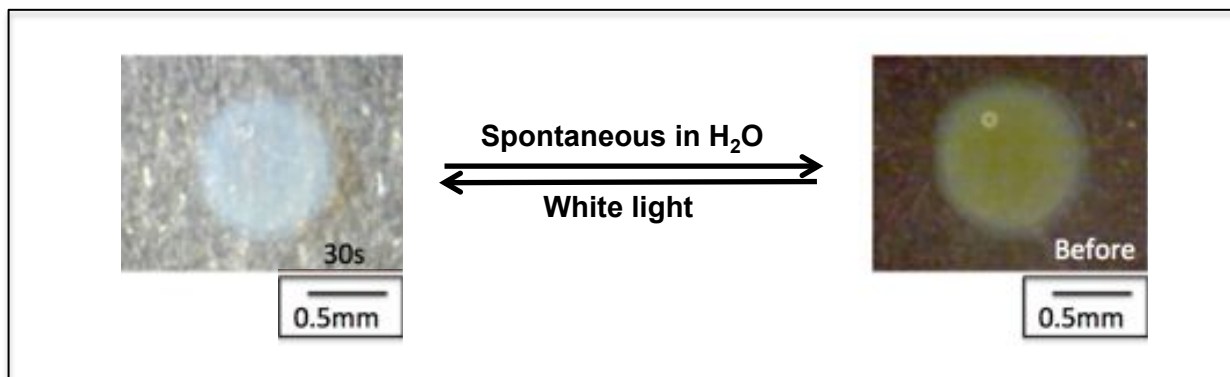
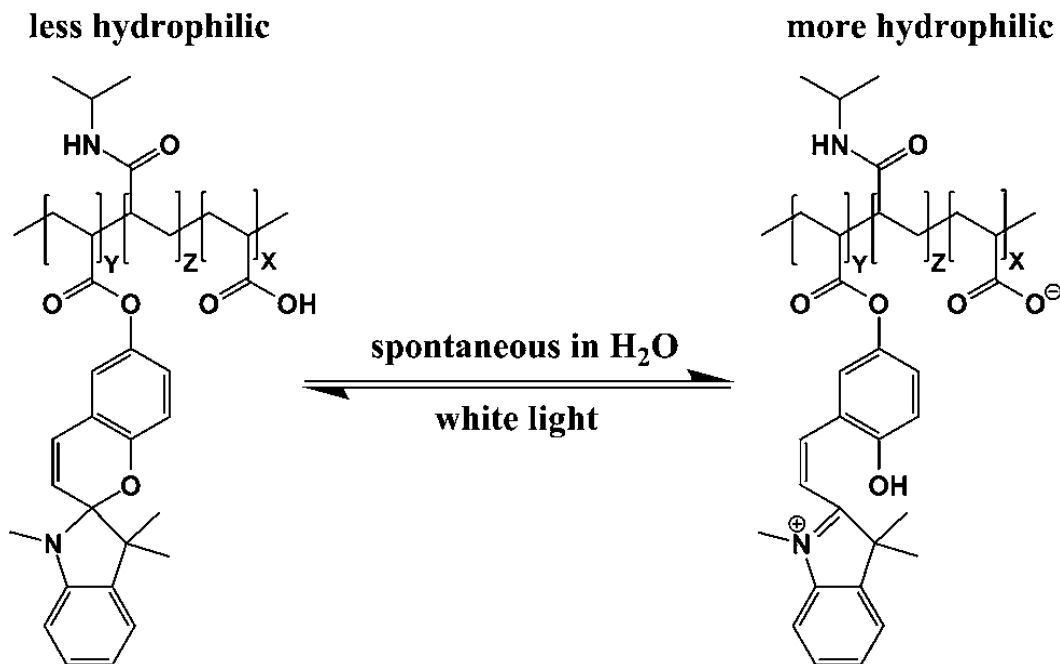


# Photo-responsive materials Capabilities and Perspectives

July 2014

Larisa Florea

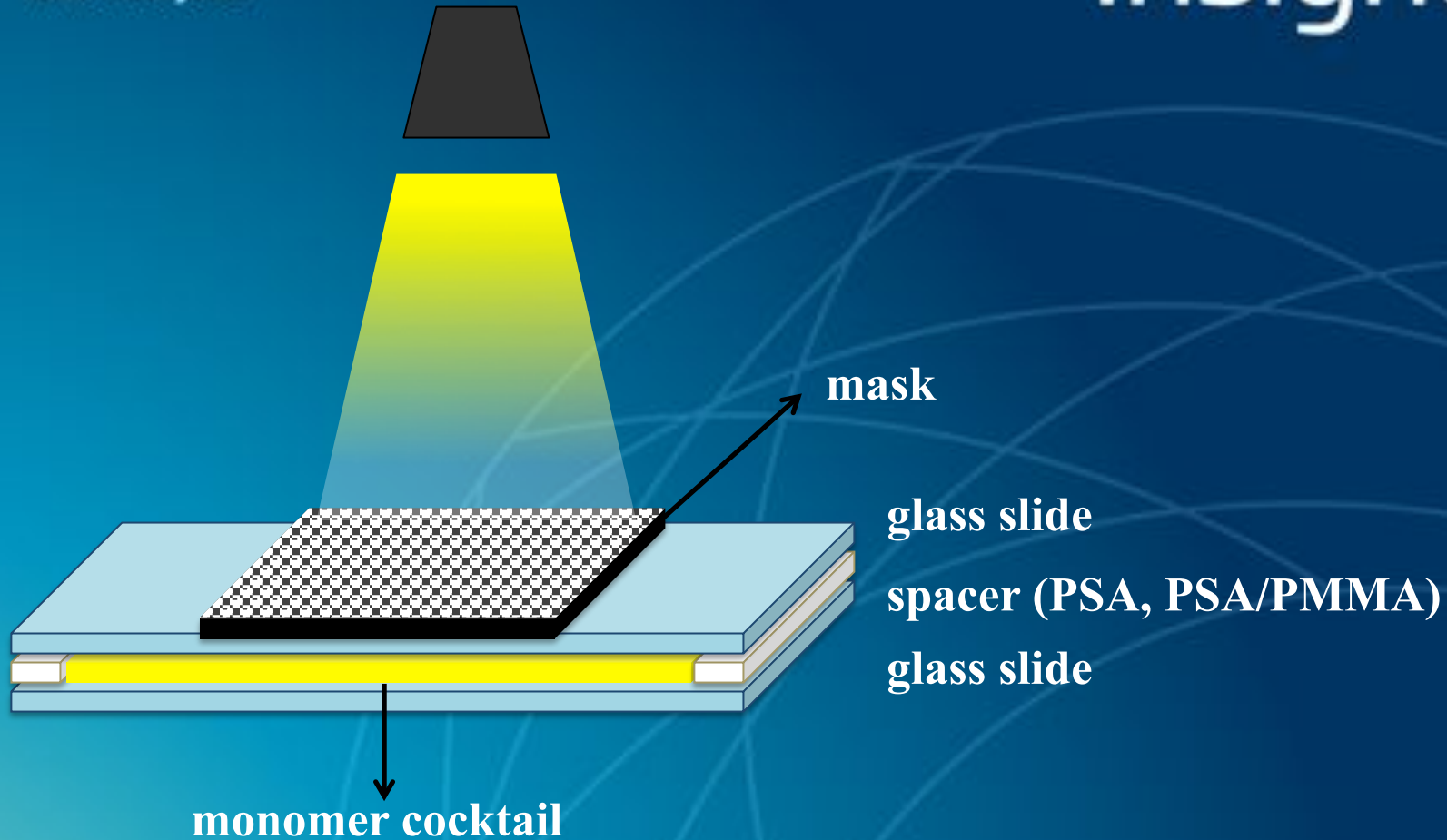
# Photo-responsive hydrogels



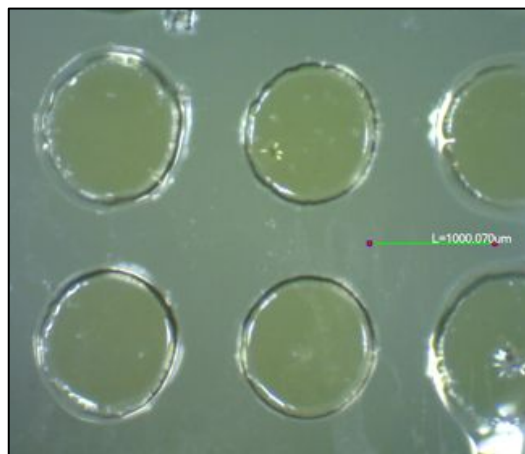
# Microstructures preparation

Centre for  
Data Analytics

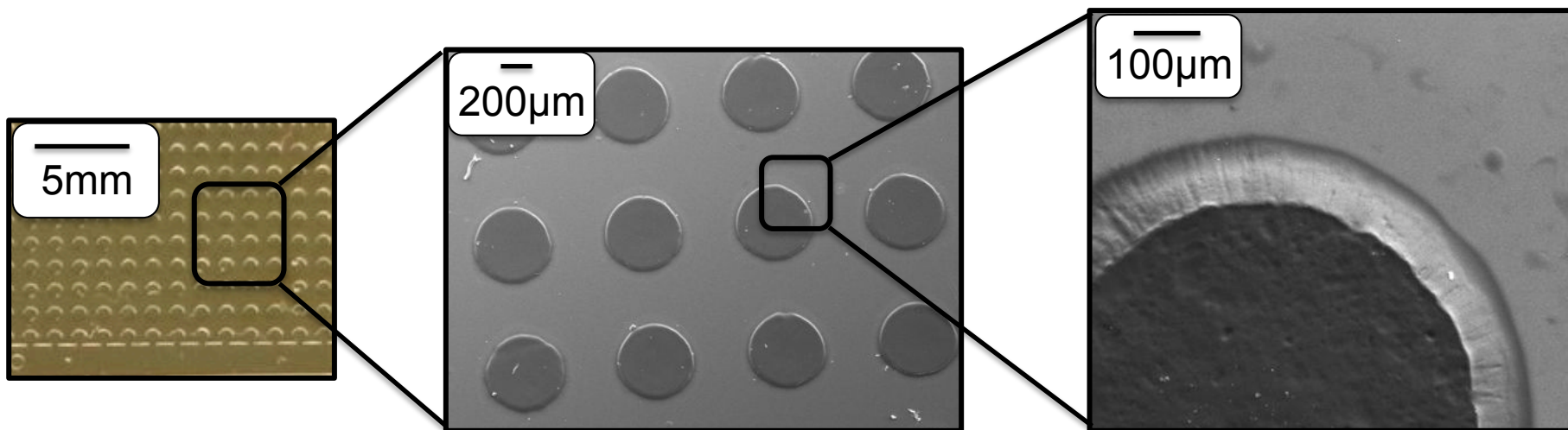
Insight



# Hydrogels microstructures



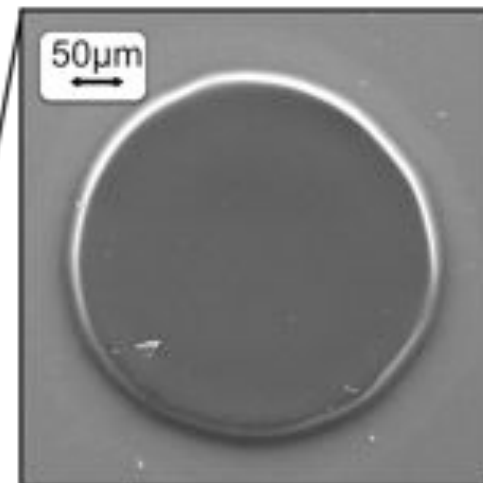
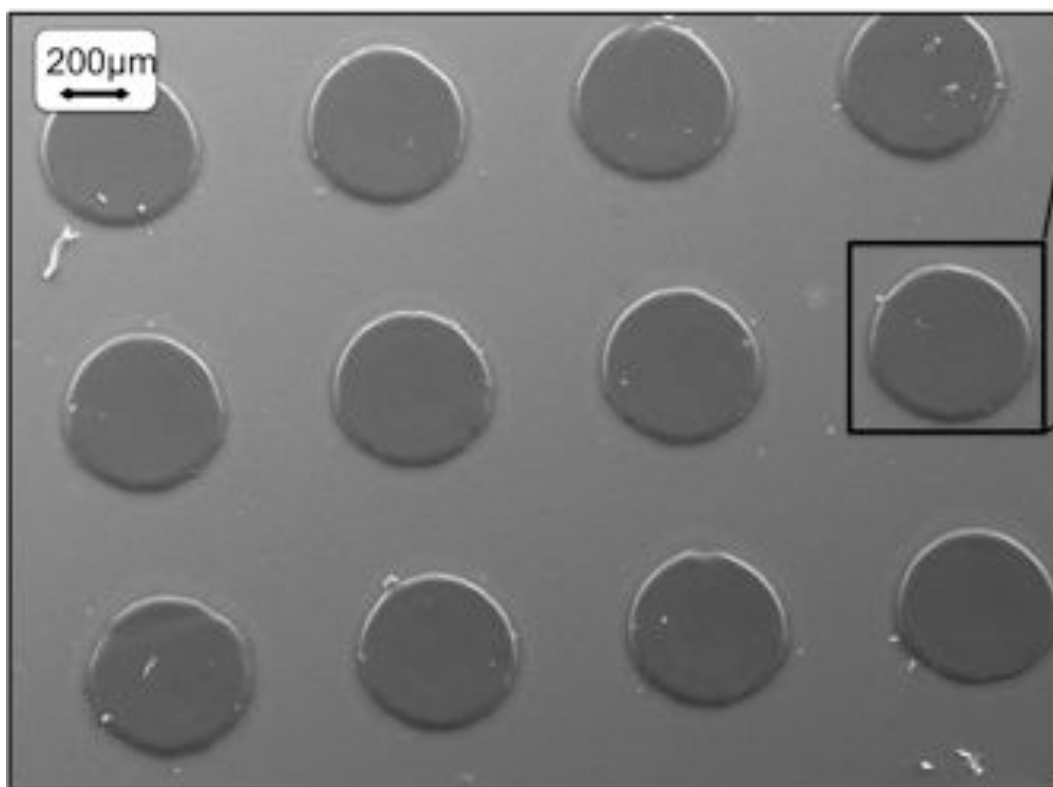
1 mol NIPAM  
5% Acrylic acid  
1% acrylated-Spiropyran  
3% MBIS  
1% PBPO  
Polymerization solvent



