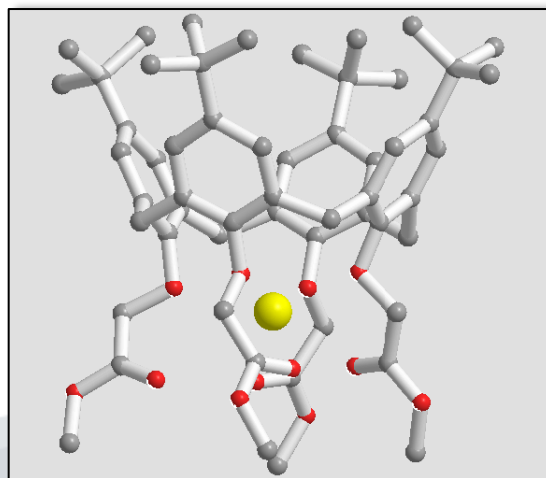
History: Calixarenes, 1984/5



+



Tetraester - Na⁺ selective

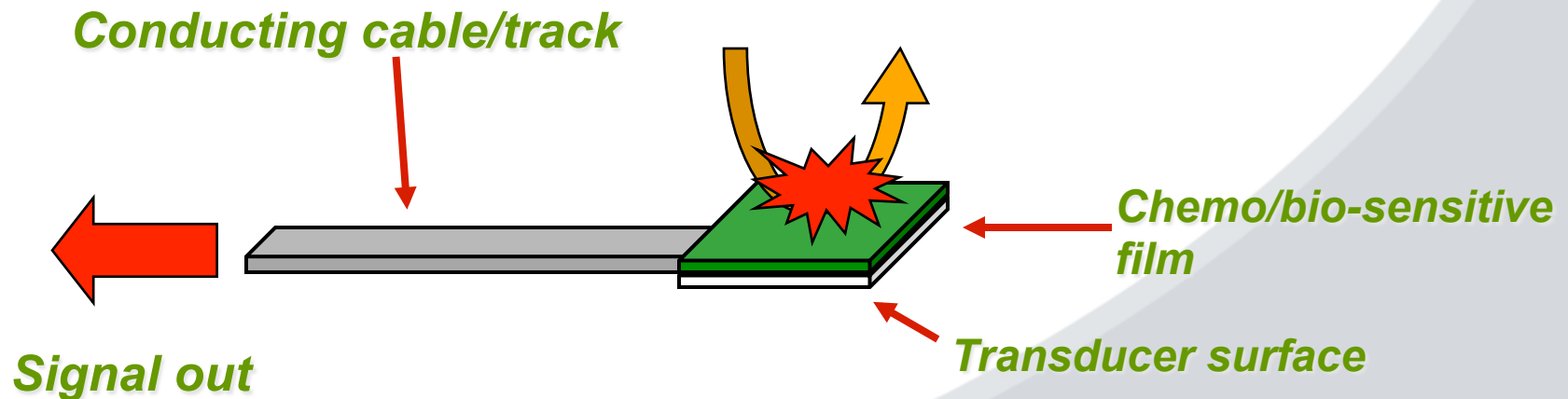


Tetraphosphine oxide - Ca²⁺ selective



What is a Chemo/Bio-Sensor?

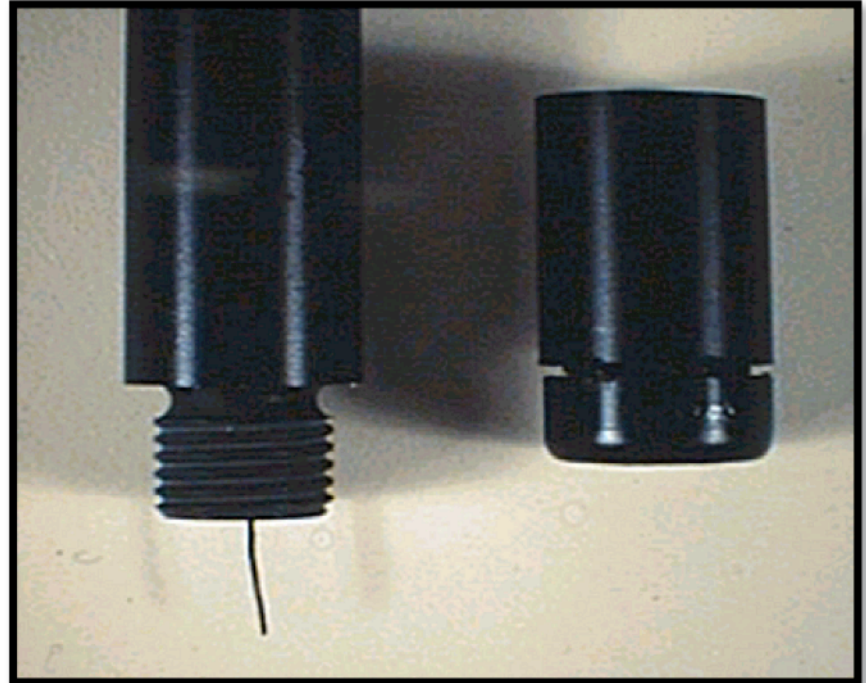
‘a device, consisting of a transducer and a chemo/bio-sensitive film/membrane, that generates a signal related to the concentration of particular target analyte in a given sample’



Chemo/Bio-sensing involves selective **BINDING & TRANSDUCTION** on the device surface; this also implies the target analyte **MUST** meet the device surface (**LOCATION & MOVEMENT**). It provides a signal observable in the macroscopic world (**COMMUNICATION**)



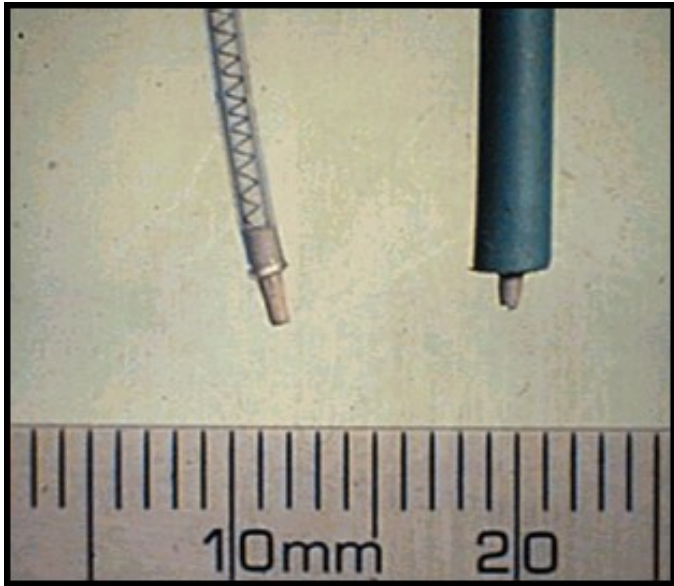
PVC - Membrane ISEs



Typical membrane cocktail (%w/w); PVC:33%, NPOE (plasticiser):66%; ionophore/exchanger: 1% (ratio at least 2:1 by mole); dissolve in a volatile solvent e.g. THF and cast membrane from this solution



Blood Analysis; Implantable Sensors



1985: Catheter Electrodes for intensive care – function for 24 hrs

Dr. David Band, St Thomas's Hospital London

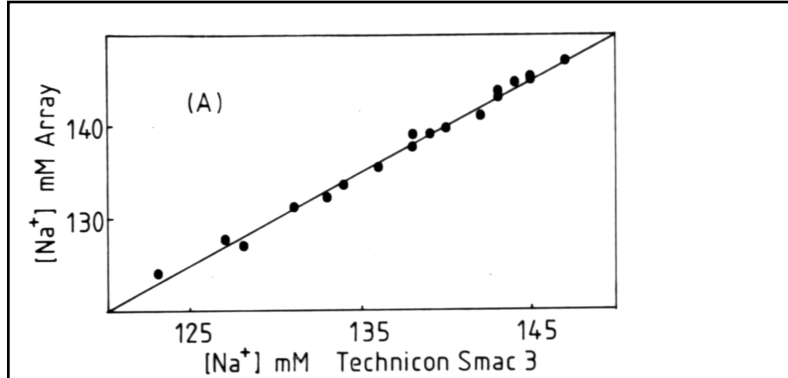
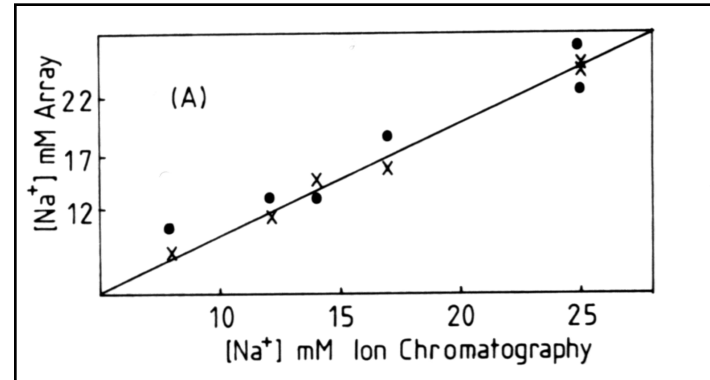


Fig. 3. Comparison of plasma sodium analysis using the array-FIA approach with a SMAC analyser. Good correlation without bias is obtained [5].



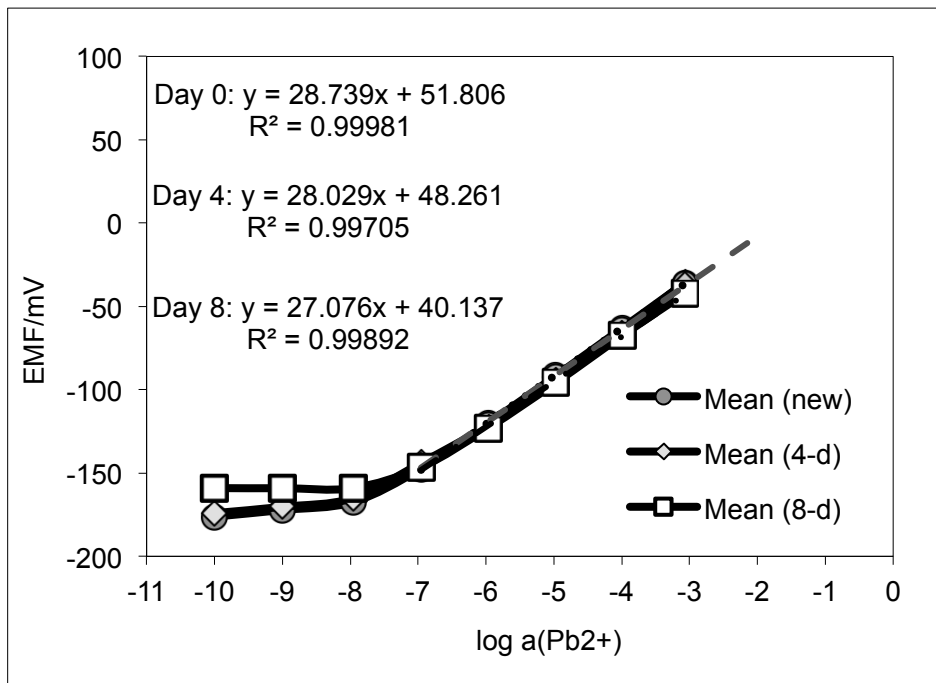
Anal. Chem., **64** (1992) 1721-1728.

Ligand (and variations of) used in many clinical analysers for blood Na^+ profiling