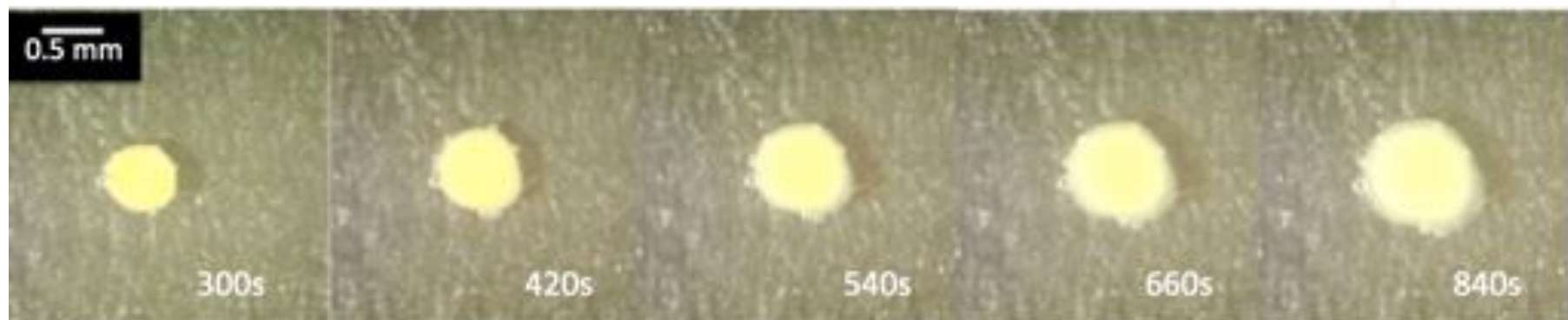
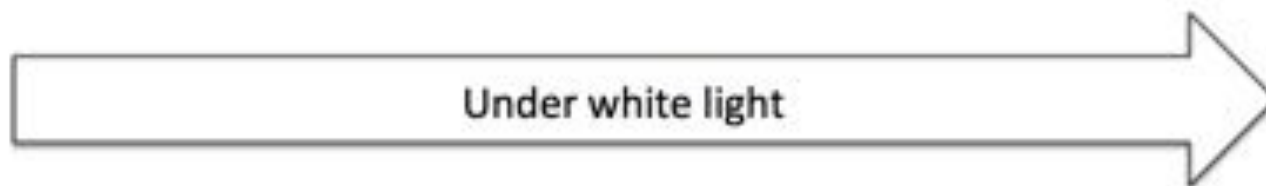
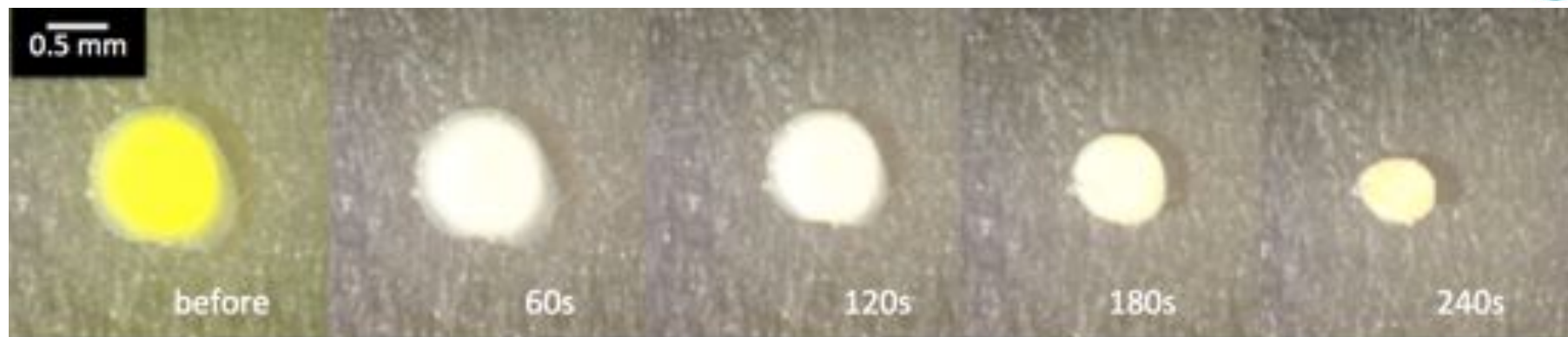


# Micro-patterned hydrogels

Hydrogel microstructures covalently attached to glass substrates were photopolymerised through micro-patterned masks using white light.



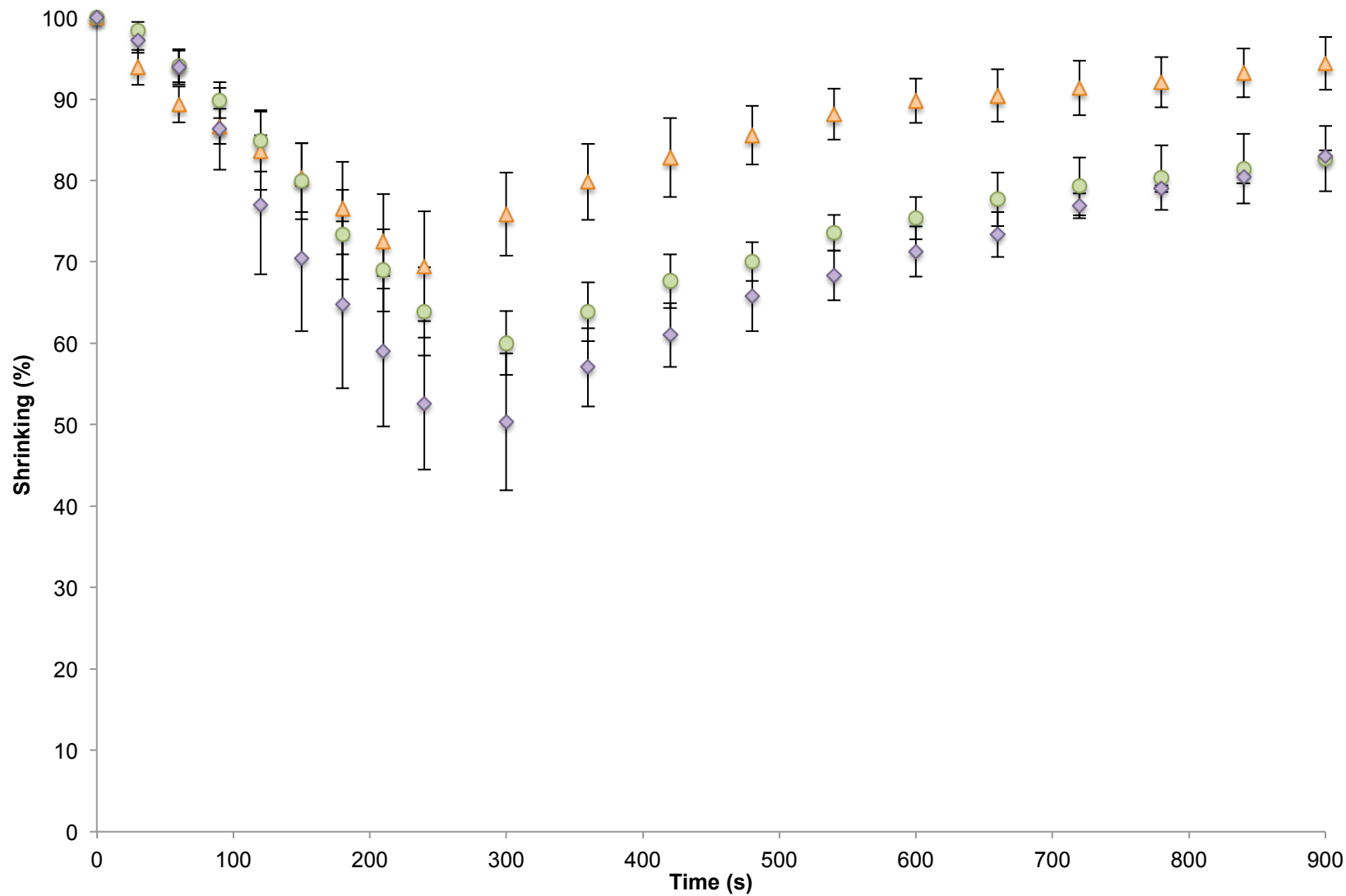
- Circular masks of 1mm diameter.
- Hydrogel height: 60μm.
- Polymerisation time: 10-20 seconds.





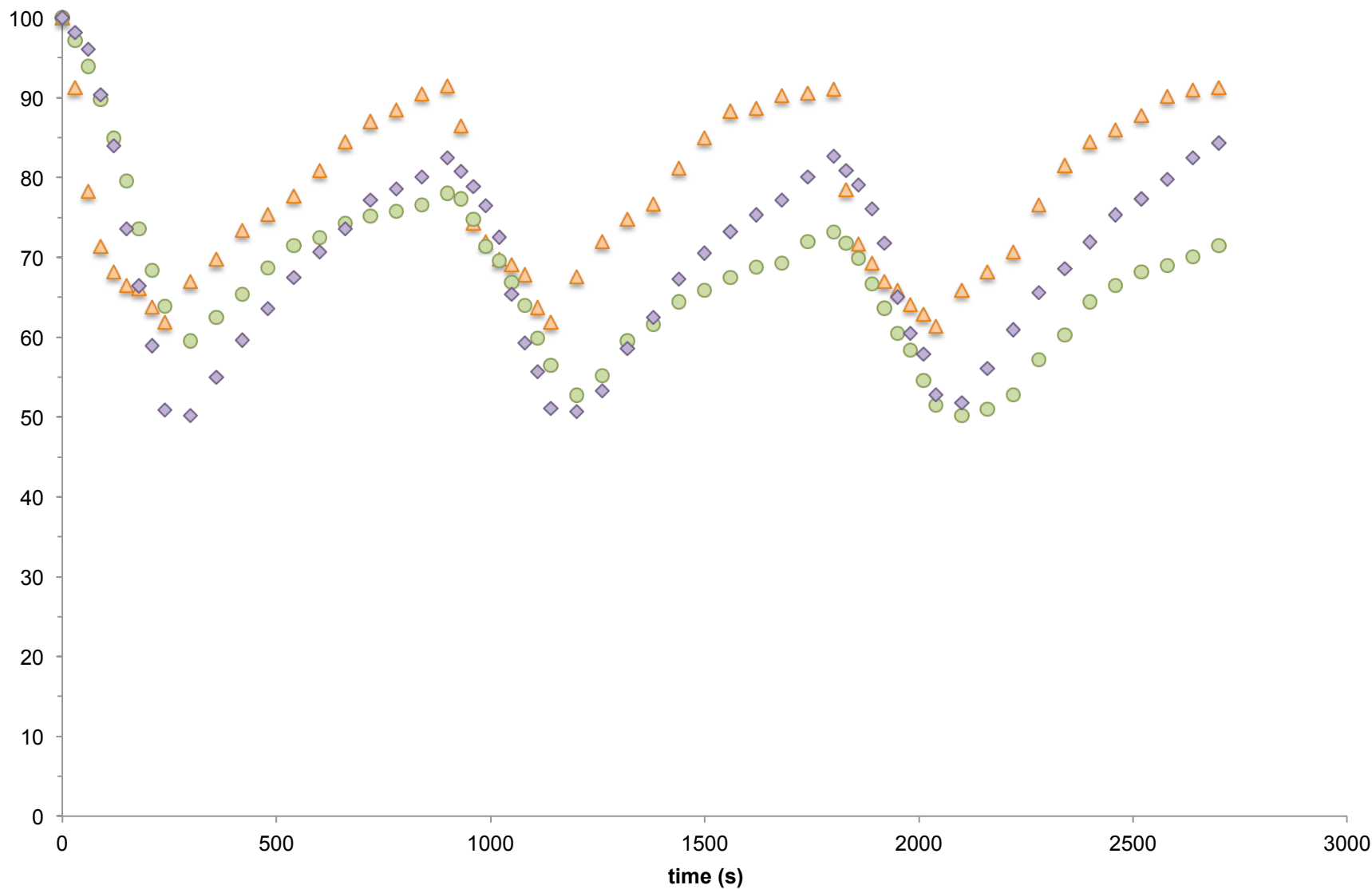


# 1<sup>st</sup> Irradiation Cycle





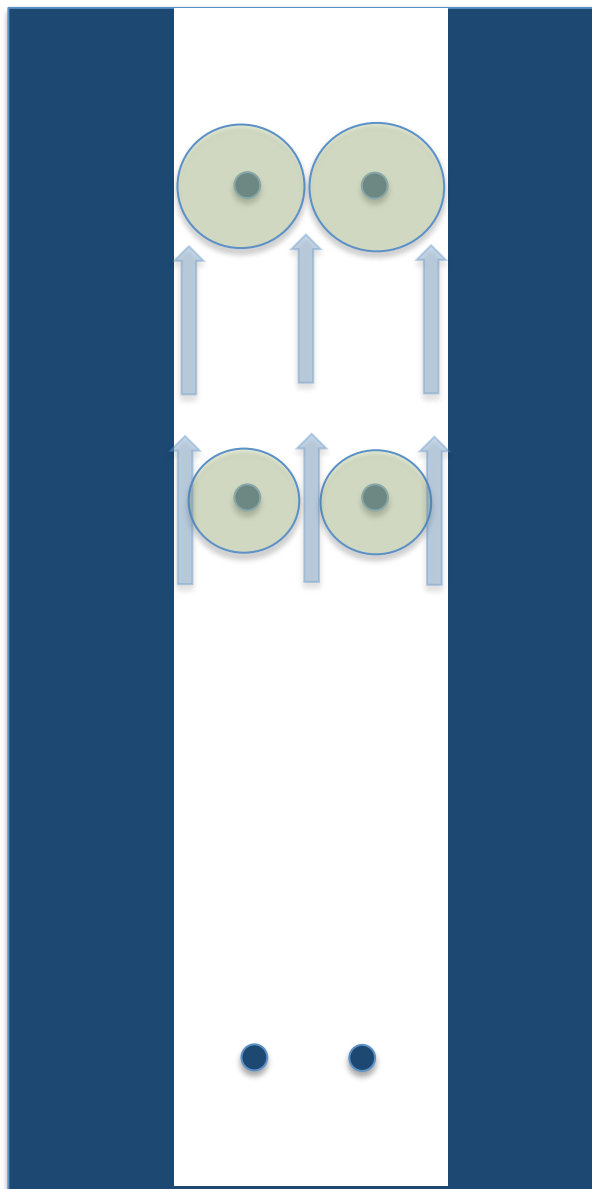
# Multiple Irradiation Cycles



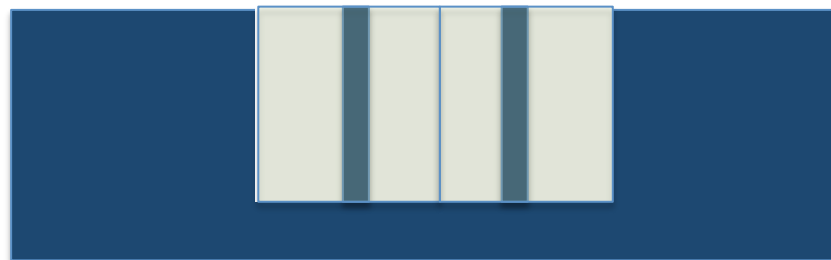


# Valve applications



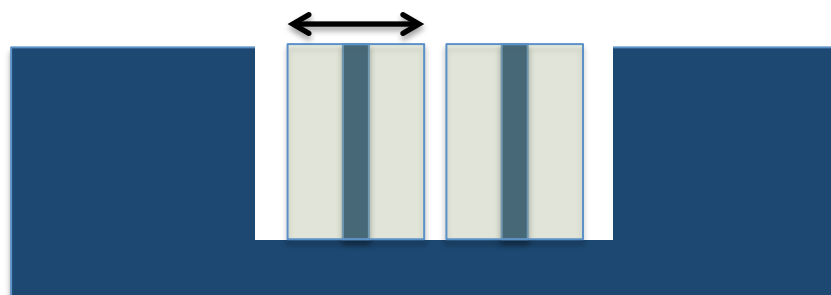


3

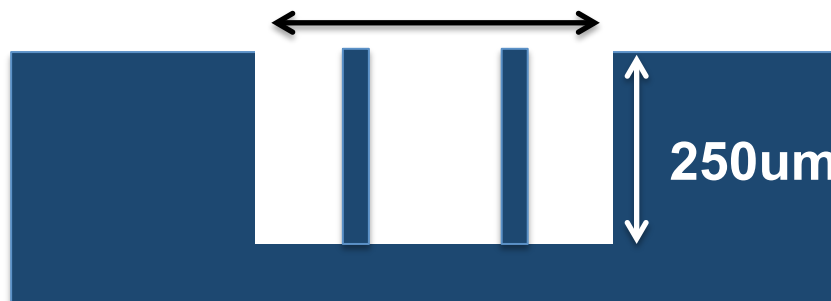


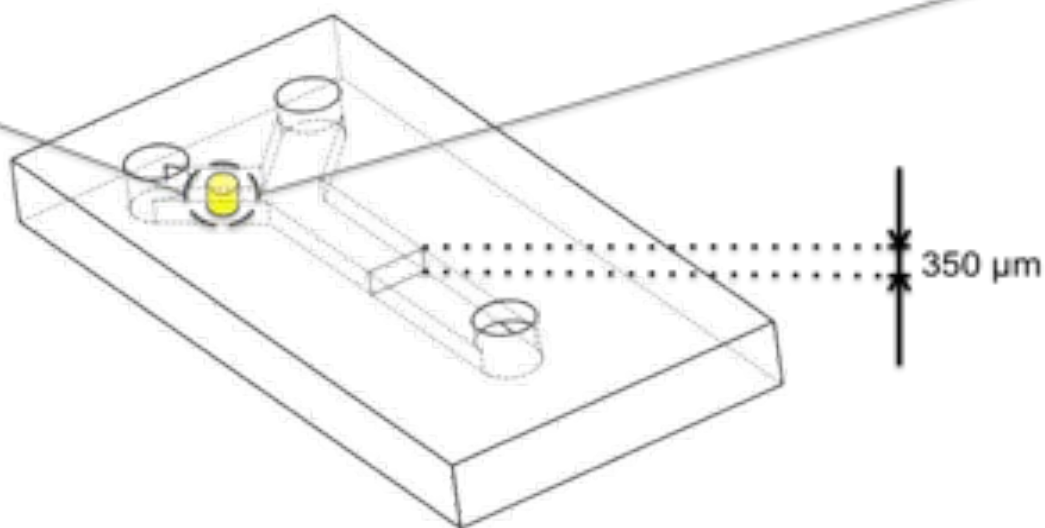
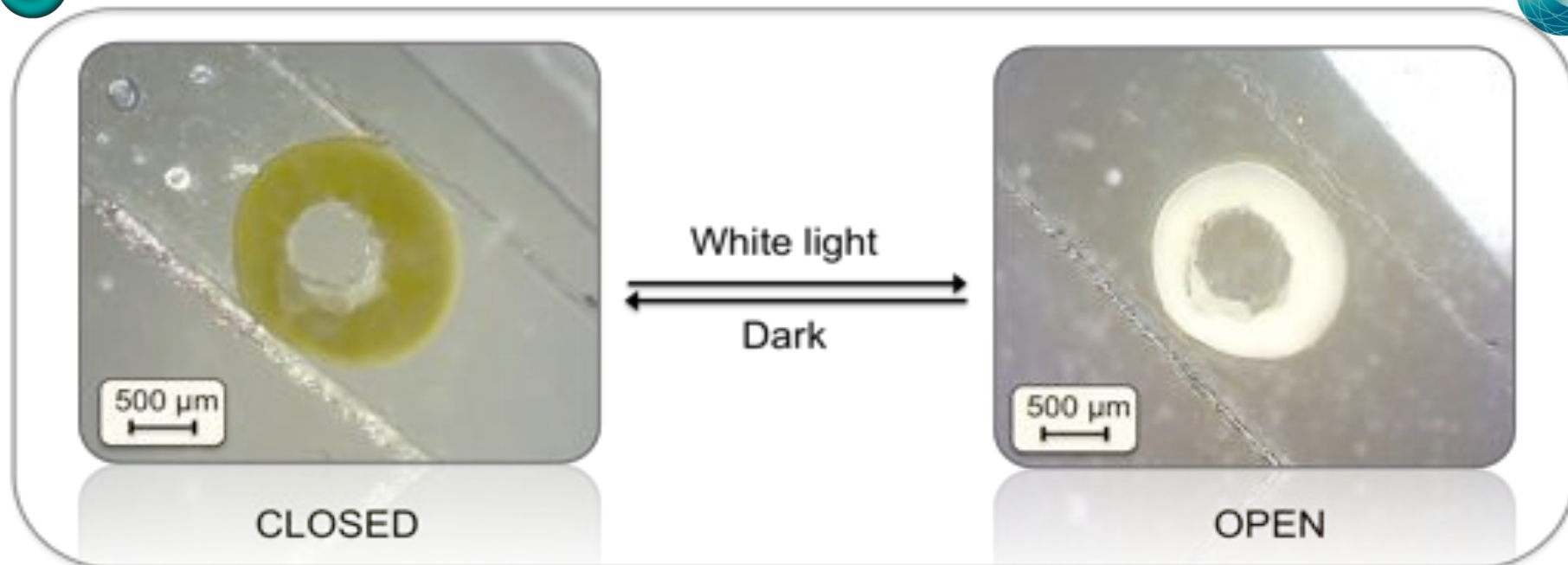
250um gel disks polymerised  
around pillar using mask