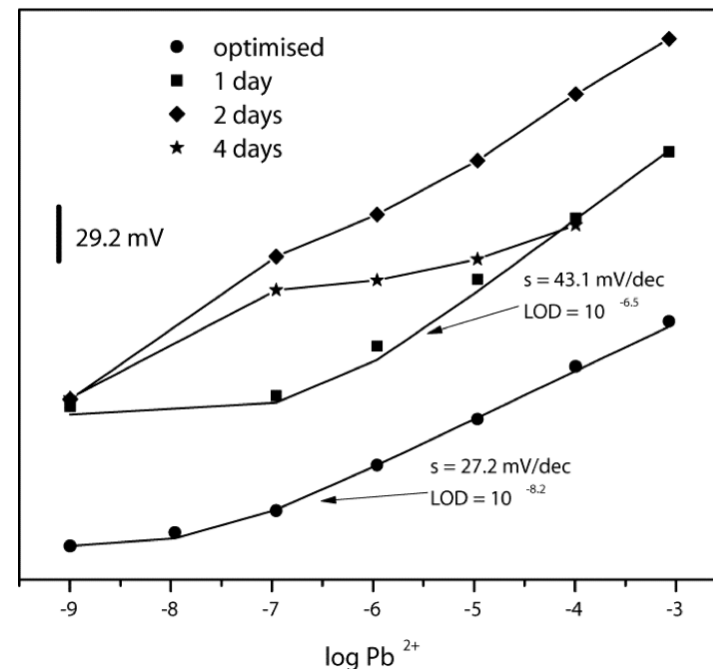


Change in Electrode Function over Time

See *Electrochimica Acta* 73 (2012) 93–97



stored in 10^{-9}M Pb^{2+} , pH=4

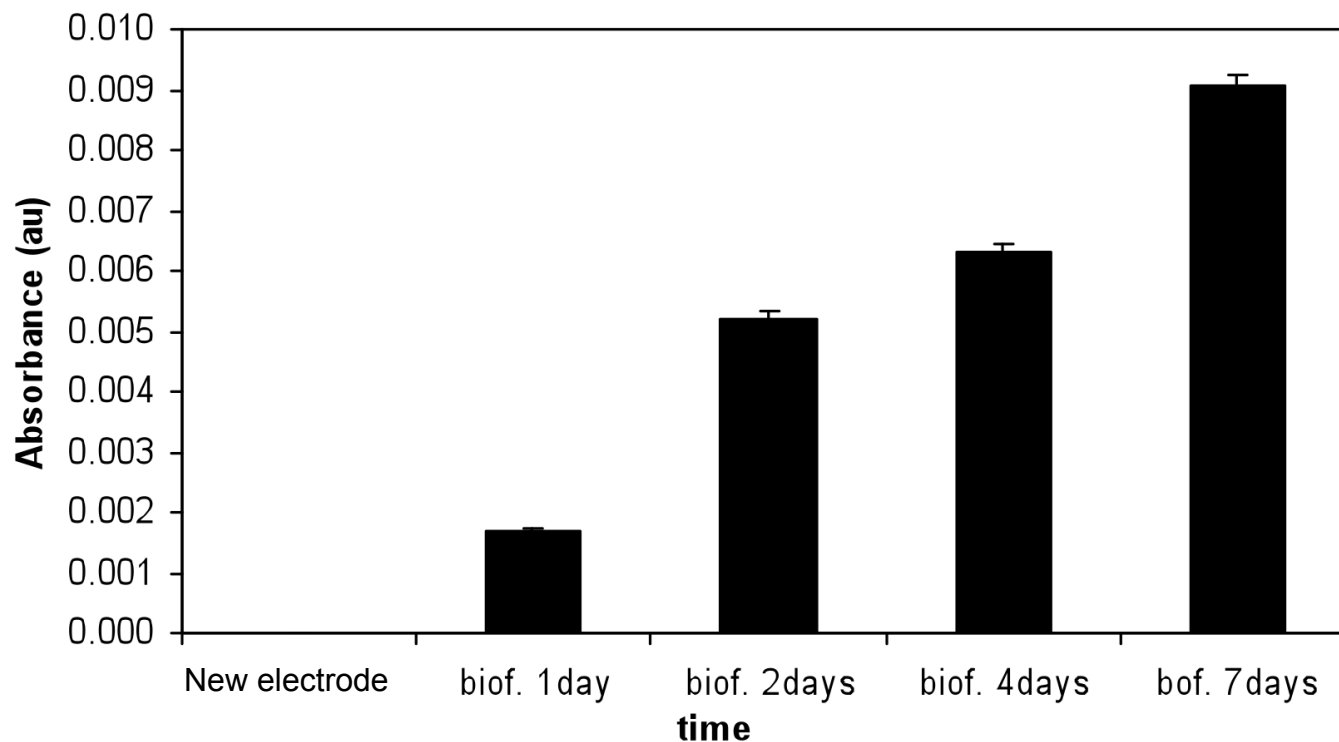


Continuous contact with river water

Conventional PVC-membrane based ISEs

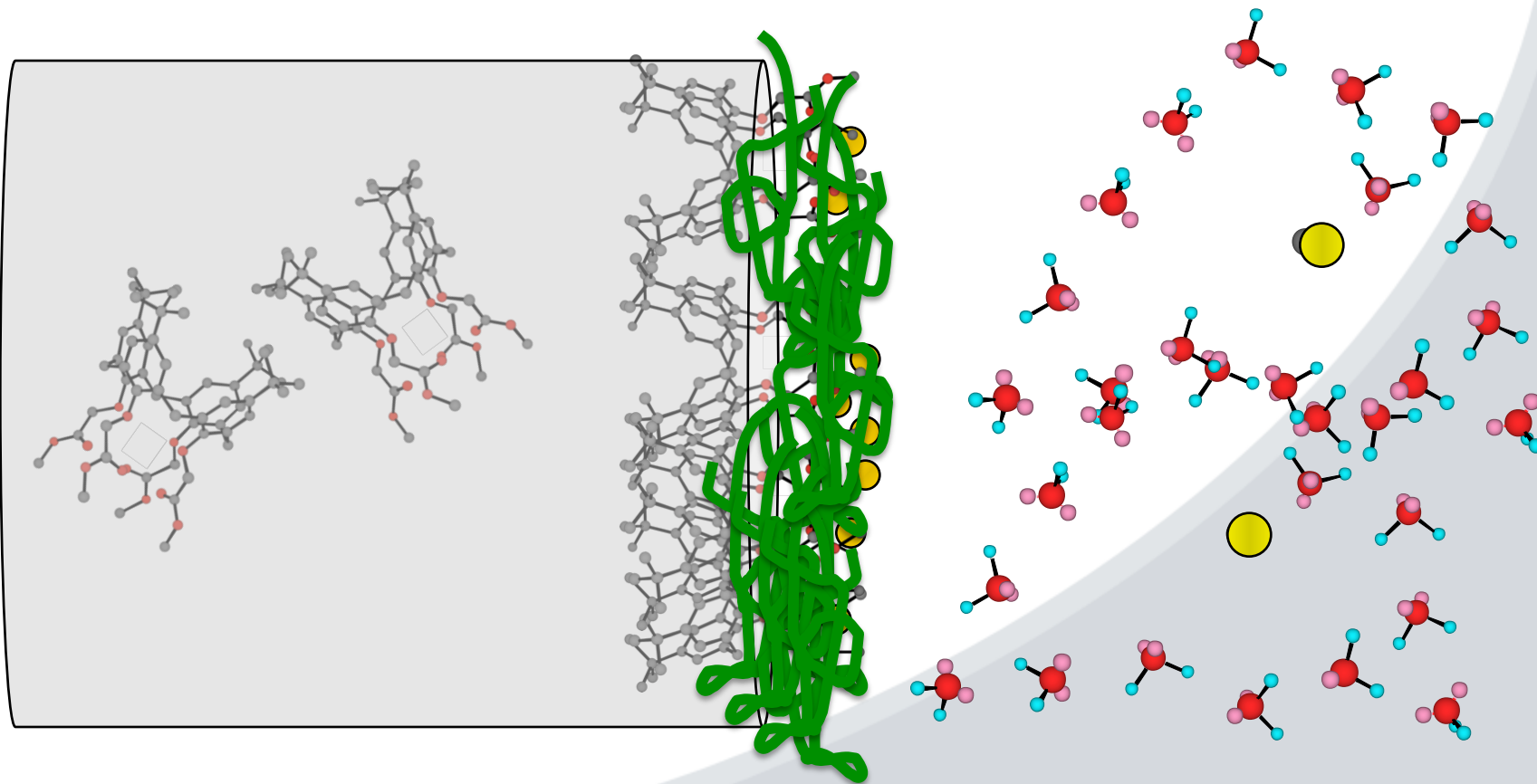


Biofilm Formation on Sensors



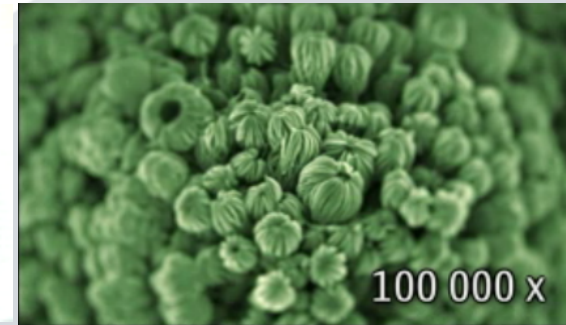
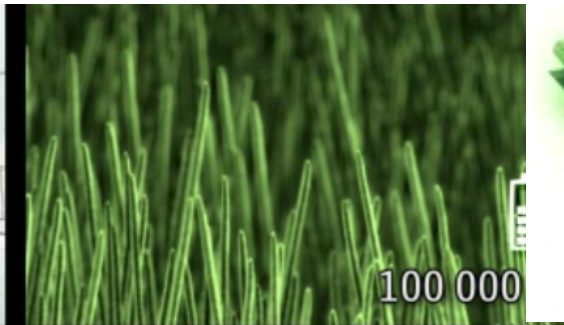
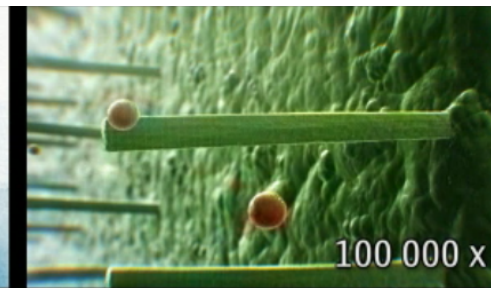
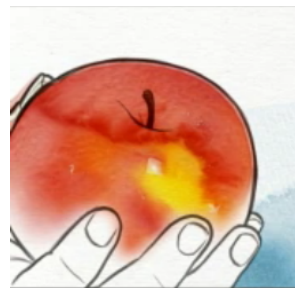
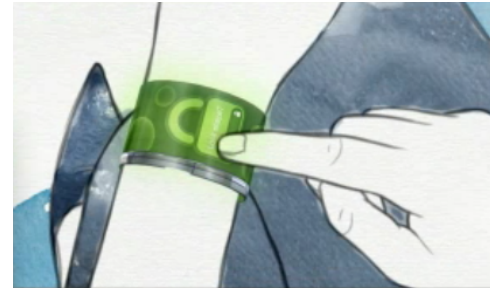
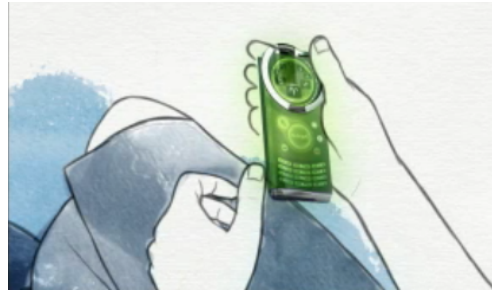
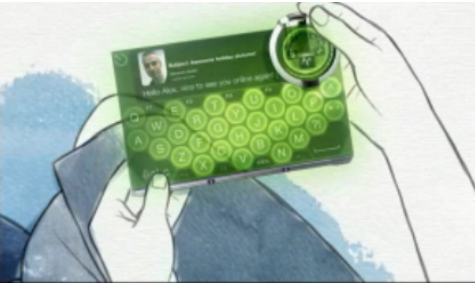
- **Electrodes exposed to local river water (Tolka)**
- **‘Slime test’ shows biofilm formation happens almost immediately and grows rapidly**

Control of membrane interfacial exchange & binding processes

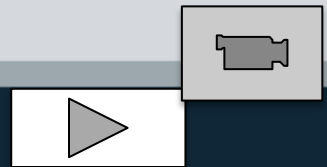


Remote, autonomous chemical sensing is a tricky business!

Nokia 'Morph' Concept



<http://research.nokia.com/morph>






INTEL....


Up to 1,200 hi-tech sensors from Intel are being placed on lamp-posts, drains, public buildings and other facilities to start zeroing in on "quality of life" issues for Dubliners.....

The sensors will host new technology in the form of Intel 'Quark' chips, which will record sensitive environmental information and feed it back to computers, tablets and smartphones operated by Dublin City Council.


Network of hi-tech sensors to monitor capital's 'quality of life'

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HOW IT WORKS



1. Small sensors will be placed on a street lamppost, drain or public building
2. The sensors will monitor air quality, noise levels, temperature and other things
3. The data collected by the sensors (pictured) will be wirelessly (probably by phone signal) sent back to Dublin City Council headquarters, where it will pop up on a map of Dublin on a PC
4. If there's a problem somewhere, an alert will go off highlighting where the problem is on the map
5. Dublin City Council then sends someone out to assess the problem (eg a blocked drain or a nightclub making too much noise)



An app called Citywatch allows the public to inform the council of a problem



**Dermot Diamond, Anal. Chem., 76 (2004) 278A-286A
(Ron Ambrosio & Alex Morrow, IBM TJ Watson)**

Sensor Hierarchy and Deployment Scenarios

- **Physical Transducers –low cost, reliable, low power demand, long life-time**

- Thermistors (temperature), movement, location, power,, light level, conductivity, flow, sound/audio,

- **Chemical Sensors – more complicated, need regular calibration, more costly to implement**

- Electrochemical, Optical, .. For metal ions, pH, organics...

- **Biosensors – the most challenging, very difficult to work with, die quickly, single shot (disposable) mode**

- Due to the delicate nature of enzymes, antibodies....

Increasing difficulty

- **Gas/Air Sensing – easiest to realise**

- Reliable sensors available, relatively low cost
- Integrate into platforms, develop IT infrastructure, GIS tools, Cloud Computing

- **On-land Water/ Monitoring**

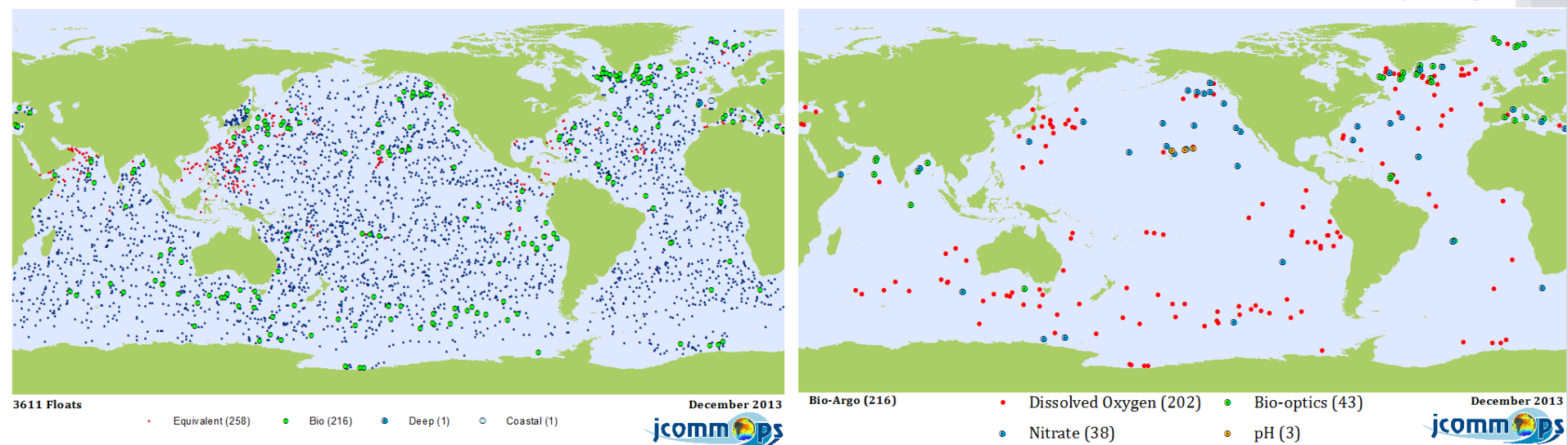
- More accessible locations
- Target concentrations tend to be higher
- Infrastructure available

- **Marine Water**

- Challenging conditions
- Remote locations & Limited infrastructure
- Concentrations tend to be lower and tighter in range



Argo Project (accessed March 9 2014)



- Ca. 3,600 floats: temperature and salinity
 - Only 216 reporting chem/bio parameters (ca. 6%)
 - Of these nitrate (38), DO (202), Bio-optics (43), pH (3) @€60K ea!
- DO is by Clark Cell (Sea Bird Electronics) or Dynamic fluorescence quenching (Aanderaa)
- See <https://picasaweb.google.com/JCOMMOPS/ArgoMaps?authuser=0&feat=embedwebsite>

‘calibration of the DO measurements by the SBE sensor remains an important issue for the future’, Argo report ‘Processing Argo OXYGEN data at the DAC level’, September 6, 2009, V. Thierry, D. Gilbert, T. Kobayashi