2



1



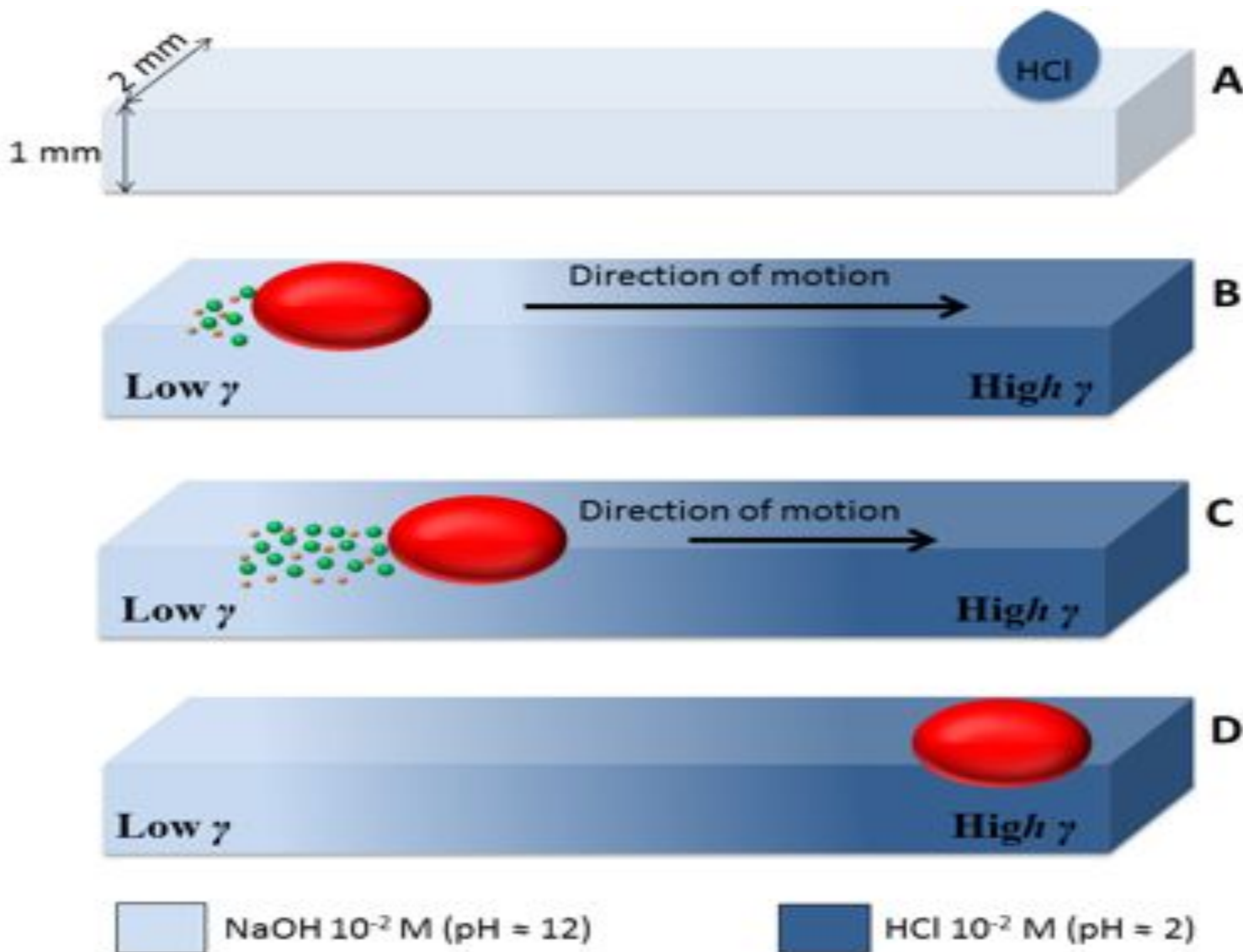




# Surfactant-driven vehicles

## Chemotactic Droplets

# Controlled-release of surfactants





# Video 1



**Speed X 4**





## Video 2:



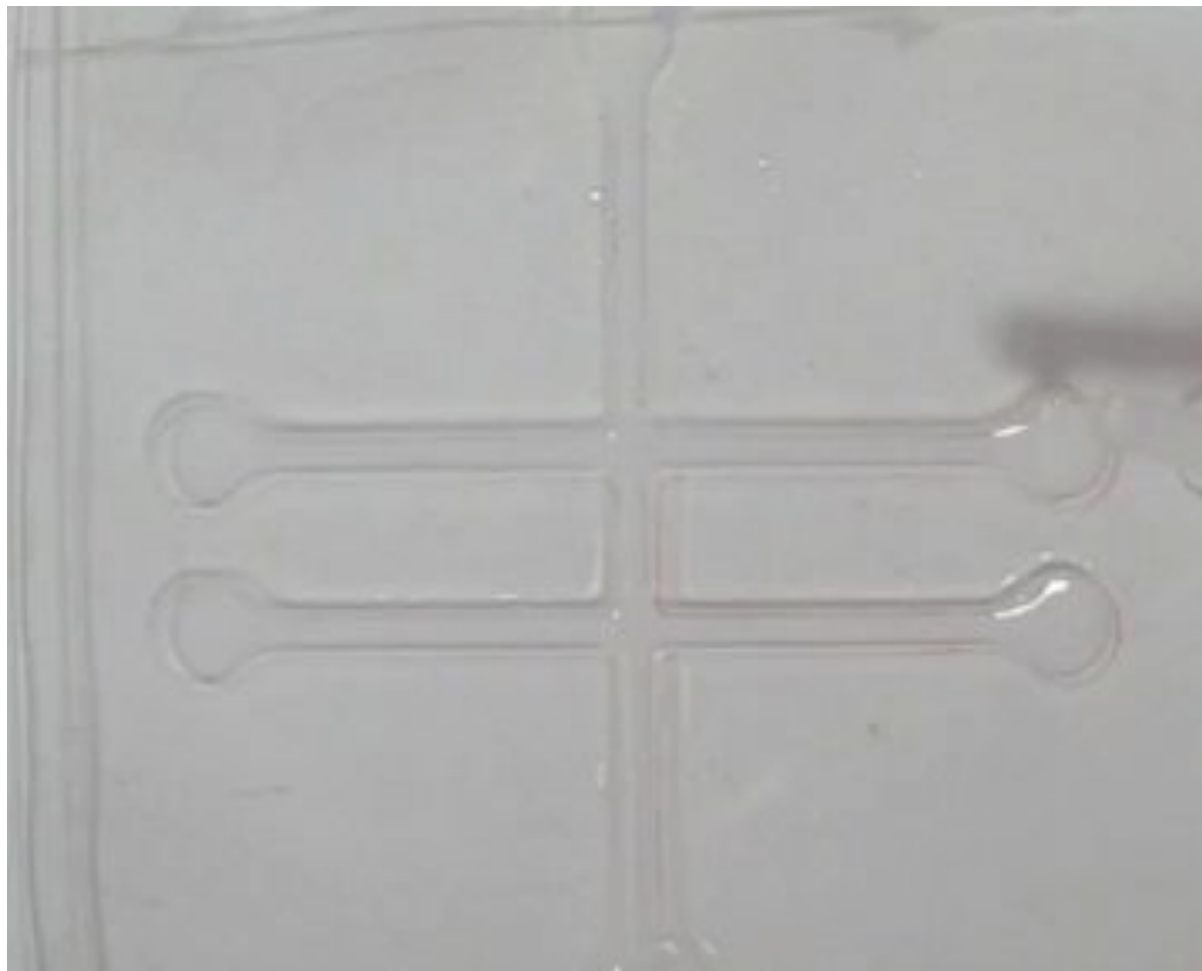
**Speed X 2**







# Multiple Chemotactic Droplets

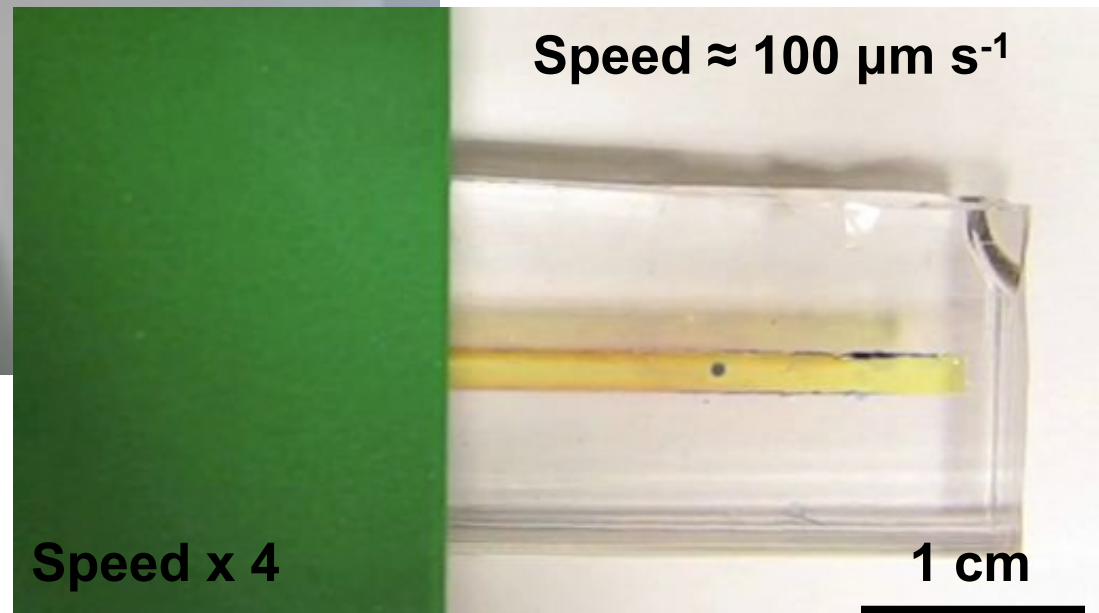


**Speed X 2**

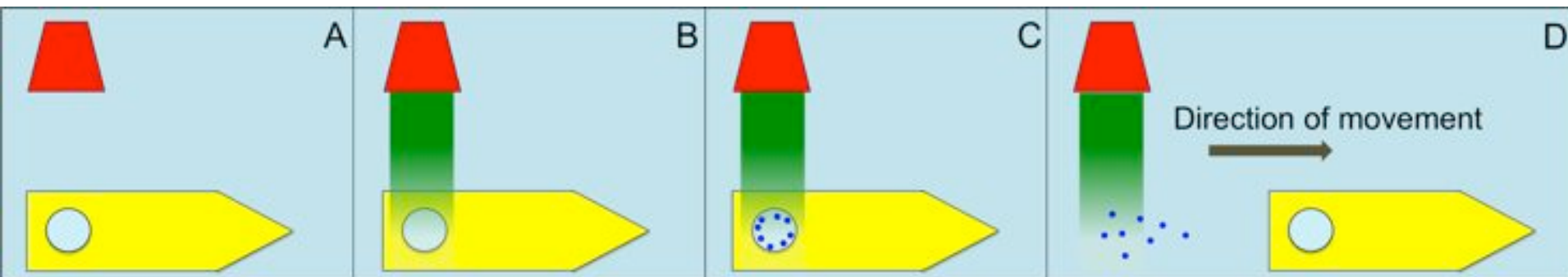
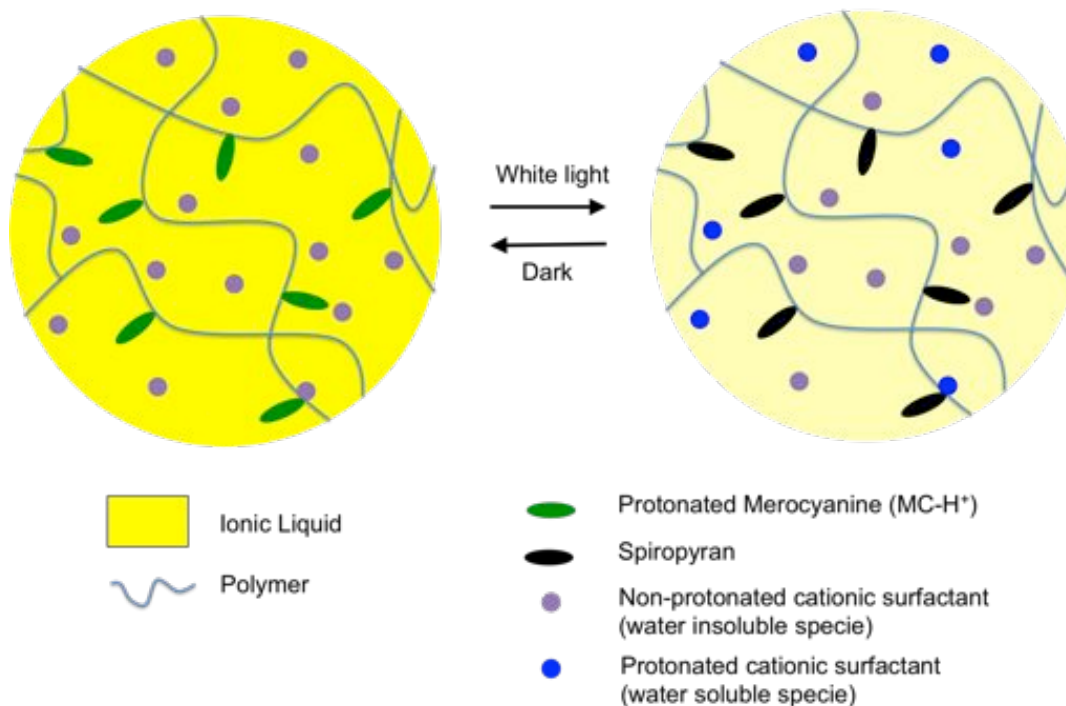


# Photo-activated chemopropulsion



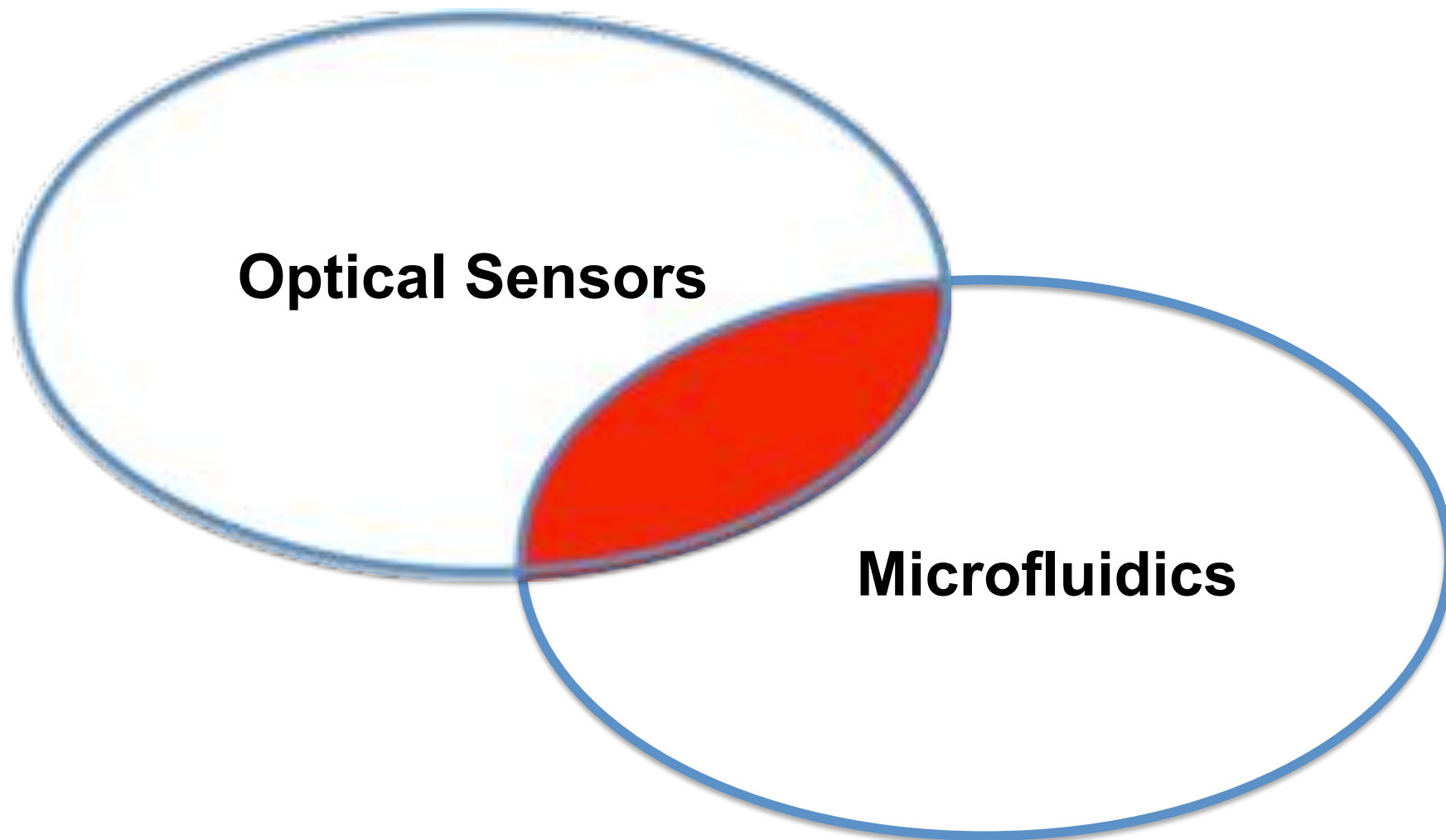


# Perspectives



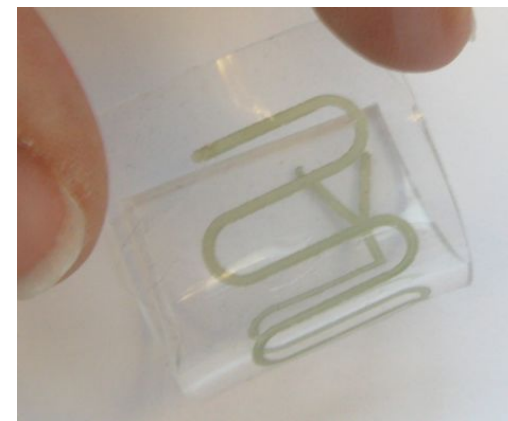
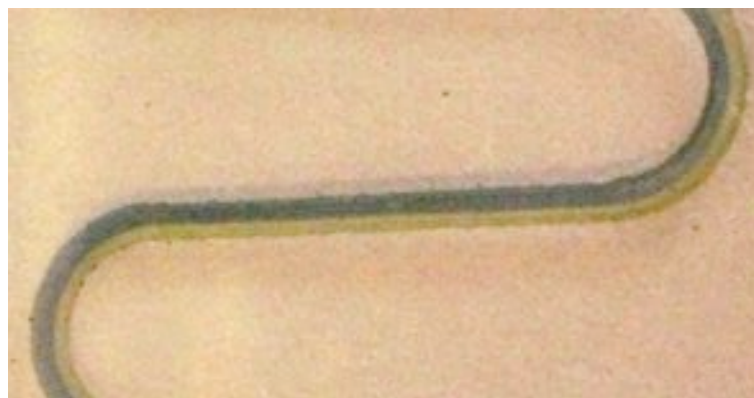


# Optical Sensors in Microfluidics



# Polyaniline functionalised micro-capillaries and micro-fluidic channels

- pH sensing
- ammonia sensing
- diffusion study

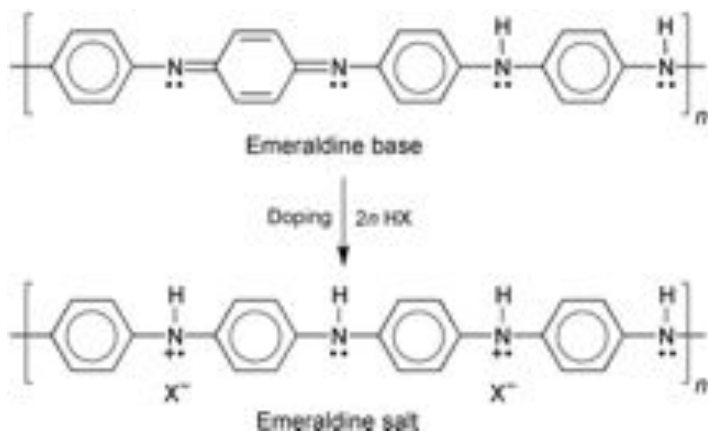






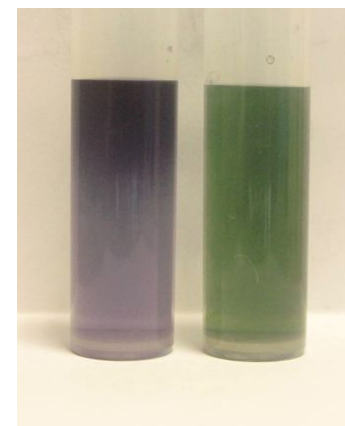
# Polyaniline Nanofibres

- low cost, easy synthesis
- reversible acid-base doping-dedoping chemistry
- environmental stability



Insulating State  
Blue/Violet Colour

Conducting State  
Green

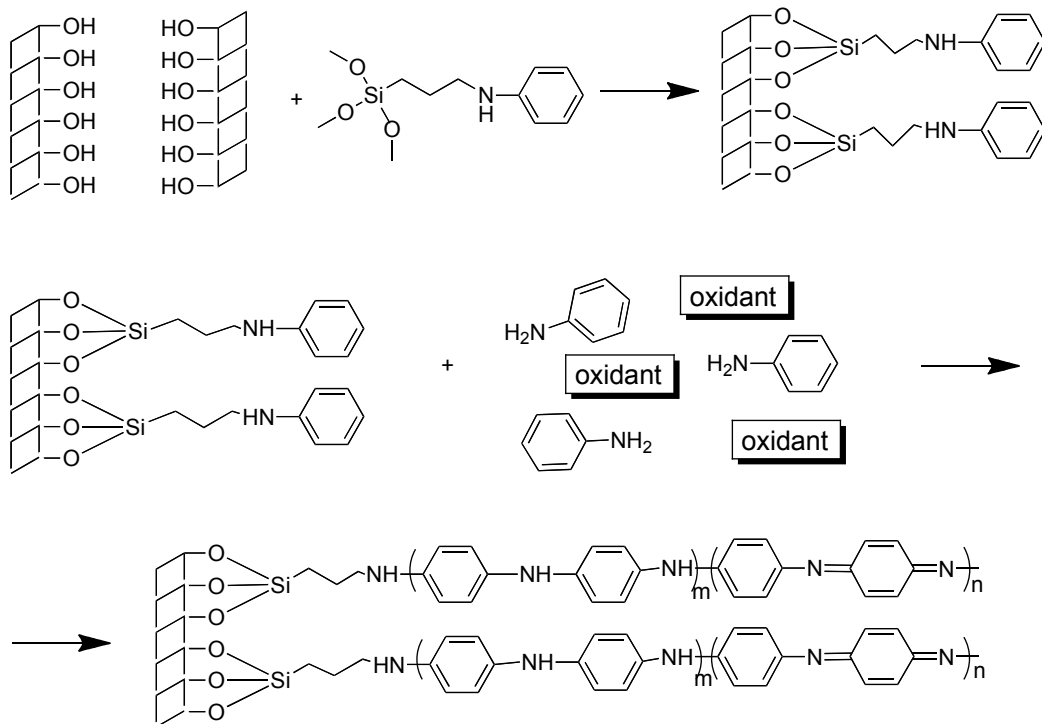


J.X. Huang, S. Viril, B.H. Weller, R.B. Kaner / J.Am.Chem.Soc. 125 (2003), 314-315