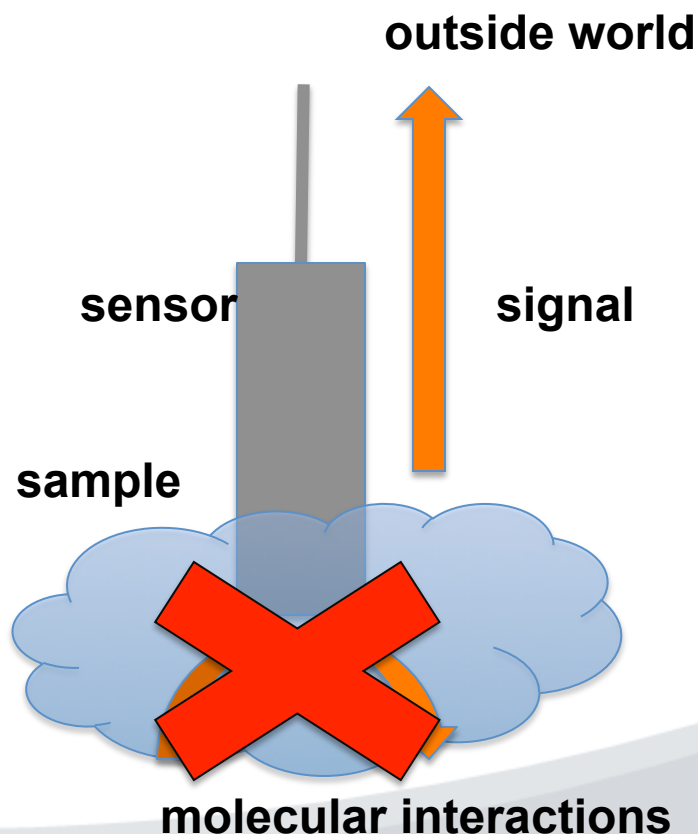


**After decades of intensive research,
our capacity to deliver successful
long-term deployments of chemo/bio-
sensors in remote locations is still
very limited**

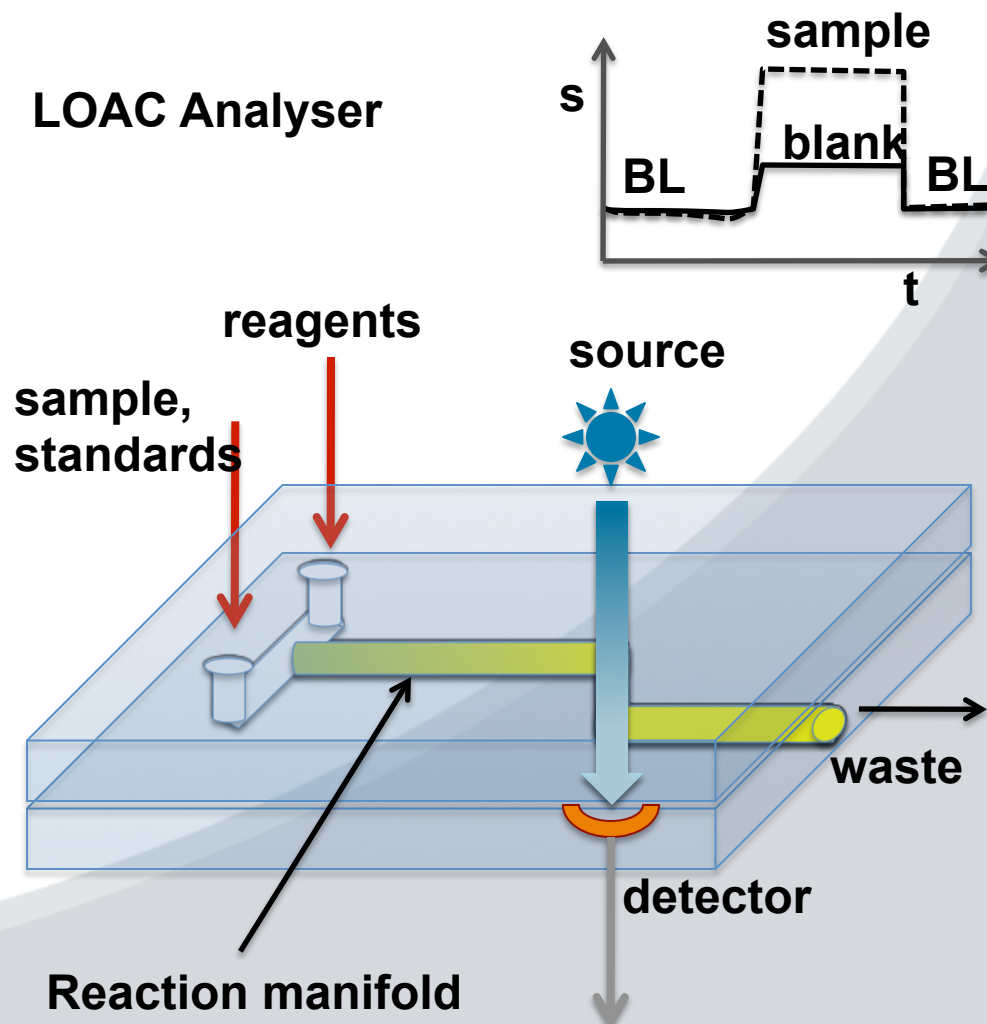


Direct Sensing vs. Reagent Based LOAC/ufluidics

Direct Sensing



LOAC Analyser

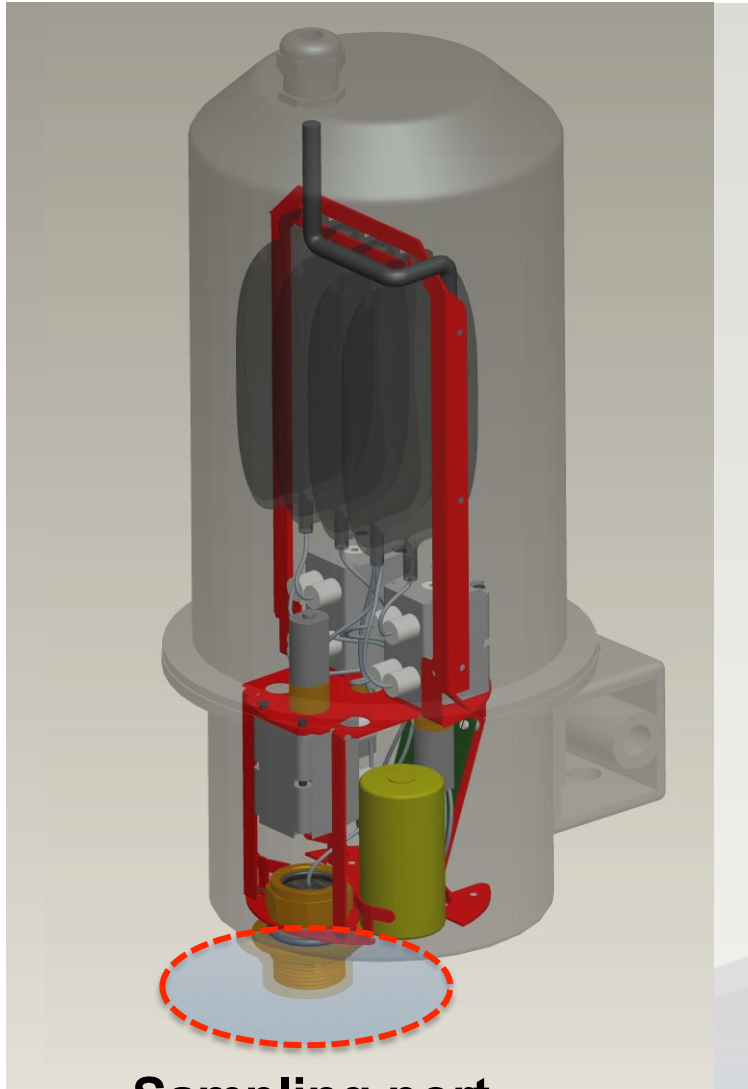


Many people, myself included, expected that the ability to manipulate fluid streams, in microchannels, easily, would result in a proliferation of commercial LoC systems, and that we would see applications of these devices proliferating throughout science. In fact, it has not (yet) happened.

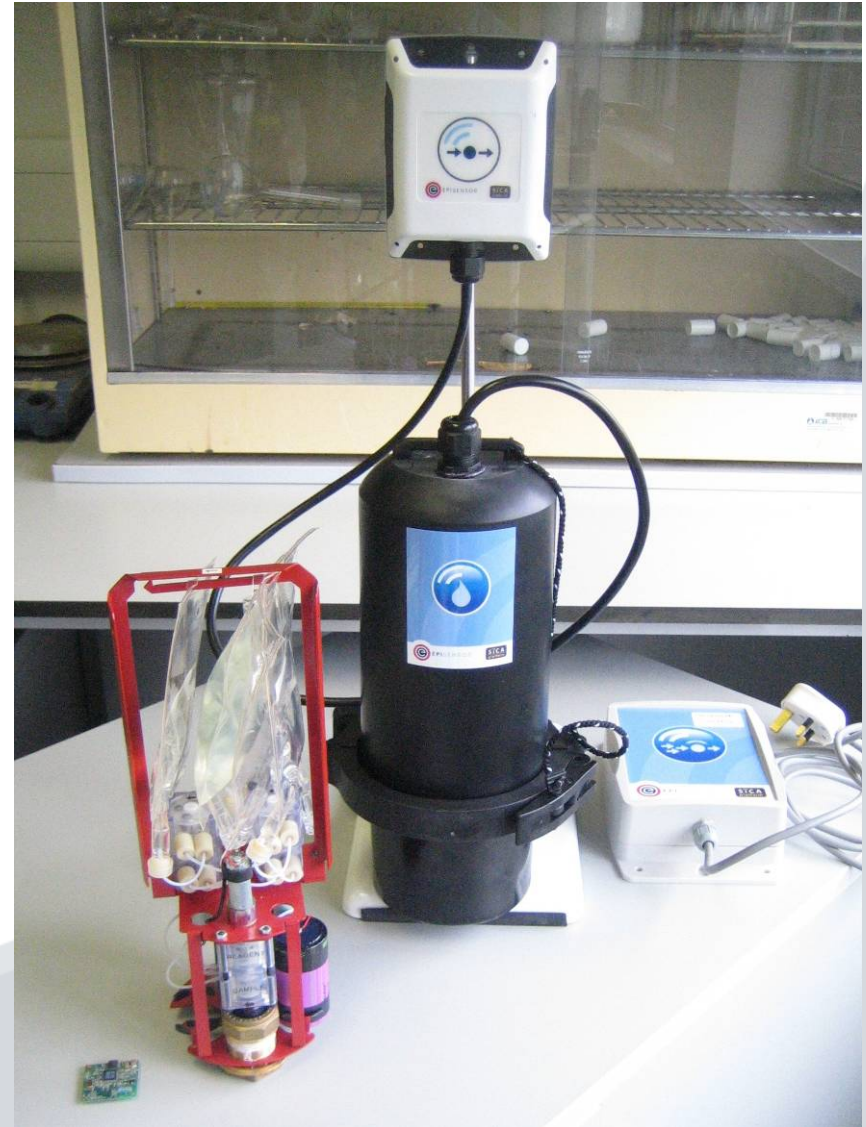
Microfluidics, to date, has been largely focused on the development of science and technology, and on scientific papers, rather than on the solution of problems

Editorial 'Solving Problems', George Whitesides,
Lab Chip 10 (2010) 2317-2318

2nd Generation Analyser: Design



Sampling port



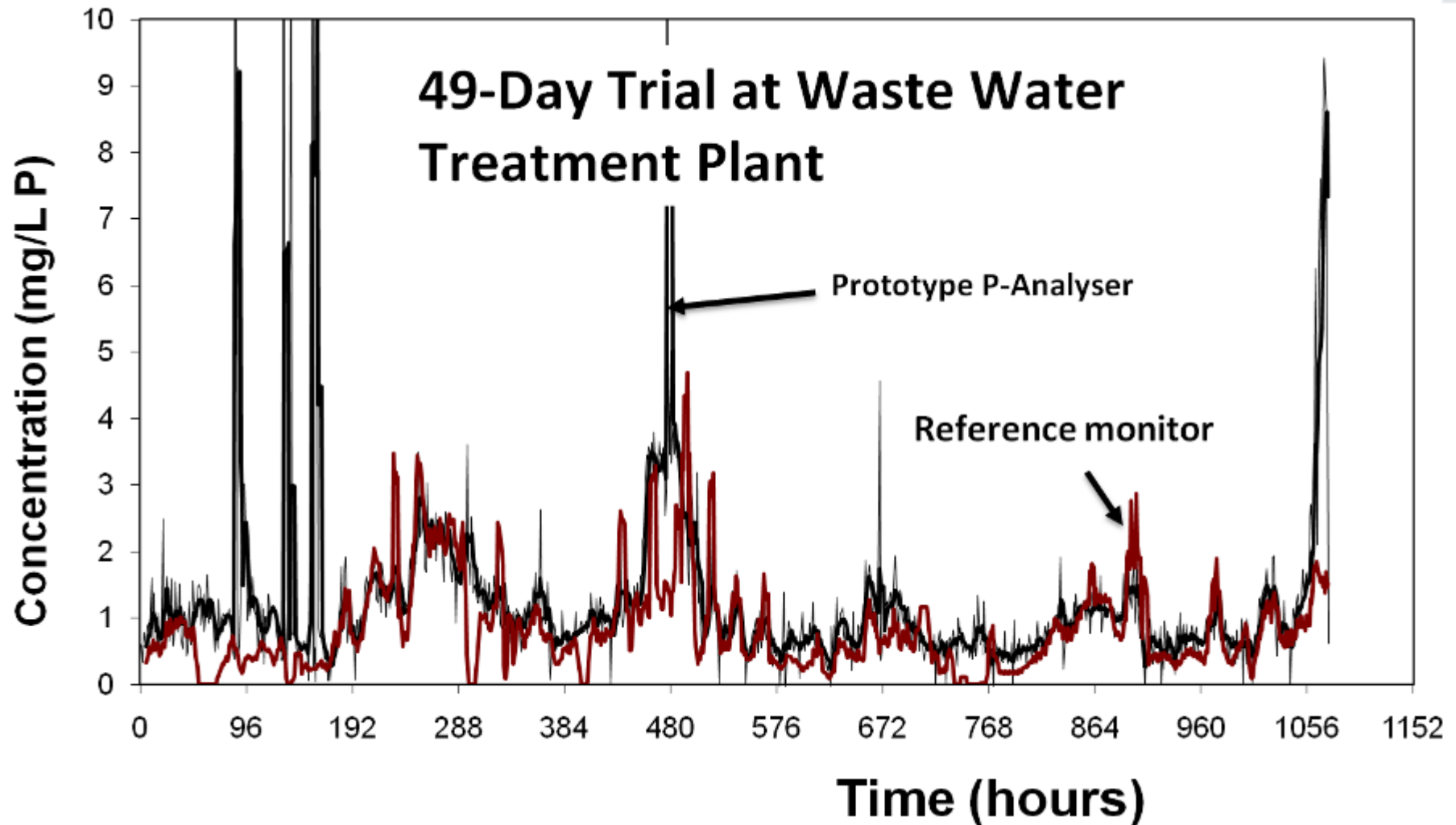
Deployment at Osberstown WWTP



- **Phosphate monitoring unit deployed**
- **System is fully immersed in the treatment tank**
- **Wireless communications unit linked by cable**
- **Data transmitted to web**