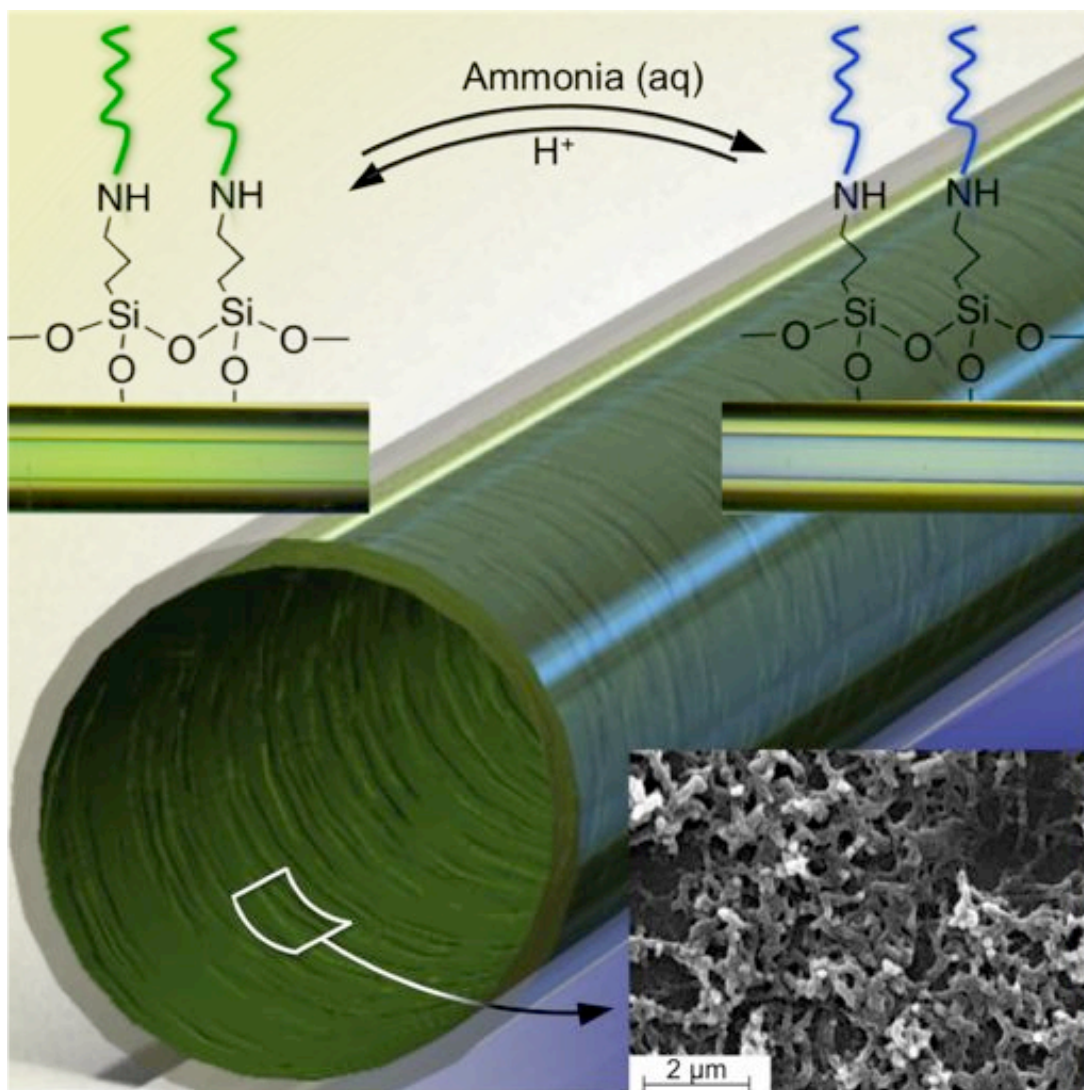
# Polyaniline-coated micro-capillaries for ammonia sensing

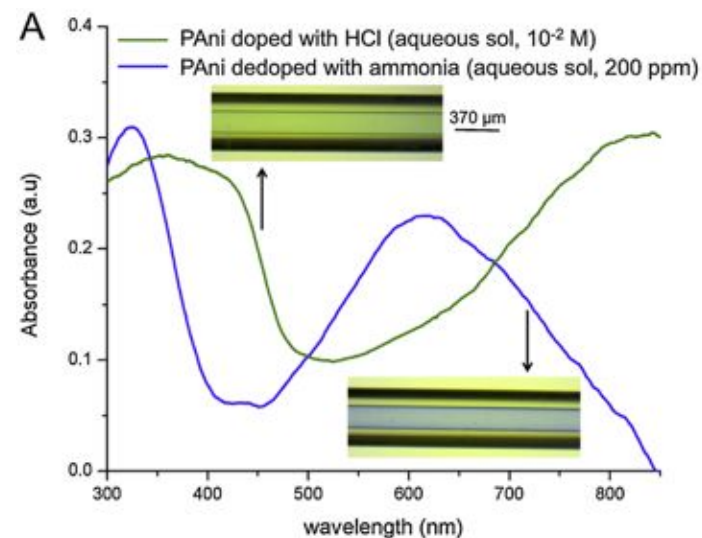
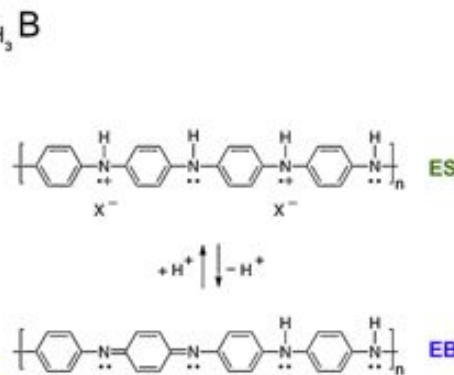
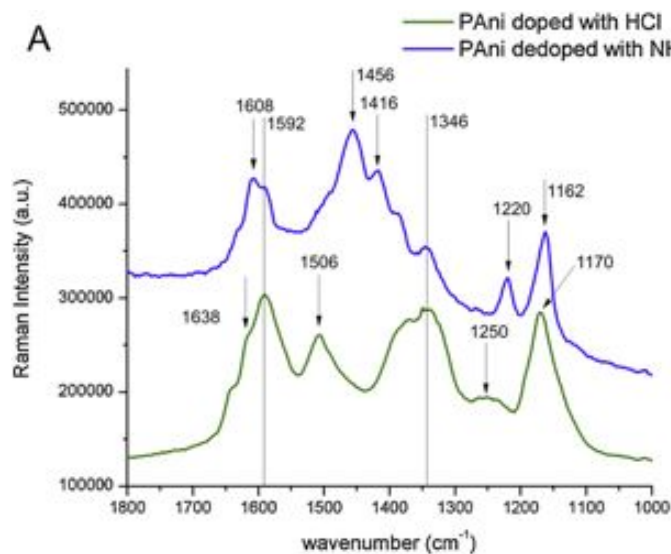
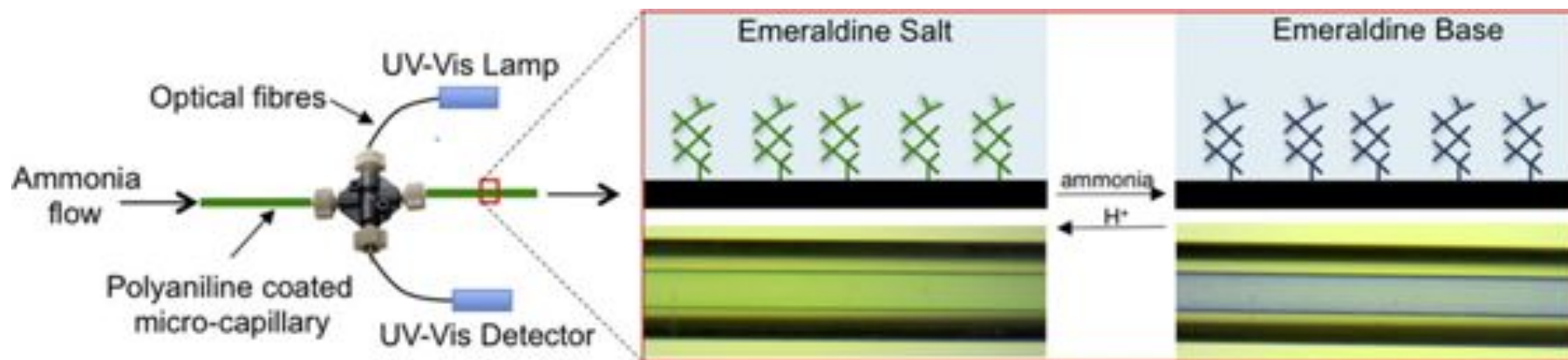
## ➤ Micro-capillary functionalisation



L. Florea, D. Diamond and F. Benito-Lopez, *Anal. Chim. Acta*, 2013, 759, 1-7

# Polyaniline-coated micro-capillaries for ammonia sensing

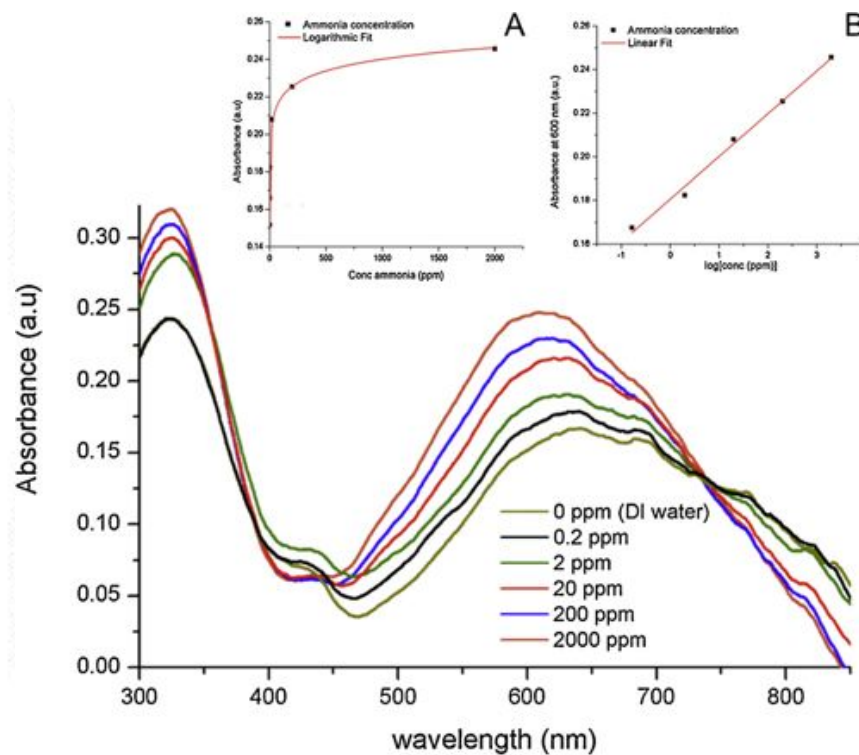
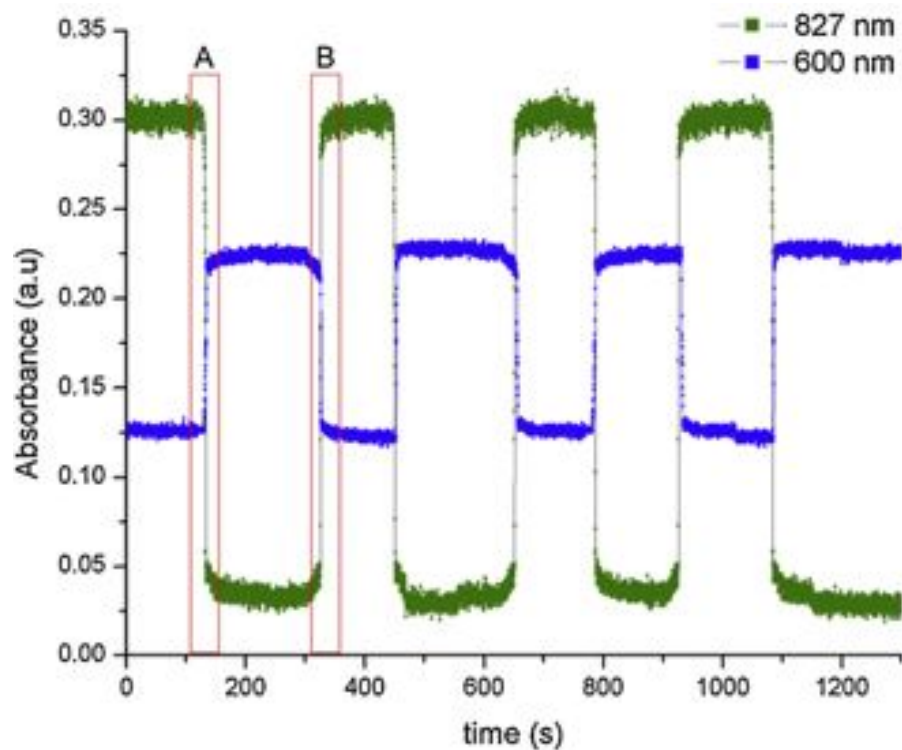








# Ammonia sensing



L. Florea, D. Diamond and F. Benito-Lopez, *Anal. Chim. Acta*, 2013, 759, 1-7



## Micro-chip fabrication

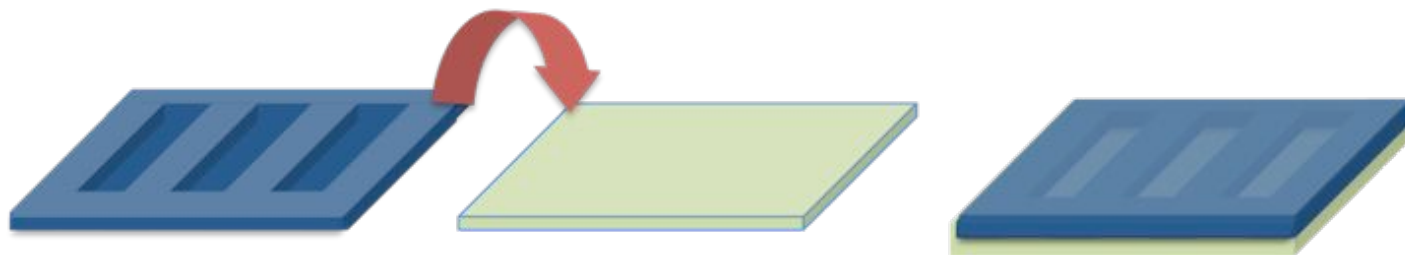
✓ PDMS is poured onto master mold, cured at 80°C for 1 h and removed from mold.



✓ PDMS and glass slide are treated with oxygen plasma.



✓ PDMS and glass slide are brought together.