Autonomous Chemical Analyser



Osberstown – 3 week deployment

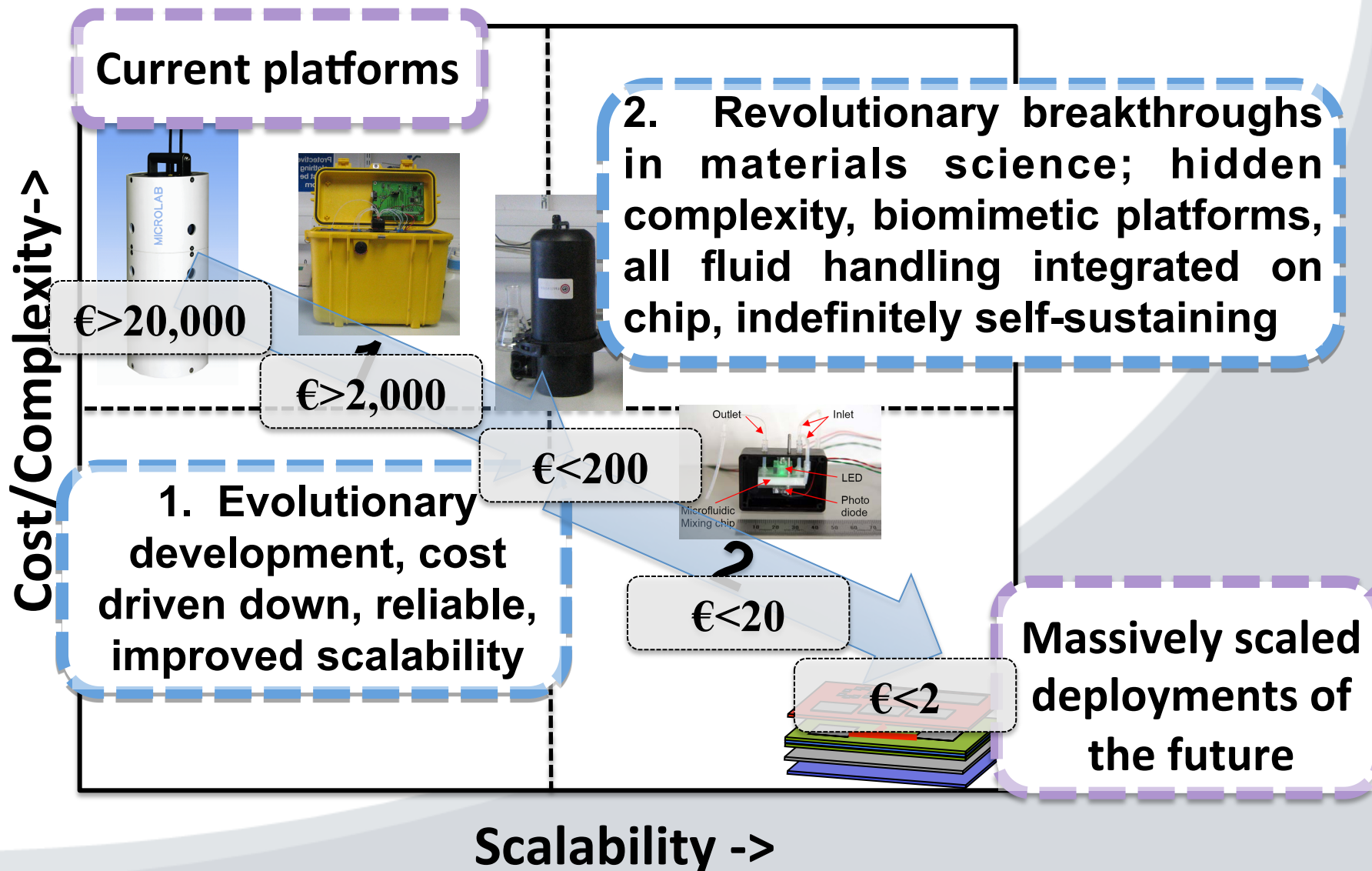


Osberstown – 3 week deployment

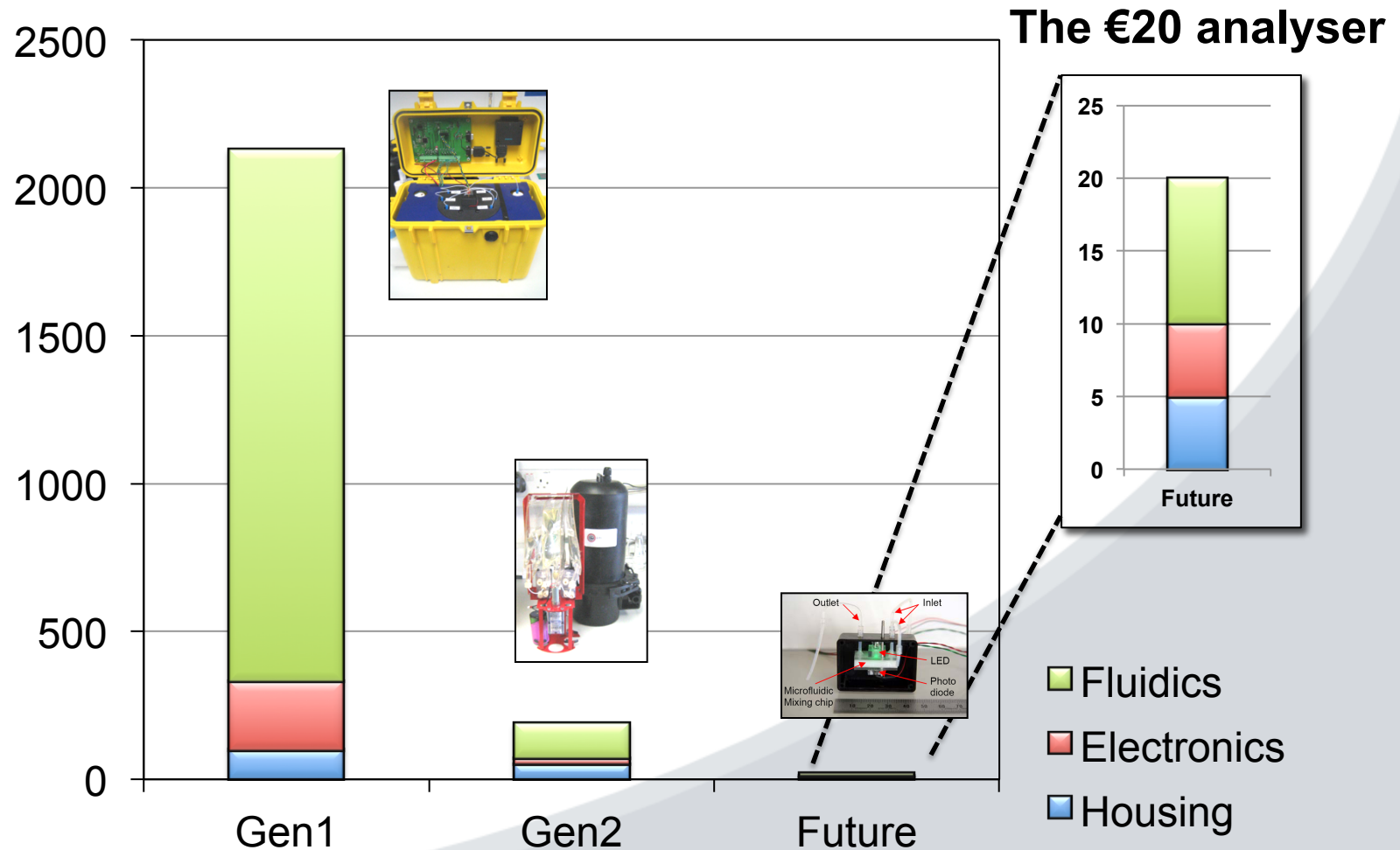


Biofouling of sensor surfaces is a major challenge for remote chemical sensing – both for the environment and for implantable sensors

Achieving Scale-up

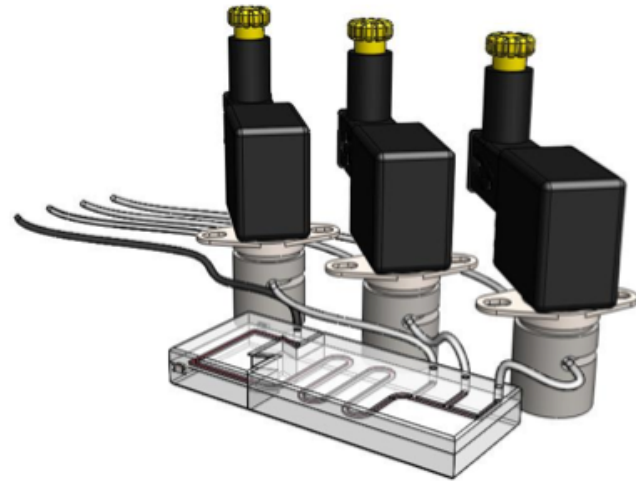


Cost Comparison Analyser (€)



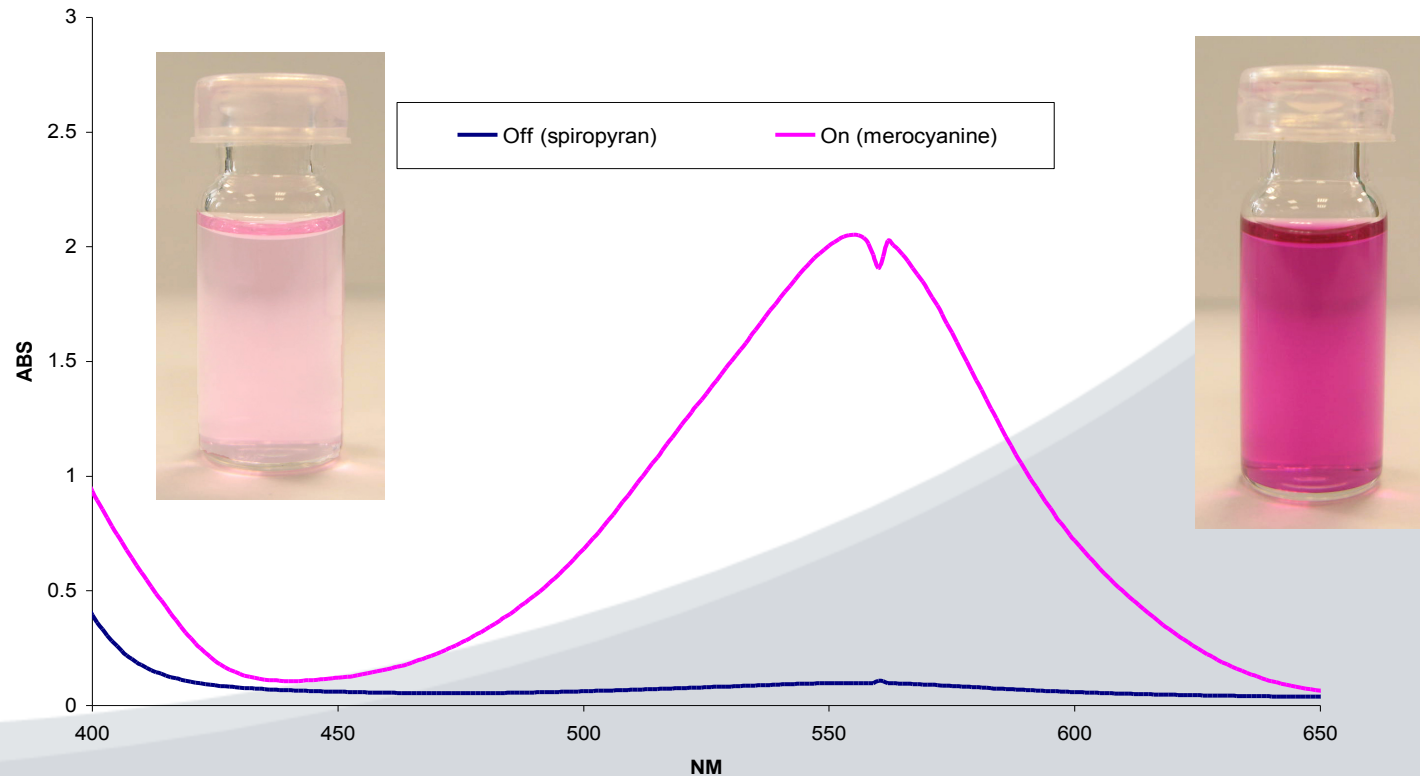
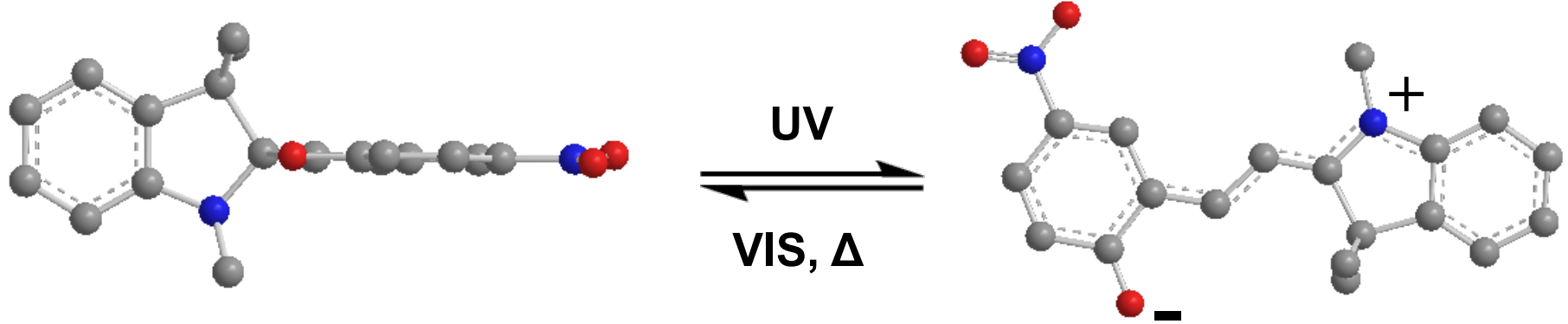
How to advance fluid handling in LOC platforms: re-invent valves (and pumps)!

- **Conventional valves cannot be easily scaled down - Located off chip: fluidic interconnects required**
 - Complex fabrication
 - Increased dead volume
 - Mixing effects
- **Based on solenoid action**
 - Large power demand
 - Expensive



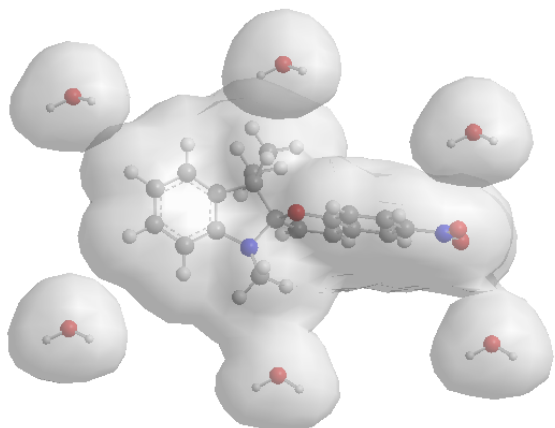
Solution: soft-polymer (biomimetic) valves fully integrated into the fluidic system

Photoswitchable Materials



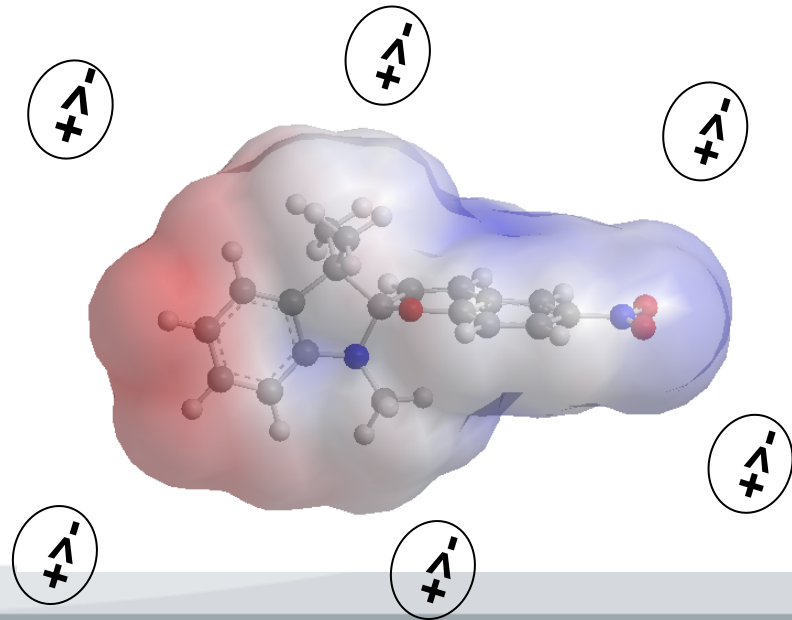
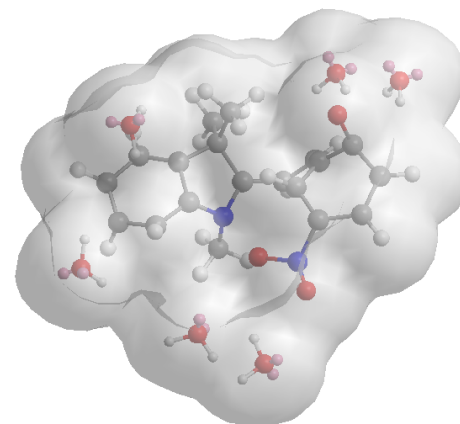
Non-specific (BSP) and specific (MC) Interactions

BSP

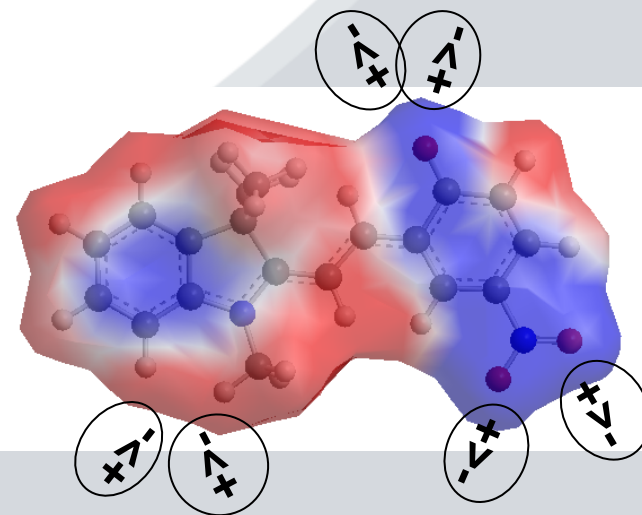


H₂O

MC



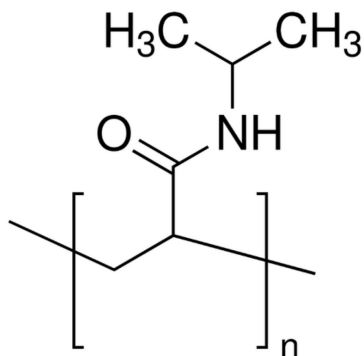
Charged species



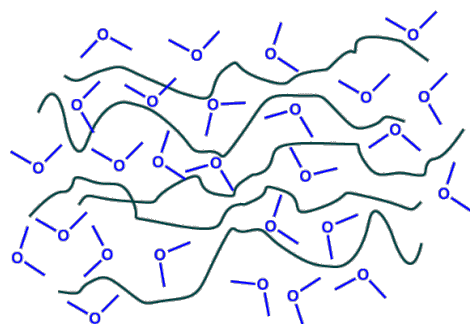
Poly(*N*-isopropylacrylamide)

- pNIPAAm exhibits inverse solubility upon heating
- This is referred to as the LCST (Lower Critical Solution Temperature)
- Typically this temperature lies between 30-35°C, but the exact temperature is a function of the (macro)molecular microstructure
- Upon reaching the LCST the polymer undergoes a dramatic volume change, as the hydrated polymer chains collapse to a globular structure, expelling the bound water in the process

pNIPAAm



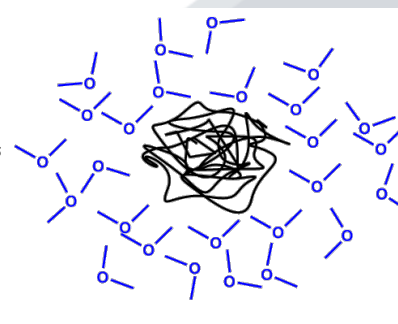
Hydrophilic



Hydrated Polymer Chains



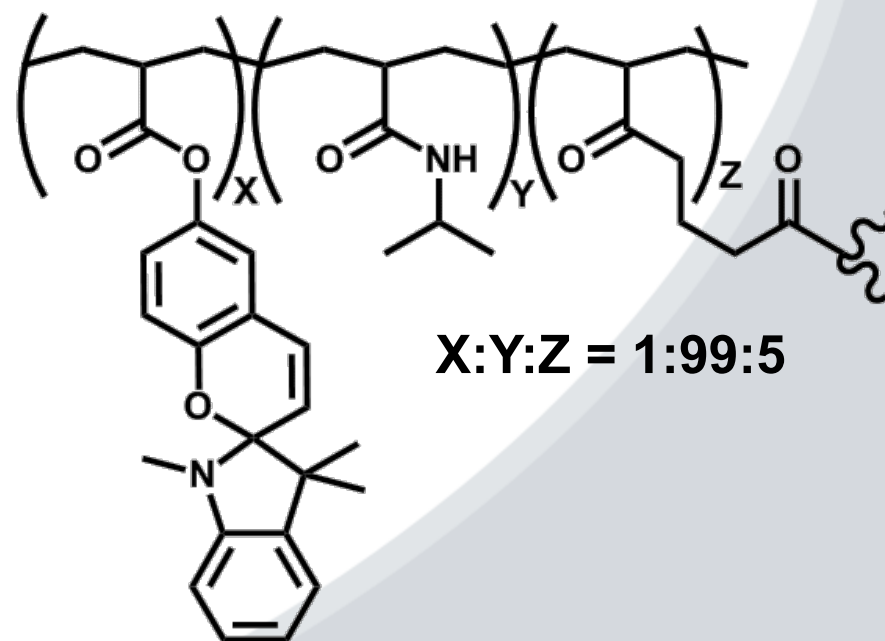
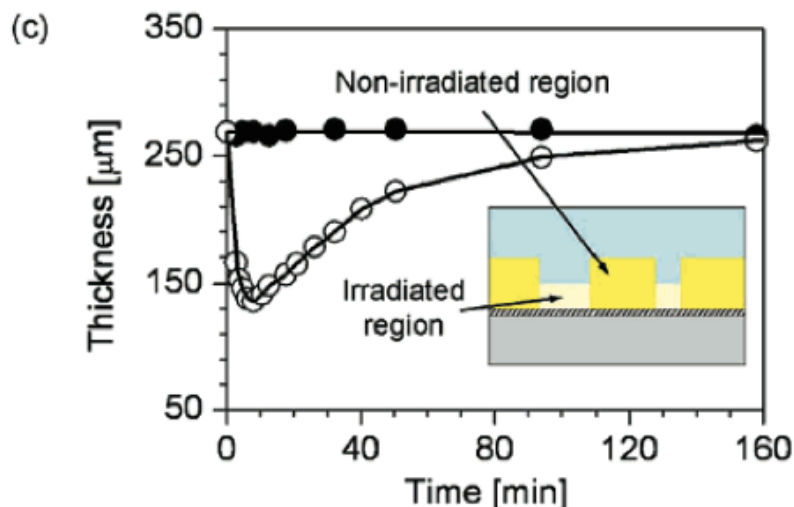
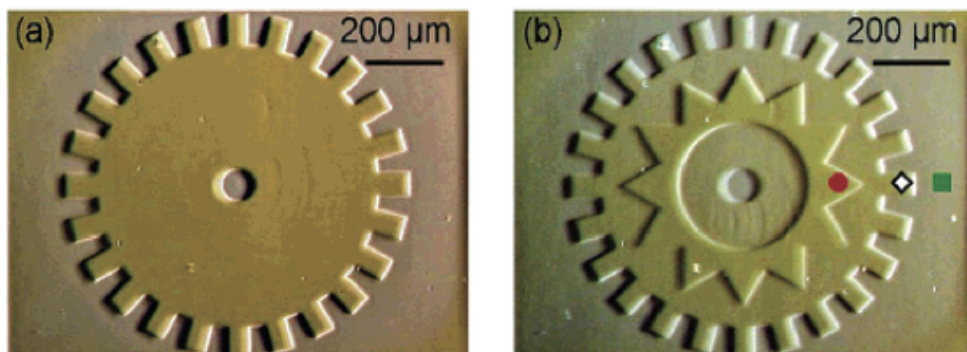
Hydrophobic



Loss of bound water
-> polymer collapse



Polymer based photoactuators based on pNIPAAm



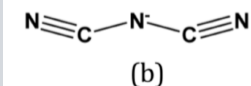
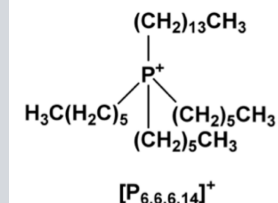
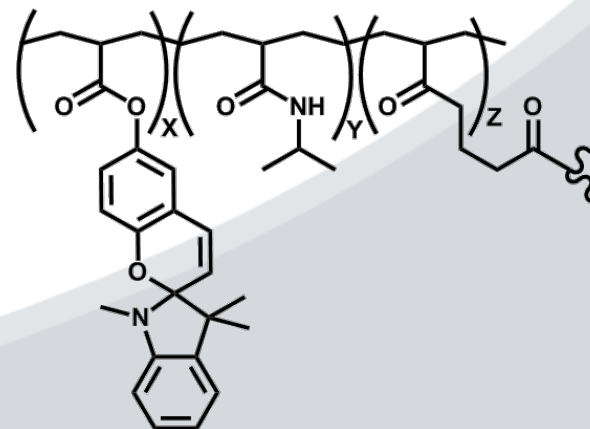
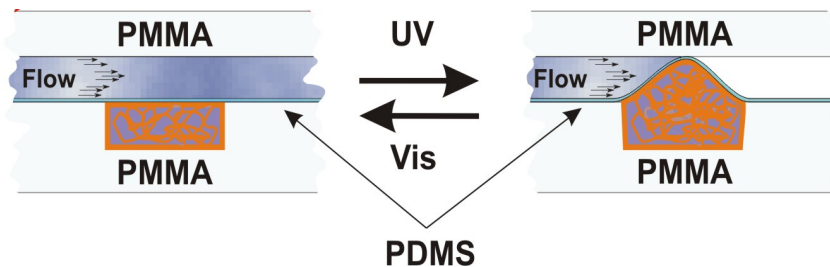
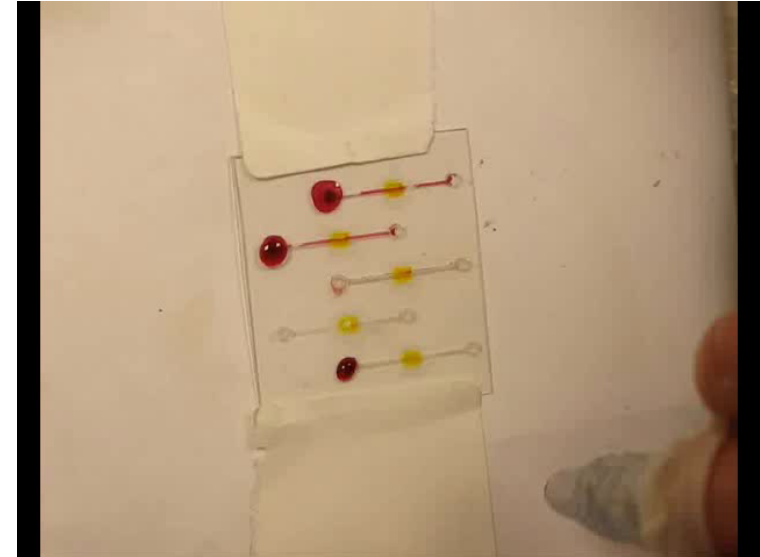
poly(N-isopropylacrylamide) (PNIPAAm)
Formulation as by Sumaru et al¹

1) *Chem. Mater.*, 19 (11), 2730 -2732, 2007.

Figure 3. (a, b) Images of the pSPNIPAAm hydrogel layer just after the micropatterned light irradiation. Duration of irradiation was (●, red) 0, (◇) 1, and (■, green) 3 s. (c) Height change of the hydrogel layer in (●) non-irradiated and (○) irradiated region as a function of time after 3 s blue light irradiation.