L. Yu, C.M. Li, Y. Liu *et al.* / Lab Chip, 9 (2009), 1243–1247.



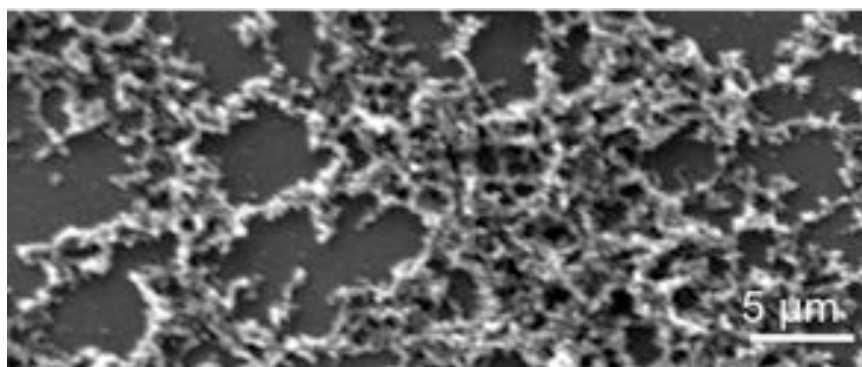
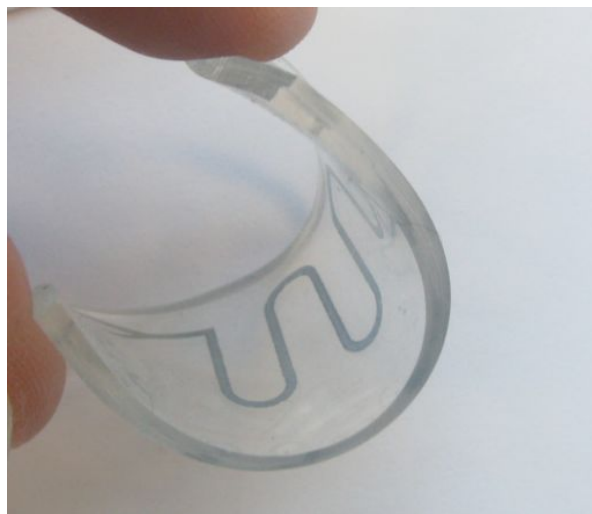


# Micro-channels

**500 $\mu$ m x 1000 $\mu$ m**



**1000 $\mu$ m x 100 $\mu$ m**

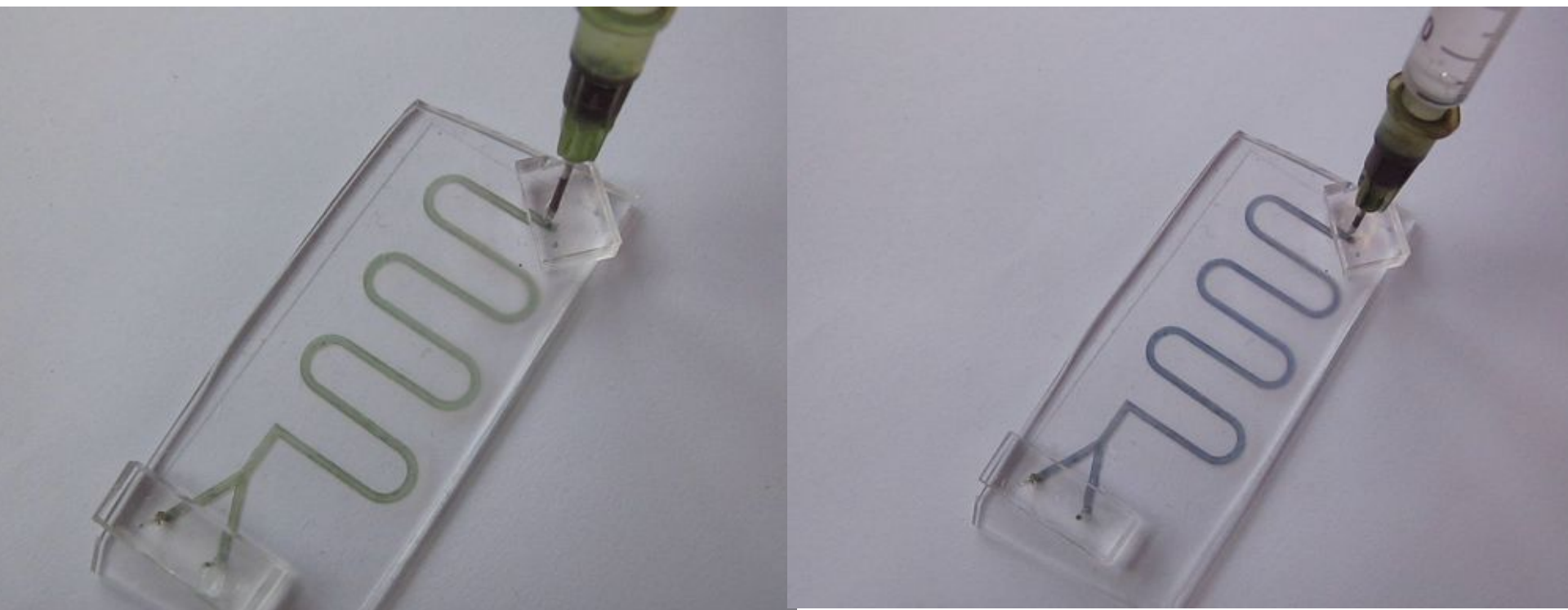


**45 $\mu$ m x 50 $\mu$ m**





# Fast Response

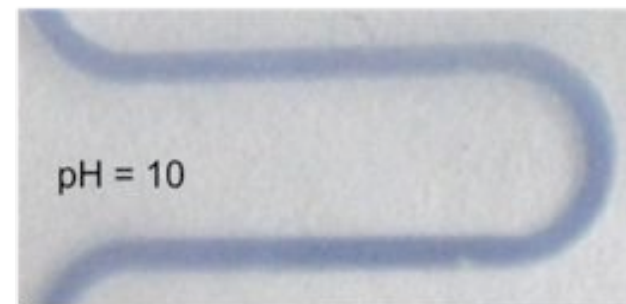
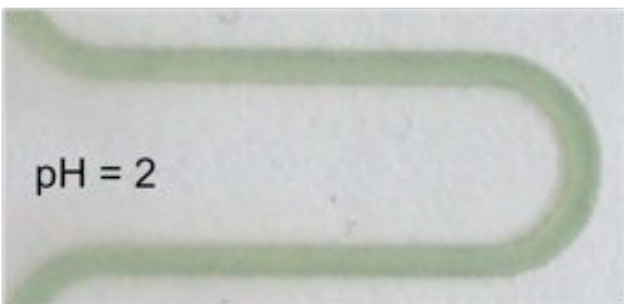


L. Florea, C. Fay, E. Lahiff, T. Phelan, N. E. O'Connor, B. Corcoran, D. Diamond and F. Benito-Lopez, *Lab Chip*, 2013, 13, 1079-1085.

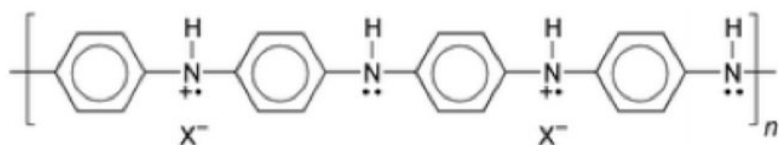




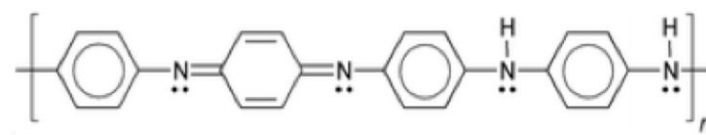
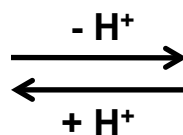
# pH sensing in continuous flow



Dedoping process



**Emeraldine Salt (ES)**



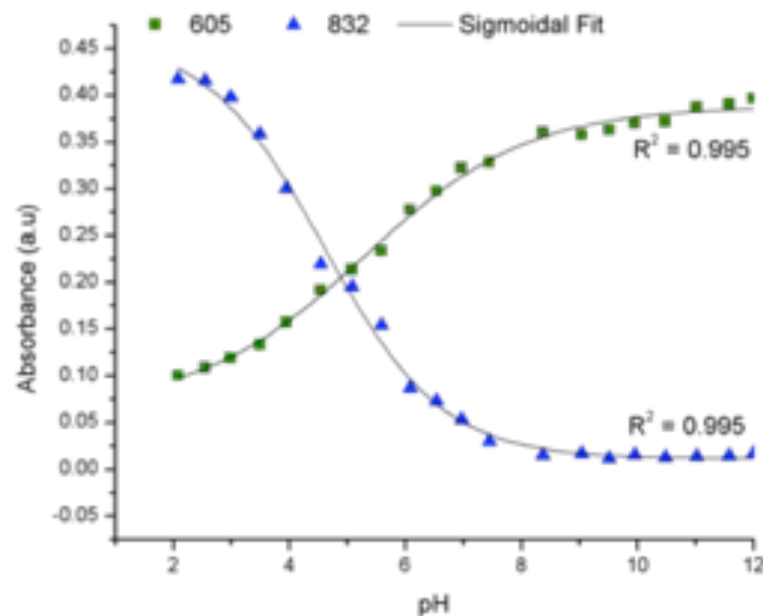
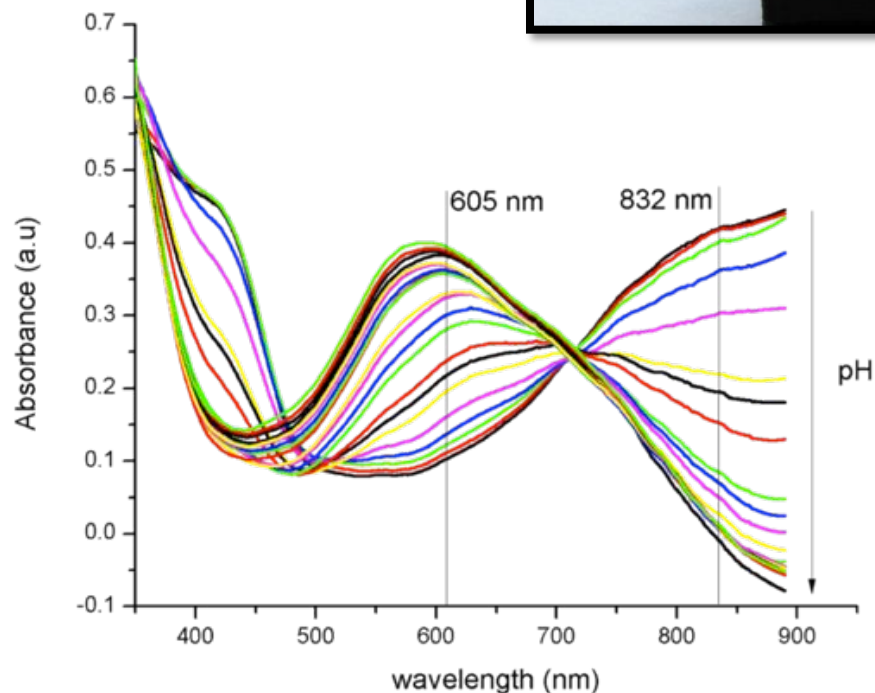
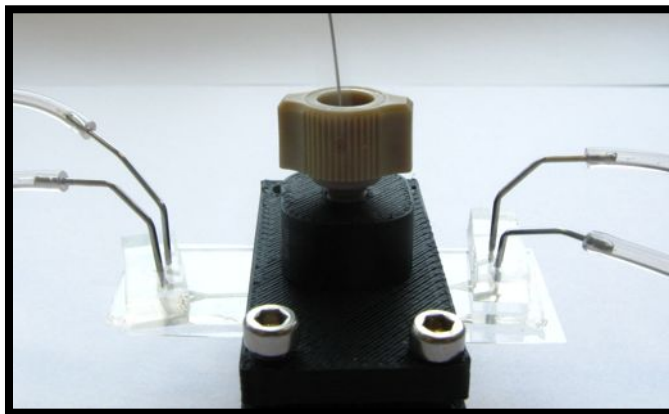
**Emeraldine Base (EB)**

L. Florea, C. Fay, E. Lahiff, T. Phelan, N. E. O'Connor, B. Corcoran, D. Diamond and F. Benito-Lopez, *Lab Chip*, 2013, 13, 1079-1085.





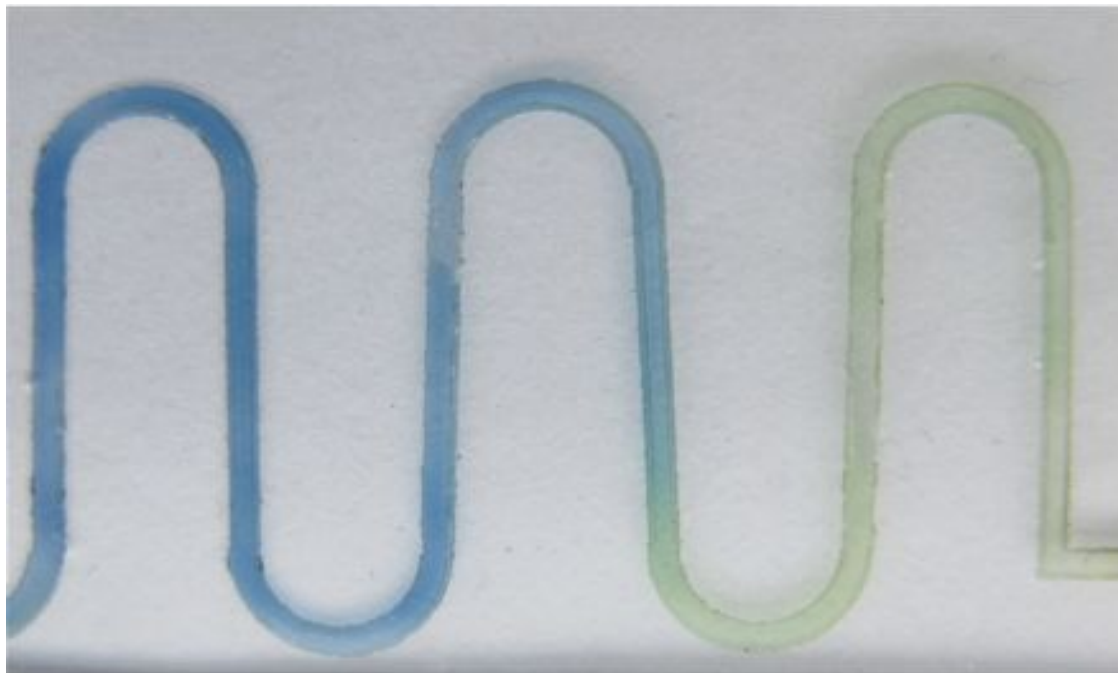
# pH sensing in continuous flow



L. Florea, C. Fay, E. Lahiff, T. Phelan, N. E. O'Connor, B. Corcoran, D. Diamond and F. Benito-Lopez, *Lab Chip*, 2013, 13, 1079-1085.



# Dynamic pH sensing

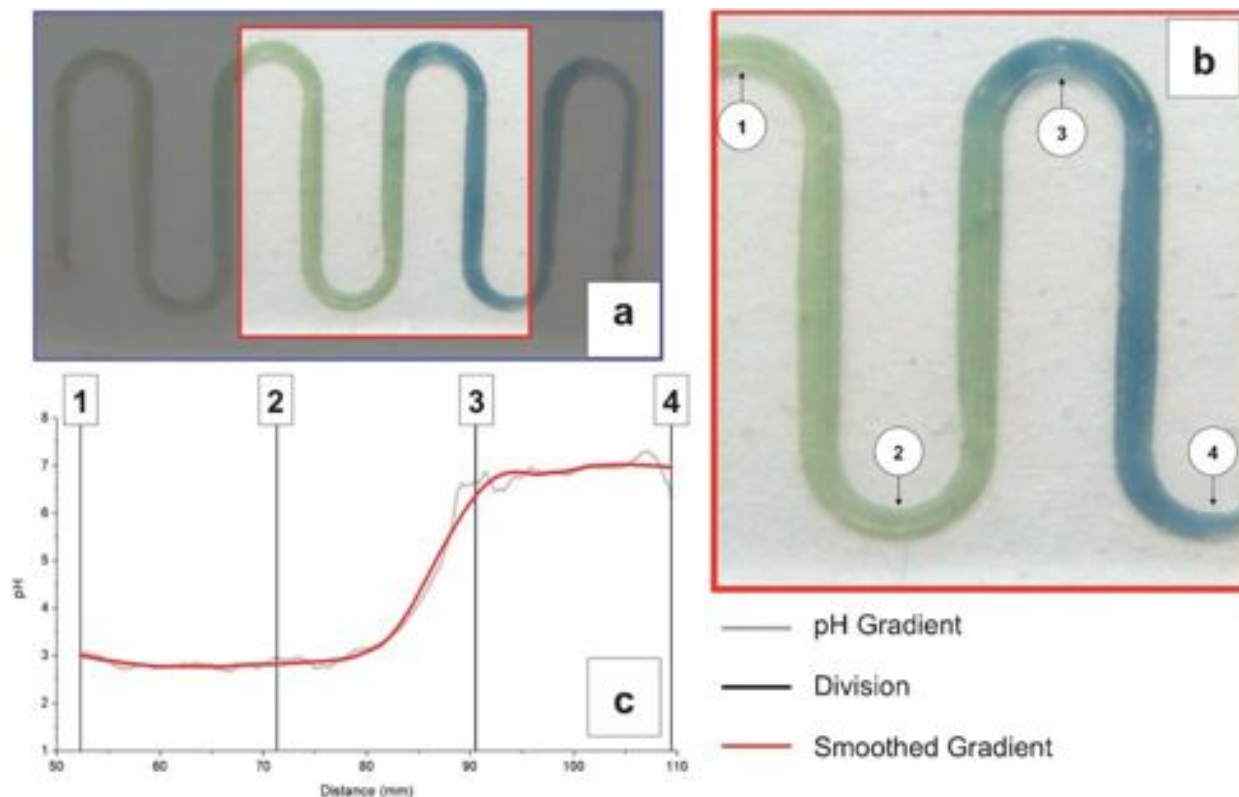


**NaOH  $10^{-2}$  M**

**HCl  $10^{-2}$  M**



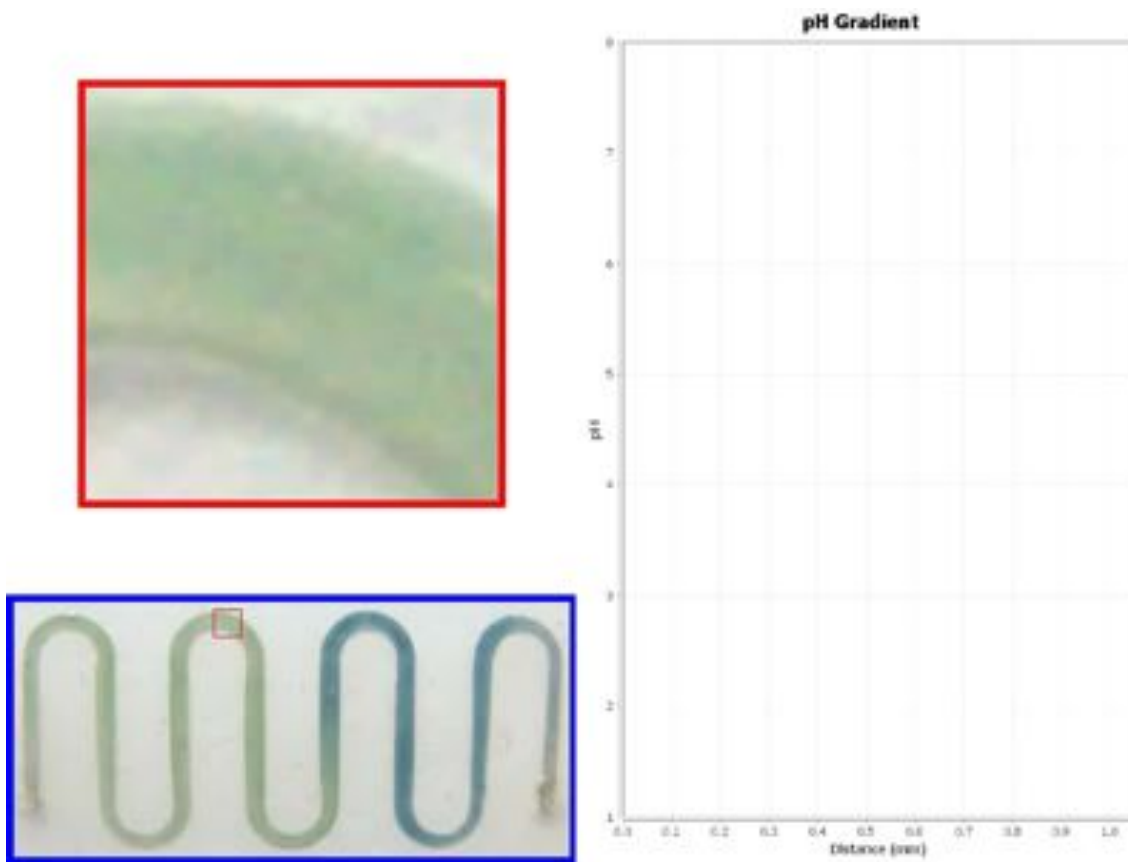




L. Florea, C. Fay, E. Lahiff, T. Phelan, N. E. O'Connor, B. Corcoran, D. Diamond and F. Benito-Lopez, *Lab Chip*, 2013, 13, 1079-1085.



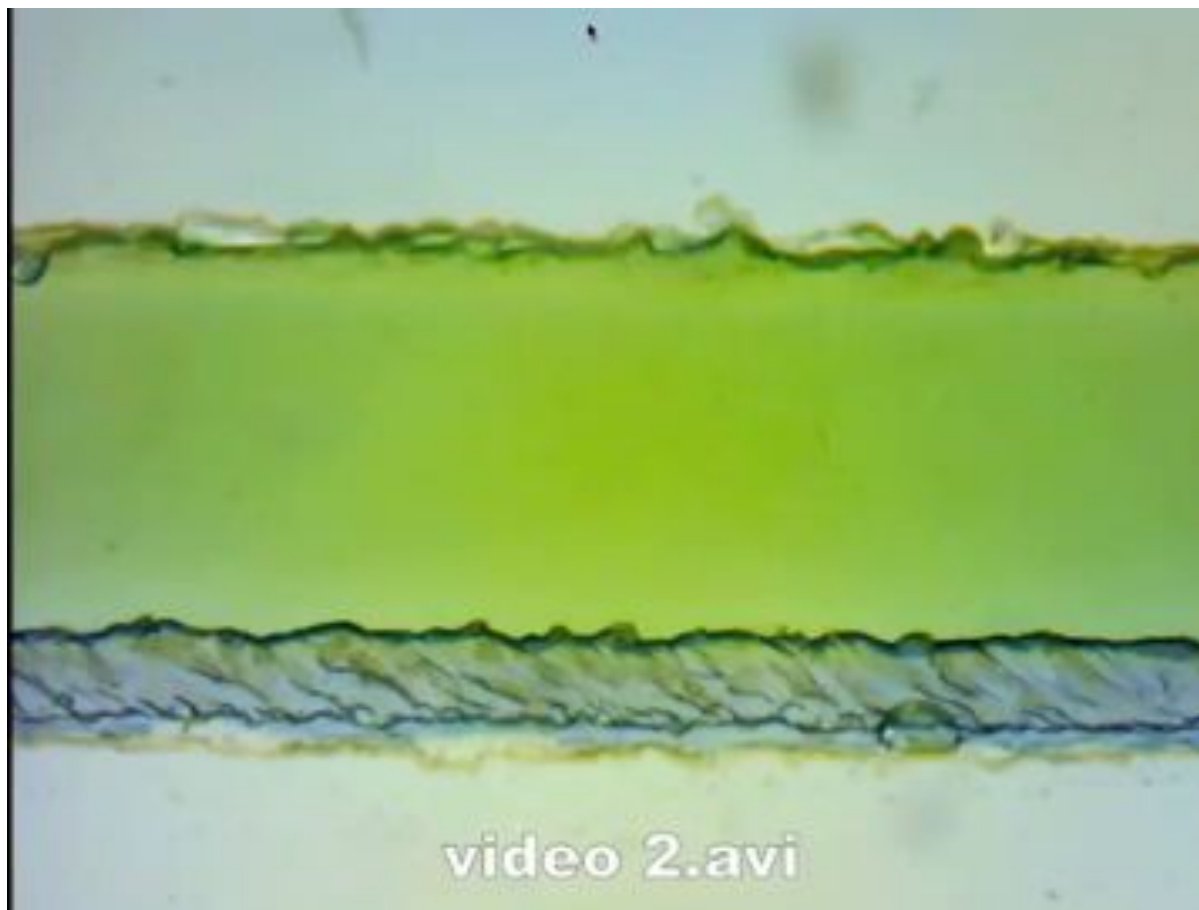
# pH gradient sensing



L. Florea, C. Fay, E. Lahiff, T. Phelan, N. E. O'Connor, B. Corcoran, D. Diamond and F. Benito-Lopez, *Lab Chip*, 2013, 13, 1079-1085.