Photo-actuator polymers as microvalves in microfluidic systems

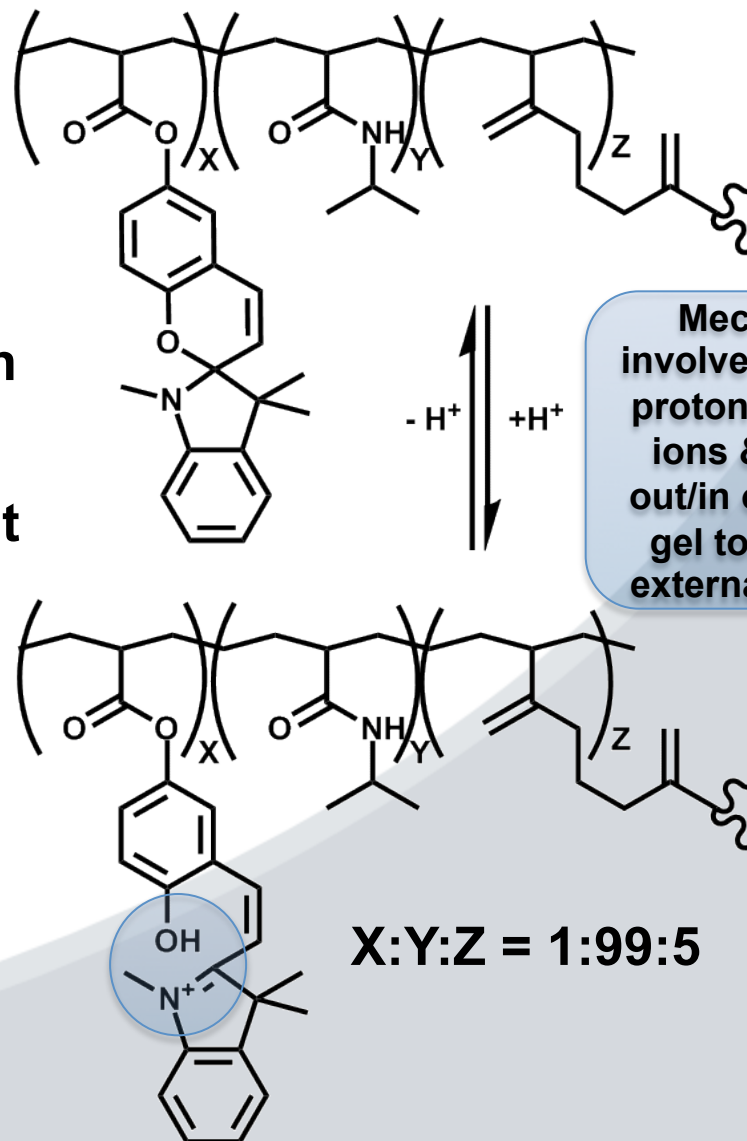
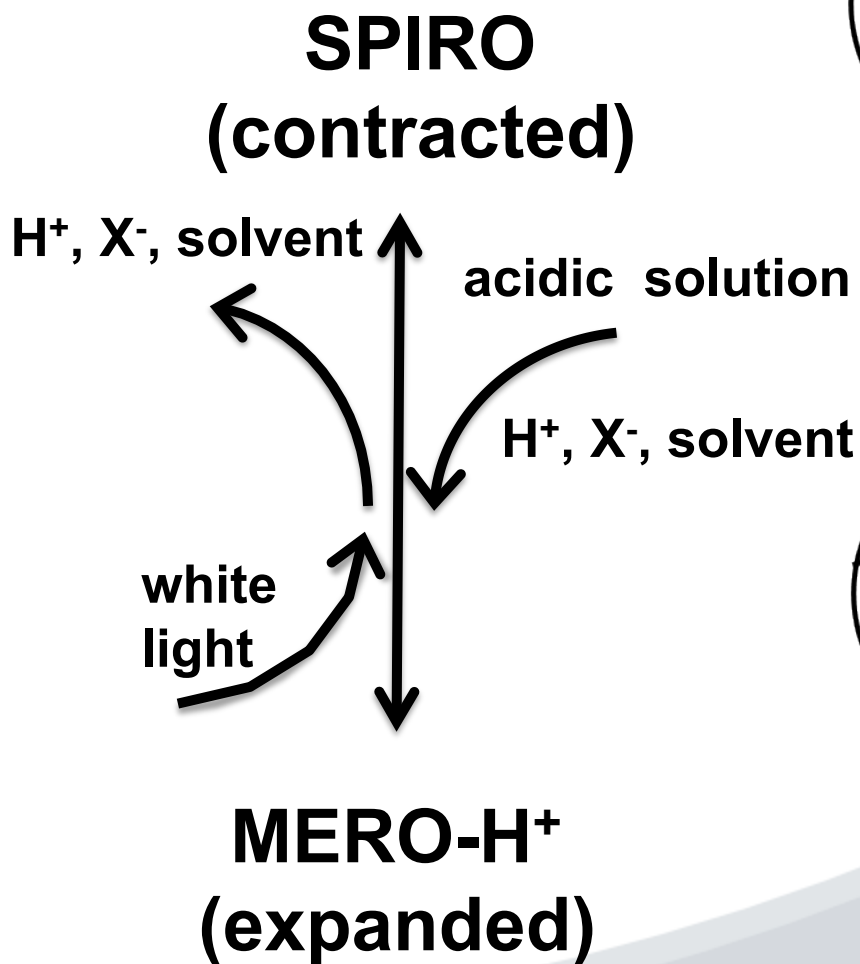


trihexyltetradecylphosphonium
dicyanoamide $[\text{P}_{6,6,6,14}]^+[\text{dca}]^-$

Ionogel-based light-actuated valves for controlling liquid flow in micro-fluidic manifolds, Fernando Benito-Lopez, Robert Byrne, Ana Maria Raduta, Nihal Engin Vrana, Garrett McGuinness, Dermot Diamond, Lab Chip, 10 (2010) 195-201.



Actuation Mechanism

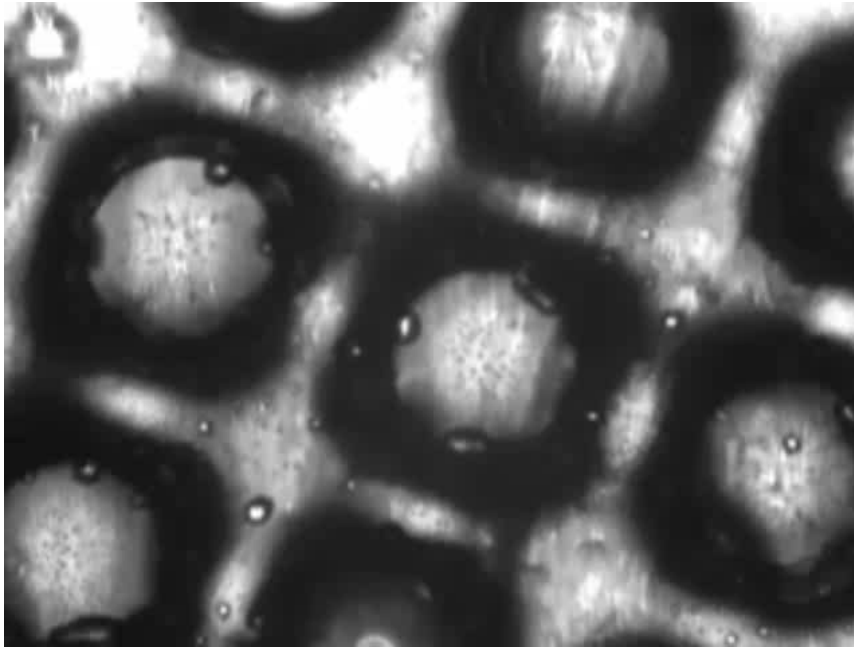


So far, so good: but what are the limitations?

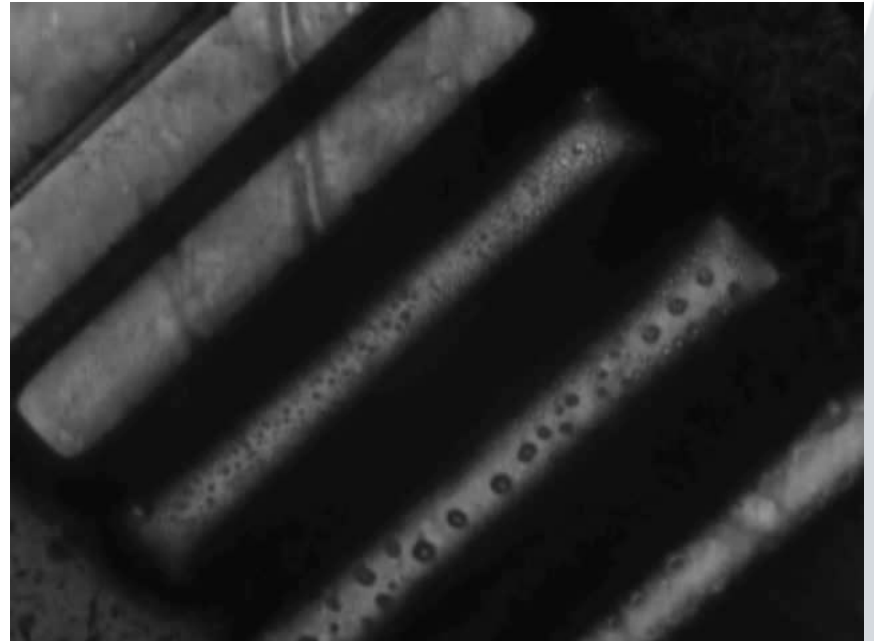
- Response time for re-swelling is slow - 10's of minutes due to diffusion mechanism
- Swelling requires protonation of the MC to MC-H^+ within the ionogel by the external bathing solution
- These issues more or less limit the applicability of the valves to single use



Build Dynamic Structures within Microchannels



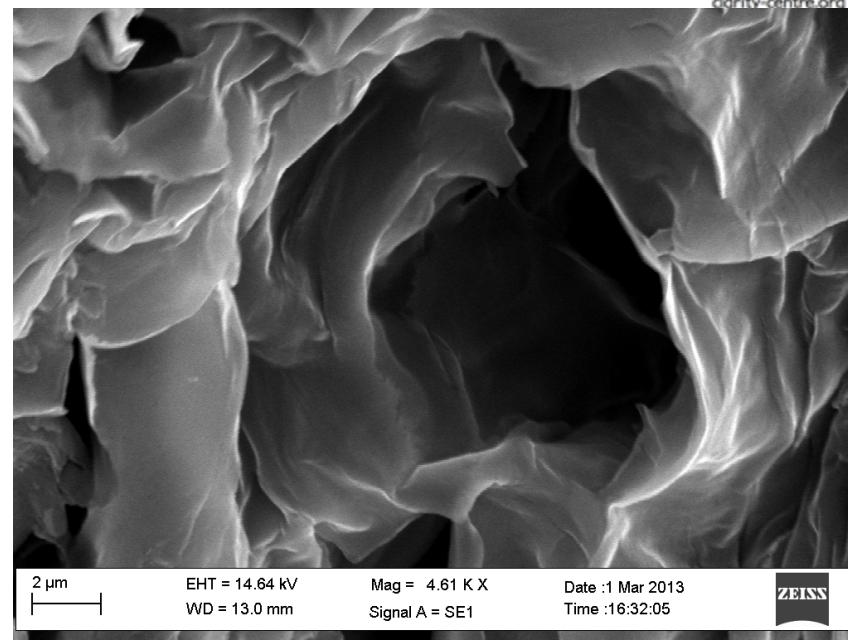
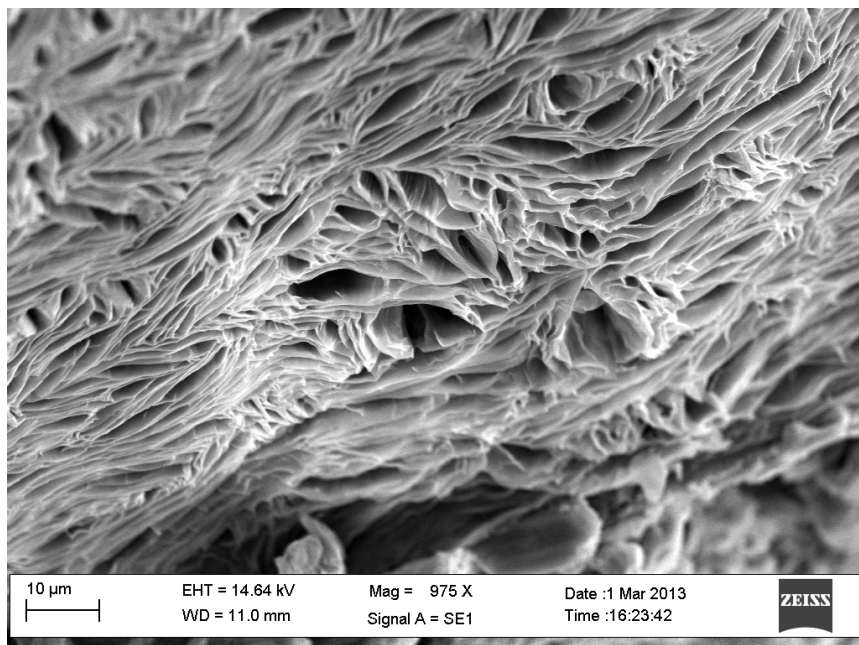
Ntf2 pillars speed x3



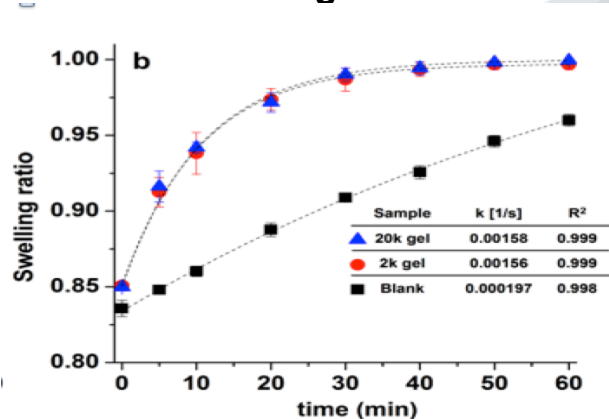
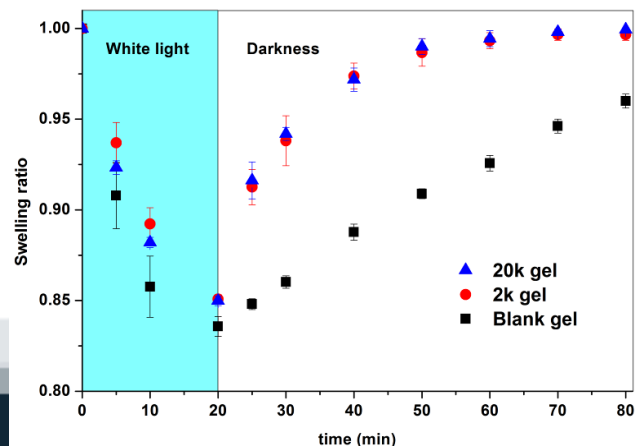
DCA lines speed x4

With Dr Peer Fischer, Fraunhofer-Institut für Physikalische Messtechnik (IPM), Freiburg

Porous Gels



Highly porous pNIPAAm gel structures generated using PEG as the porogen. This dramatically increases the surface area to bulk ratio, reducing the diffusion pathlength for water to penetrate to the gel interior, which in turns results in faster swelling/contraction rates



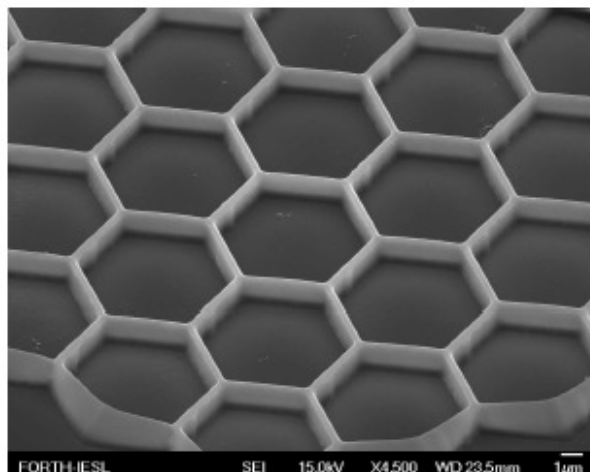
On the re-swelling side; highly porous gels now recover ca. an order of magnitude faster;

$k = 1.6 \times 10^{-3} \text{ S}^{-1}$
vs. $2.0 \times 10^{-4} \text{ S}^{-1}$

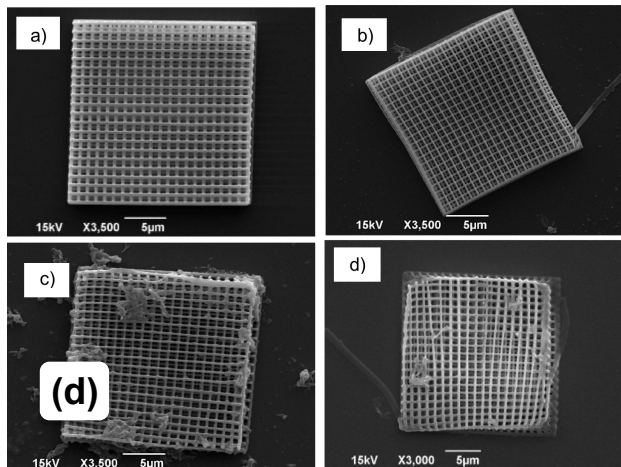


Reduce scale – increase response time

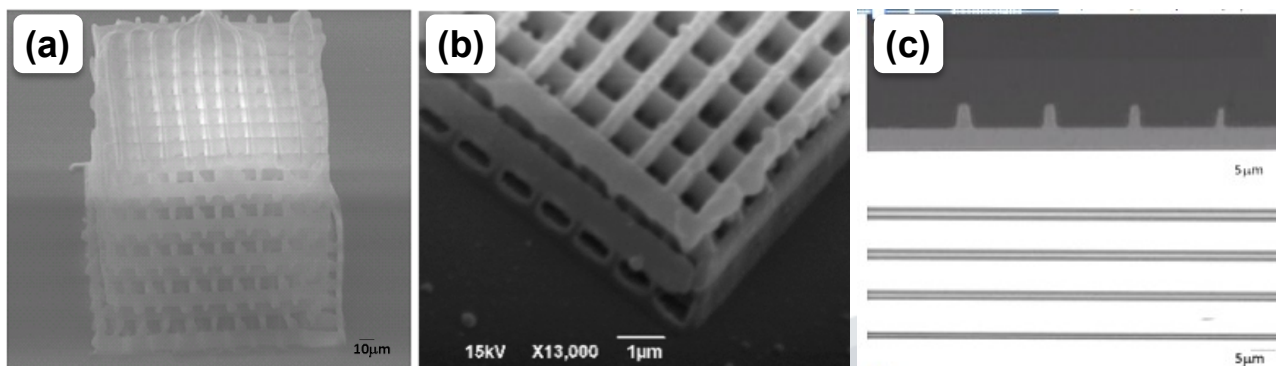
3-d Spiro-doped sol-ionogels



SEM of surface patterning produced by multi-photon polymerisation of hybrid graphene-doped ionogels



SEM images of woodpiles fabricated from material D containing a) 0%, b) 20%, c) 40% and d) 50% IL



Two photon polymerised (2PP), patterned ionogels (a) and (b), and (c) feature resolution down to 150 nm or less; (d) spiropyran co-polymerised in a gel 'woodpile' structure.

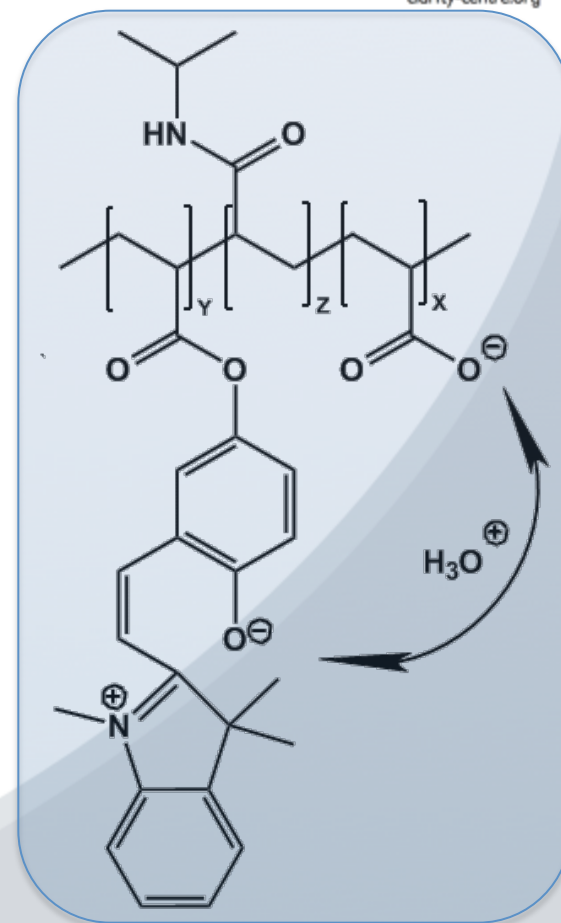
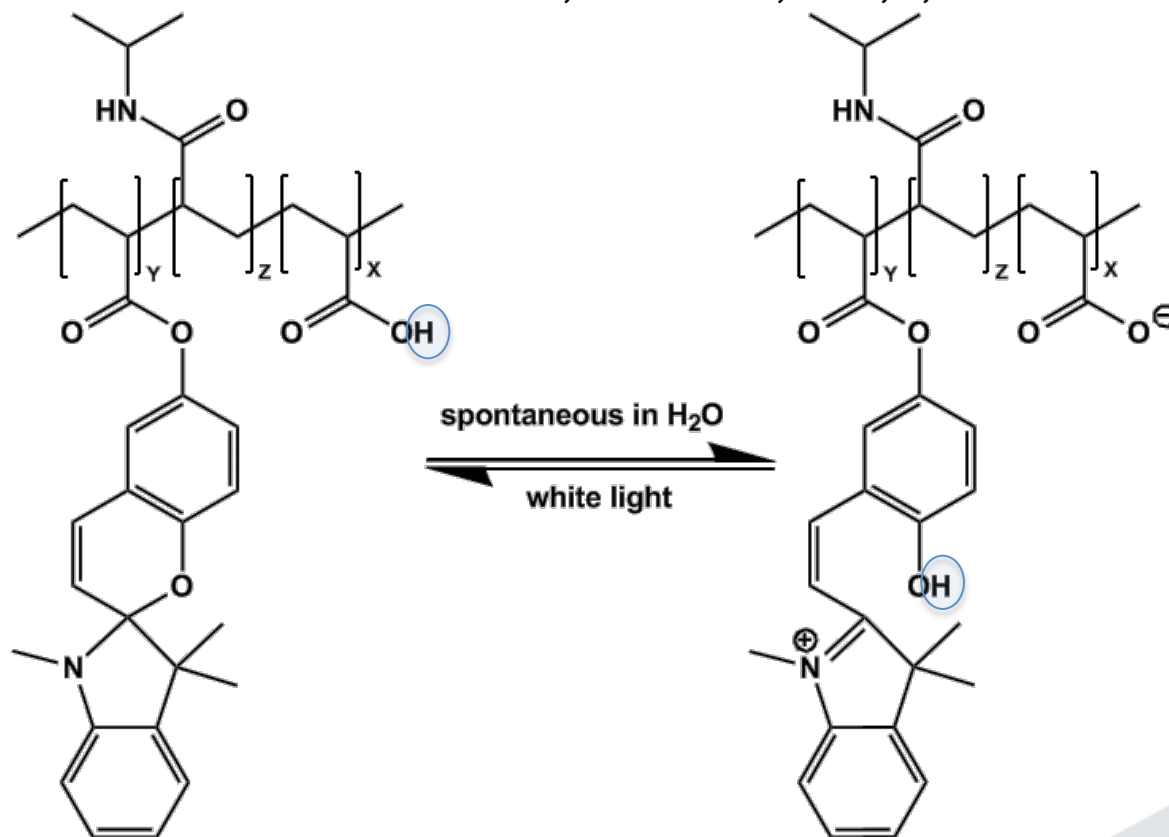
The ionogels were based on photo-curable silicato-zirconate hybrid sol-gel materials and phosphonium (trihexyltetradecylphosphonium dicyanamide [$P_{6,6,6,14}$] [DCA] ionic liquid (IL)). To optimise the dispersion of graphene within the ionogel matrices, aqueous solutions of graphene were prepared, as opposed to the conventional graphene powder approach, and employed as catalysts for the hydrolysis and condensation reactions occurring in the sol-gel process.

With Gabija Bickauskait and Maria Farsari, Institute of Electronic Structure and Laser, Foundation for Research and Technology Hellas, N. Plastira 100, GR-70013 Heraklion, Crete, Greece



Self protonating photoresponsive gel

Ziolkowski *et al.*, *Soft Matter*, 2013, 9, 8754–8760



Previously proton source was external (acidic soln. required)
Protons, counter ions & solvent diffuse into/out of the gel

Now the proton exchange is 'internalised'
The proton population is essentially conserved



Spontaneous Reformation of Acidified Merocyanine during Actuation Cycling in non-acidified water

Ziolkowski *et al.*, Soft Matter, 2013, 9, 8754–8760

Gel with 0 % AA

Colour gradually changing from yellow to purple as H^+ leaves the gel on each cycle

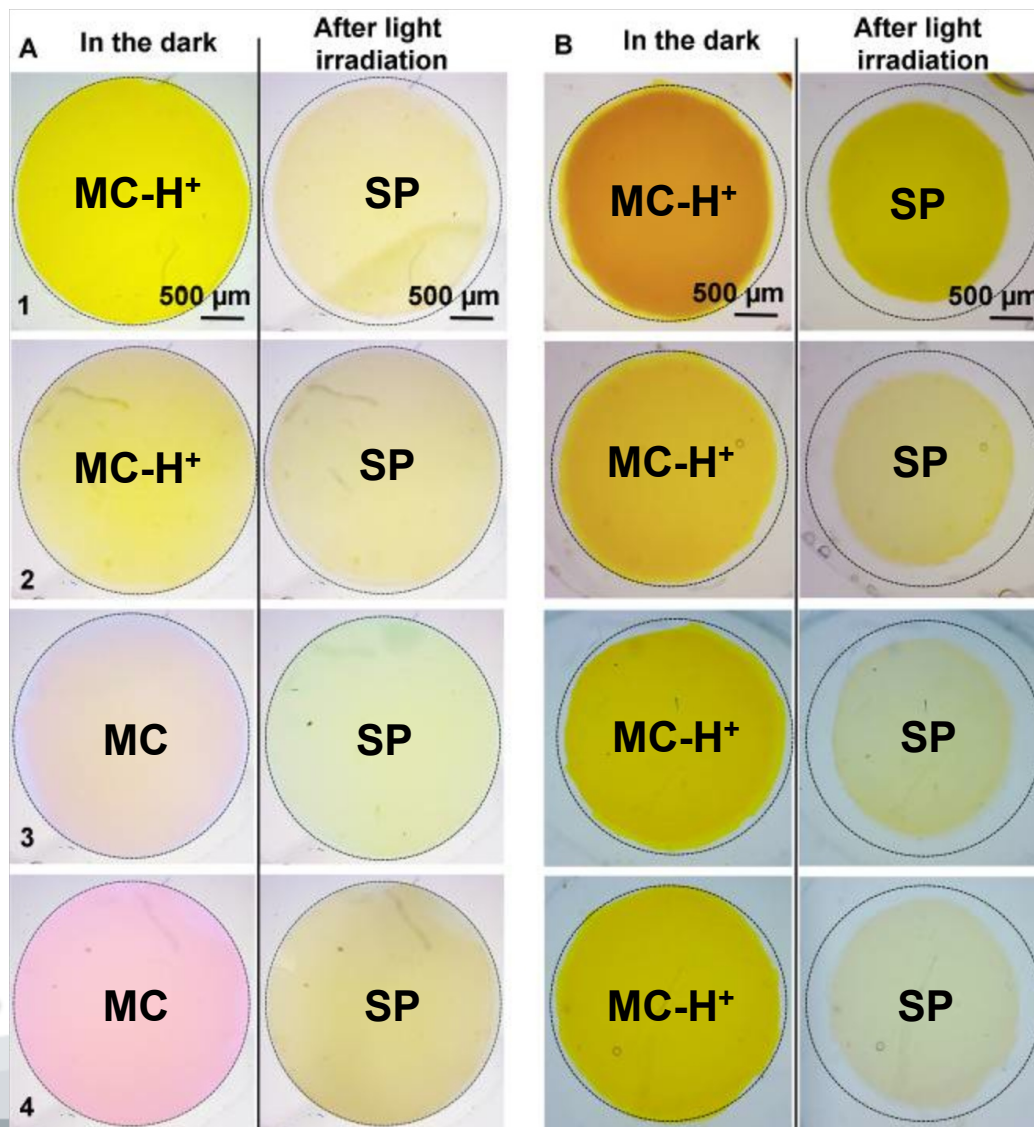
Switching changes from primarily

$MC-H^+ \rightarrow SP+H^+$

to

$MC \rightarrow SP$

Gel actuation stops



Gel with 5 % AA

Colour remains essentially the same, as H^+ stays in the gel during cycling

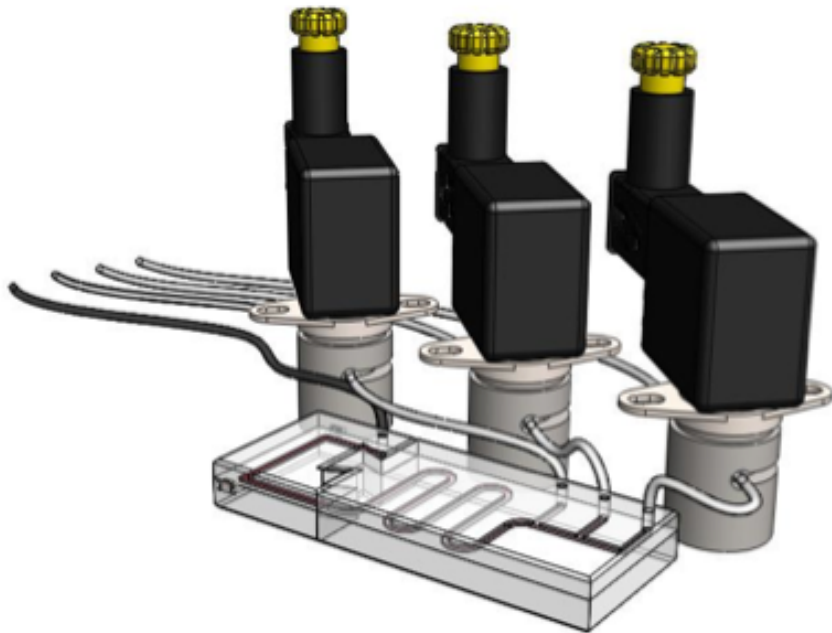
Switching stays primarily as

$MC-H^+ \rightarrow SP+H^+$

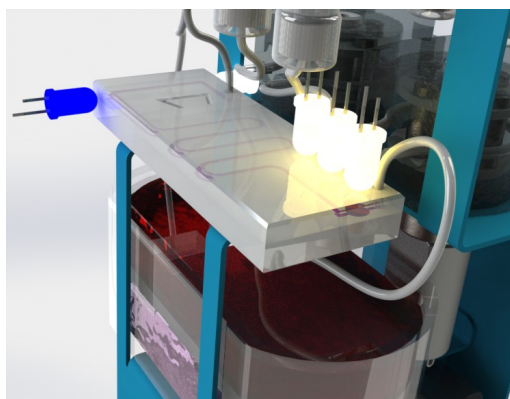
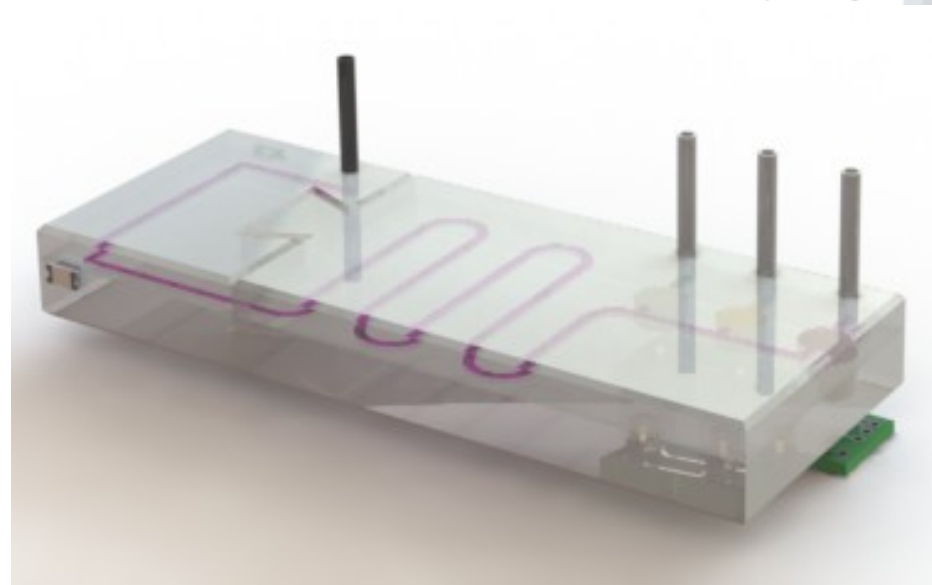
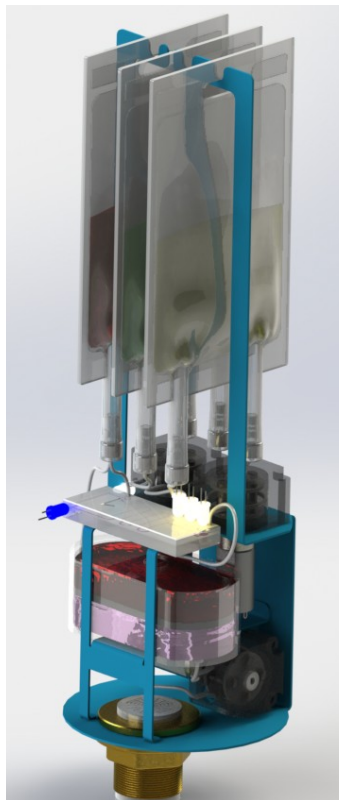
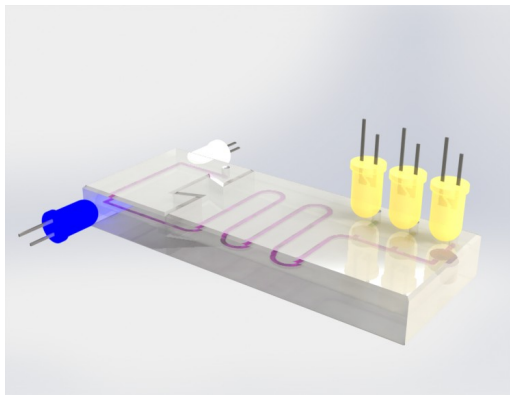
Gel actuation continues



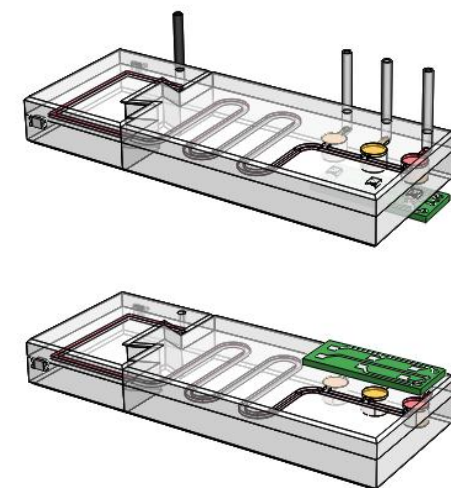
Can we go from this:



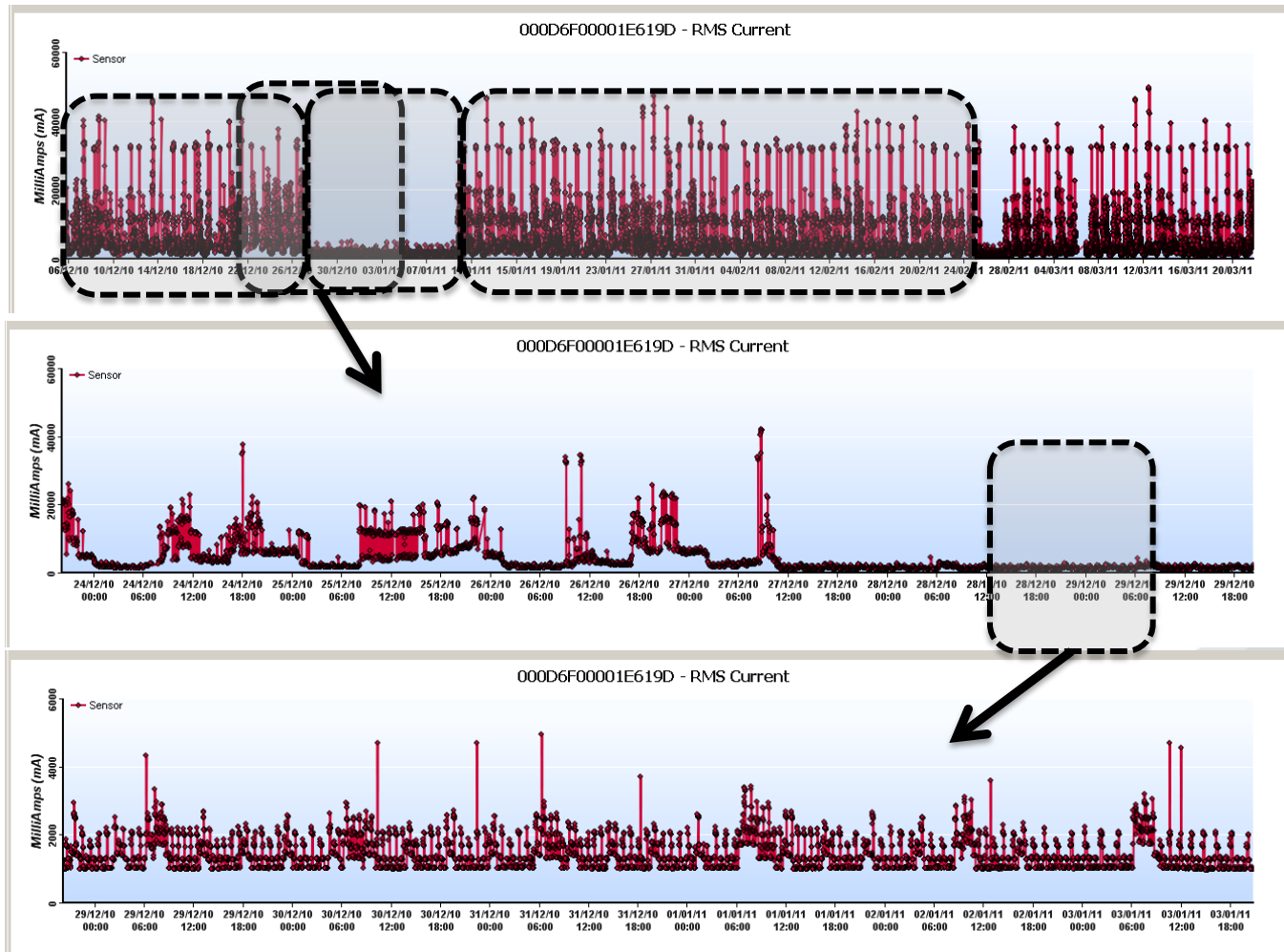
To Photo-Fluidics & Detection



- Fluidic handling completely integrated into the microfluidic chip
- Valves actuated remotely using light (LEDs)
- Detection is via LED colorimetric measurements



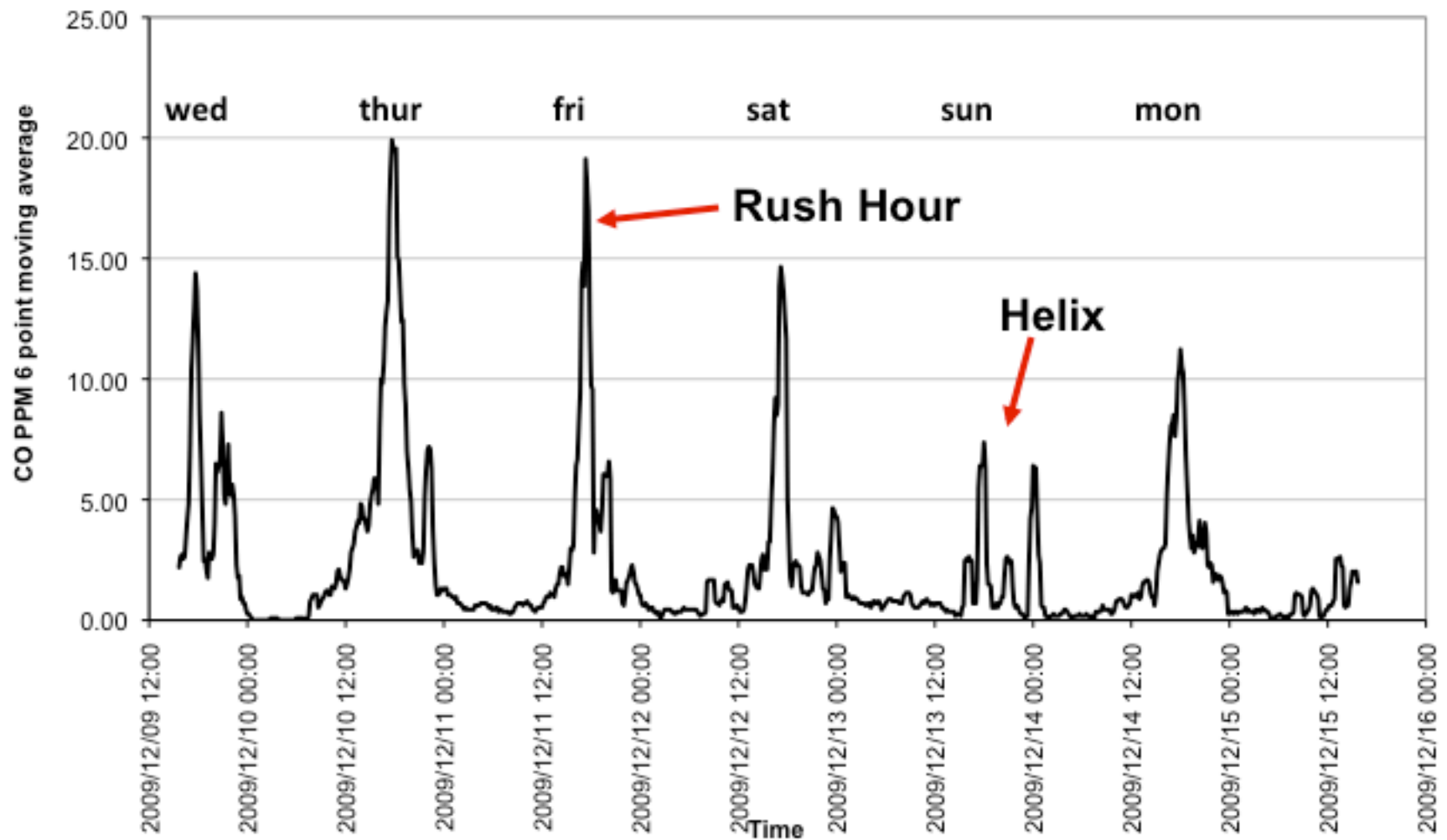
Who owns the data? Life Patterns from Home Metering



- Home/Away
- Repetitive Patterns
- Specific Appliances
- Totally unobtrusive
- Who owns the data?



CO Emissions at DCU Car Park,



Conclusions

- **Distributed sensor networks can provide tremendous benefits to society**
- **Significant advances in the performance of sensors for important chem/bio-targets are required**
- **This will happen provided there are strong links between fundamental materials science and applied research**
- **Issues around ethics, trust, privacy need to be addressed**



Thanks to.....

- Members of my research group
- NCSR, DCU
- CLARITY/INSIGHT
- Research Partners – academic and industry
- Funding sources – SFI, HEA, EI, MI, EPA, ARC, EU-FP7, IRCSET...



Thanks RSC for the invitation