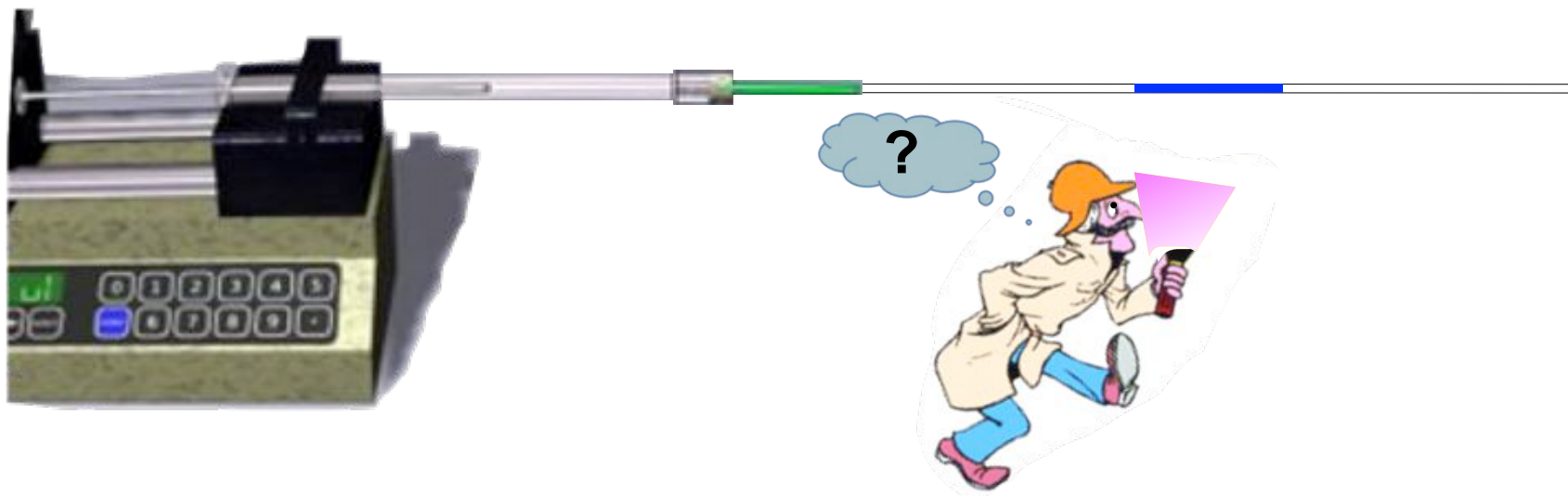


# Study of diffusion process

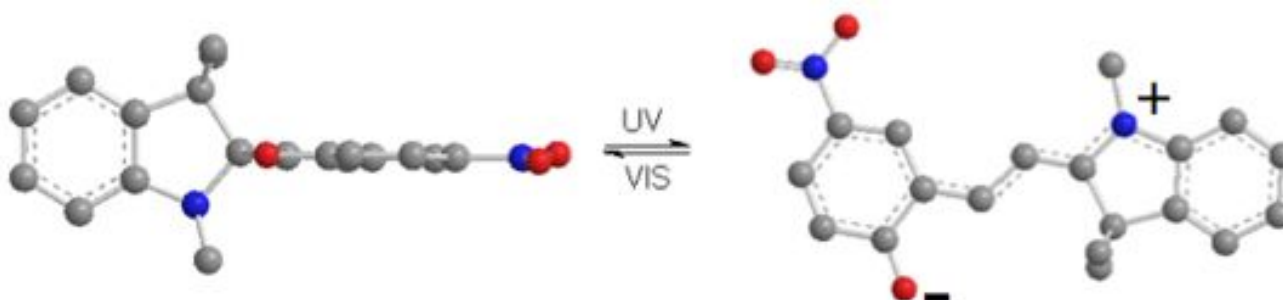


# Spiropyran polymeric brushes functionalised micro-capillaries

- **ON/OFF sensing**
- **solvent sensing**
- **metal ion sensing**

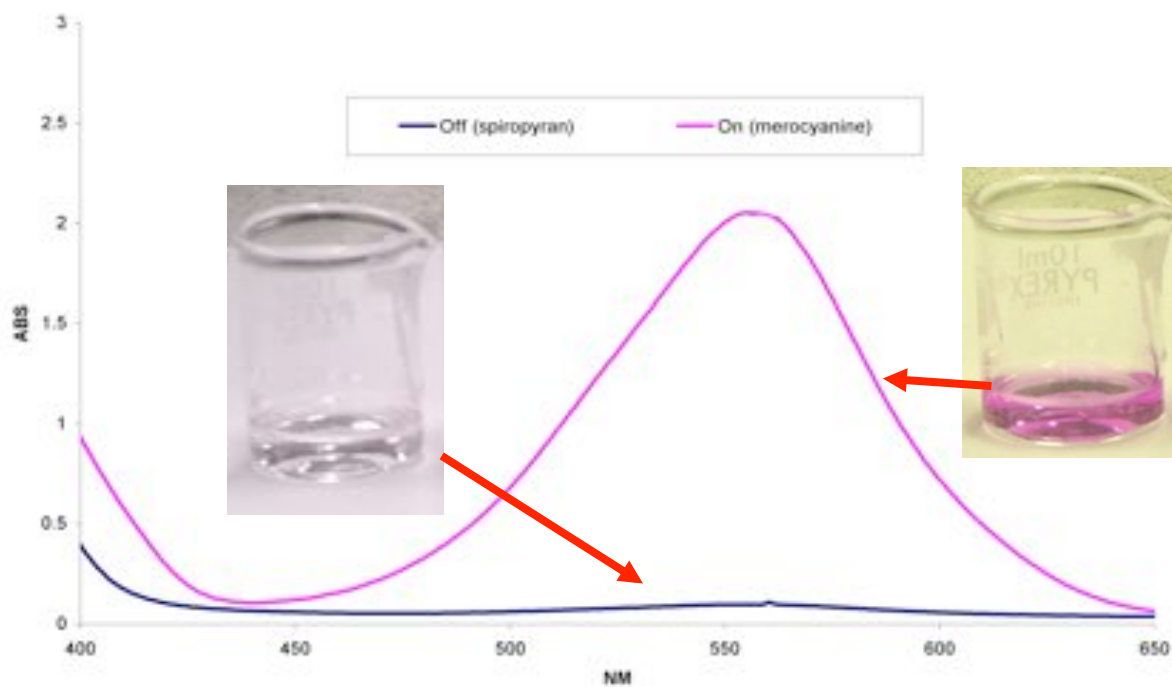


# Spiropyran



**A : Spiropyran SP (closed, colorless)**

**B : Merocyanine MC (open, colored)**



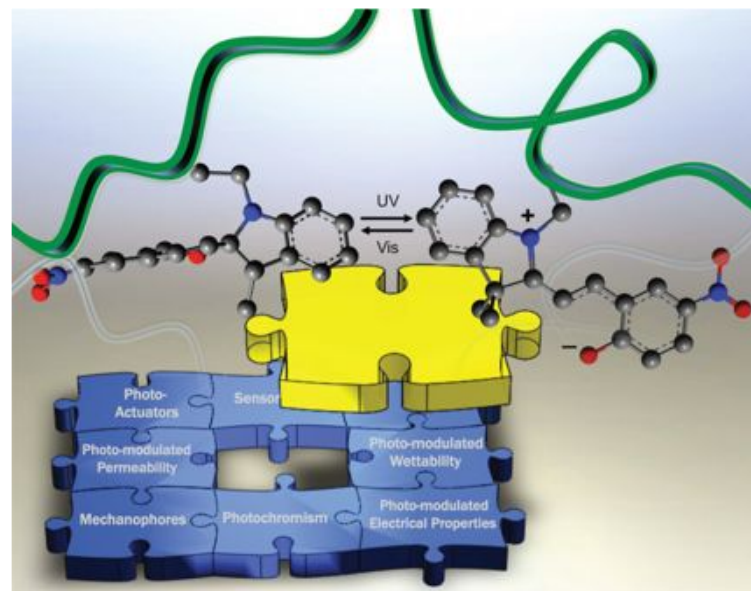
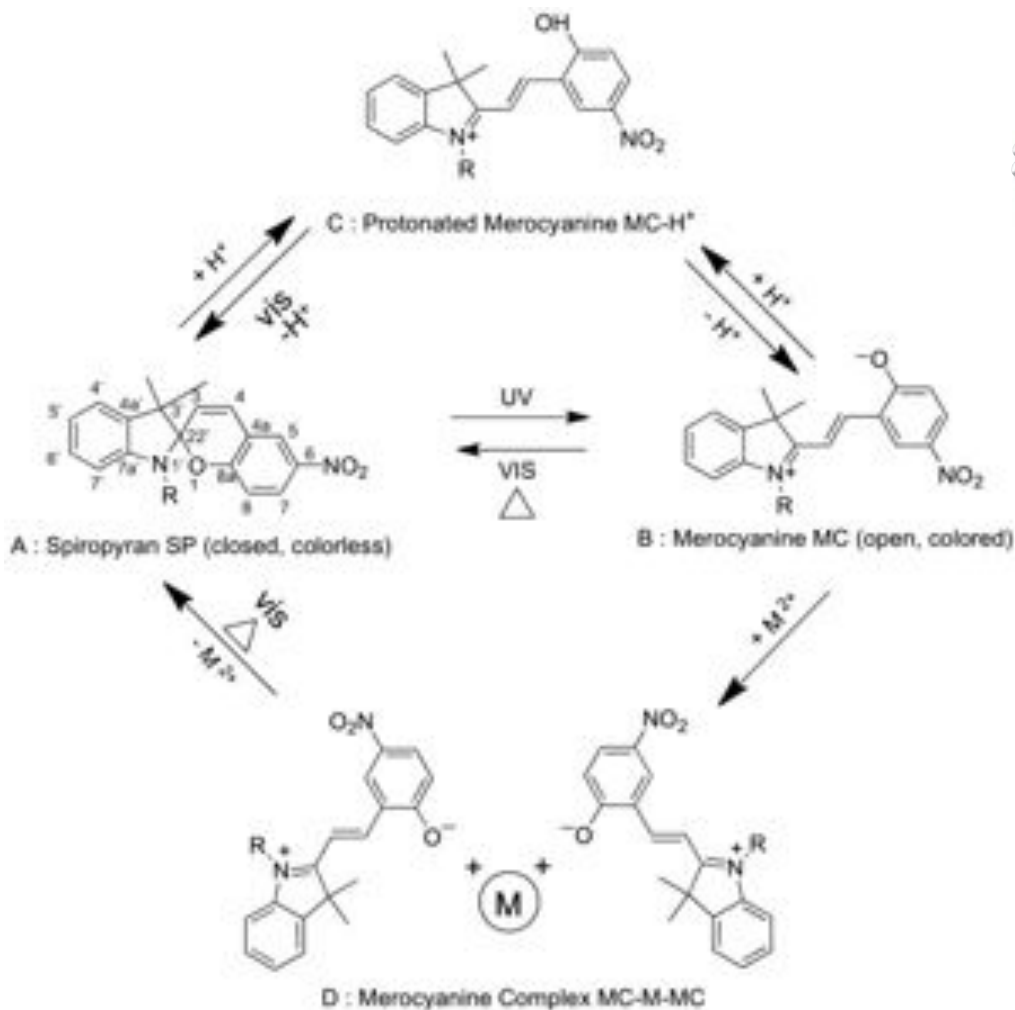
# Spiropyran

ISSN 1438-7492 · MMENFA 297 (12) 1129–1236 (2012) · Vol. 297 · No. 12 · December 2012

D 51047



## Macromolecular Materials and Engineering



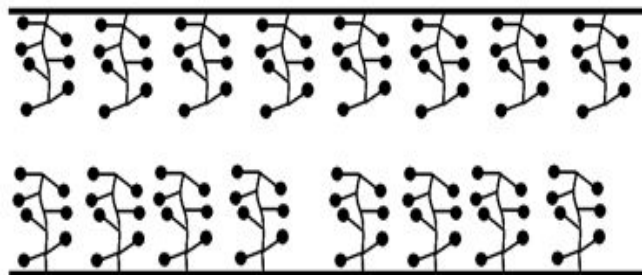
Special Issue:  
Advances in Actively Moving Polymers  
Guest-edited by Andreas Lendlein

12/2012

WILEY-VCH

L. Florea, D. Diamond and F. Benito-Lopez, *Macromolecular Materials and Engineering*, 2012, 297, 1148-1159.

# Our Approach



↑ - spiropyran molecule

- polymer brushes
- high loading of spiropyran molecule
- 3D arrangement

**Micro-capillary : Convenient platform for rapid analysis and detection**

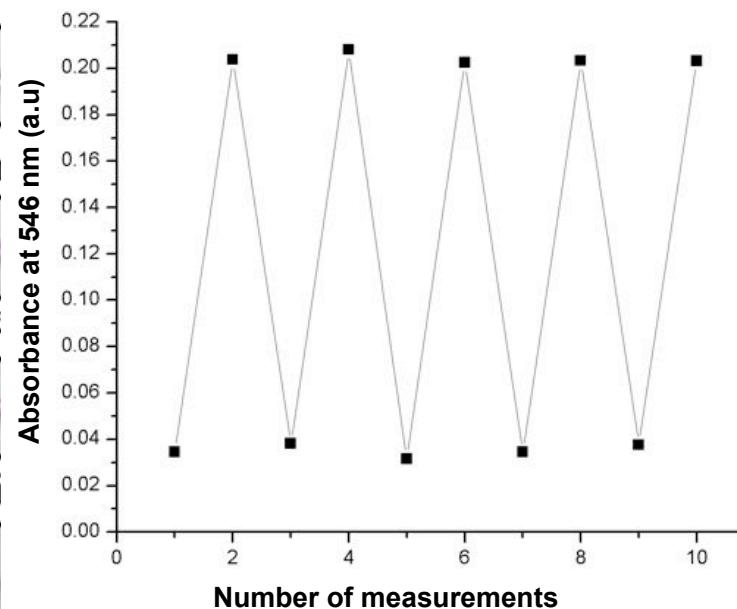
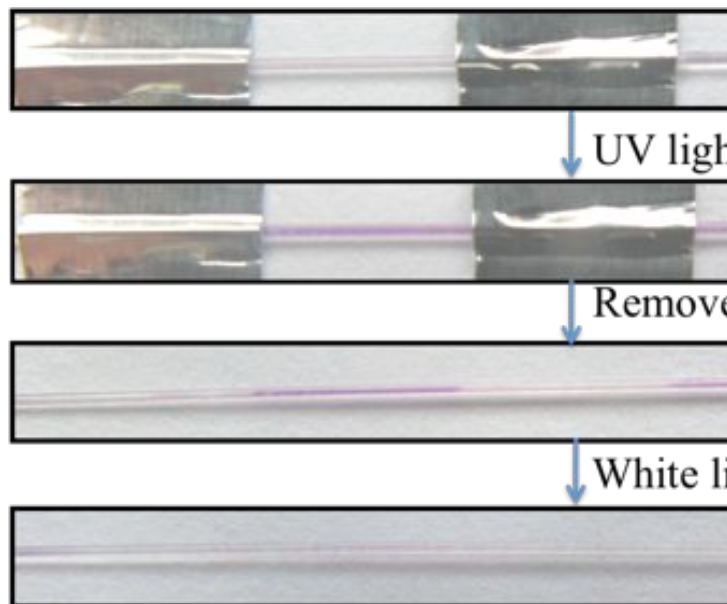
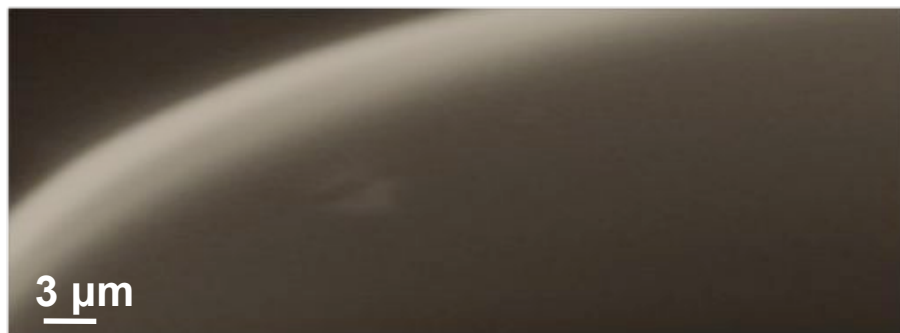
## Advantages

- act as a mechanical support for the optically sensitive layer
- represents an optical waveguide structure
- suitable for real-time continuous flow analysis
- requires very small volume of analyte





# Characterisation



L. Florea, A. Hennart, D. Diamond and F. Benito-Lopez, *Sens. Actuators B: Chem.*, 2012, 175, 92-99.

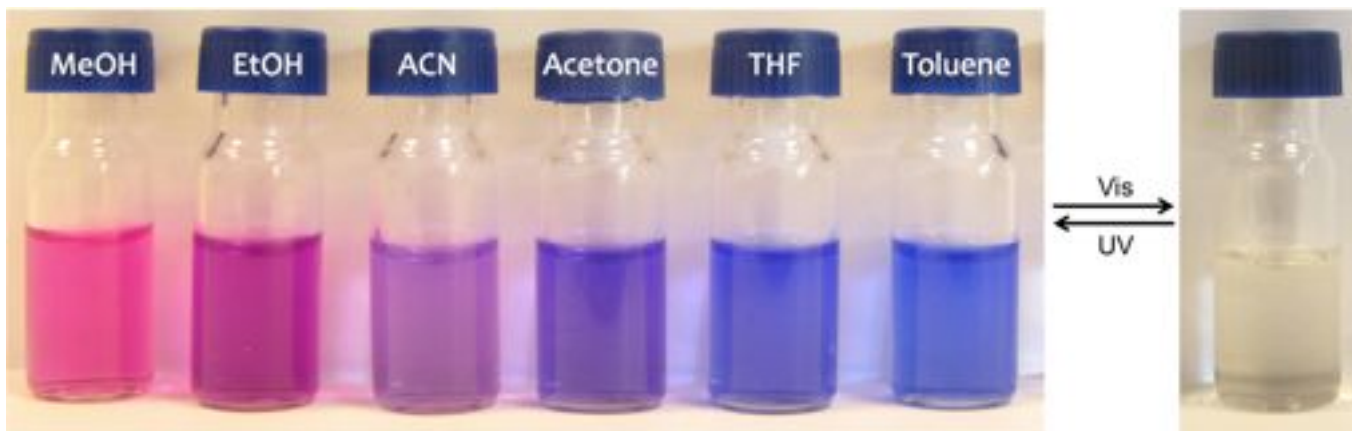




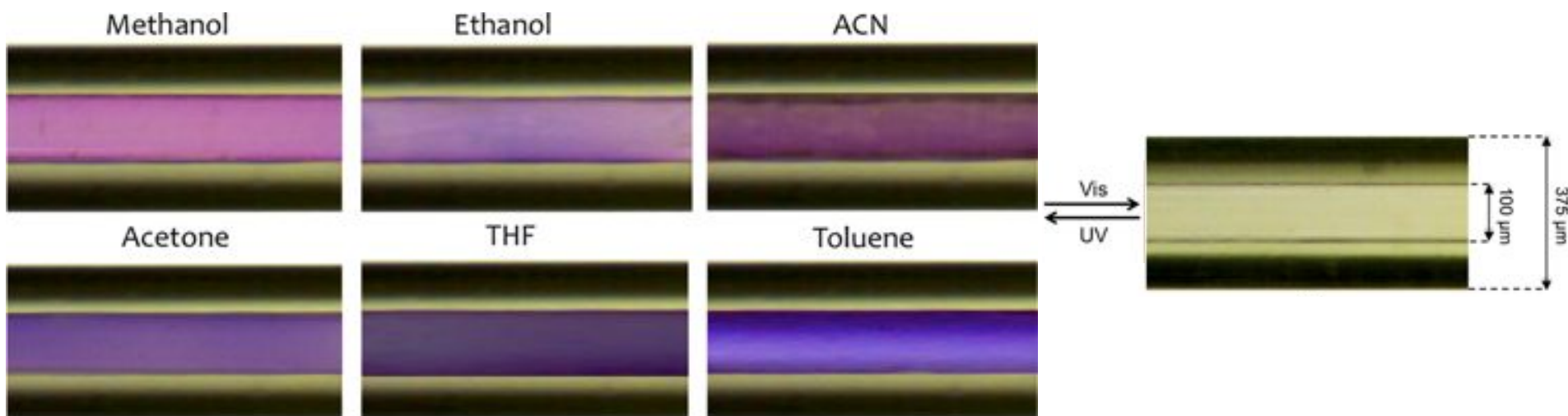
# Solvatochromic Behaviour

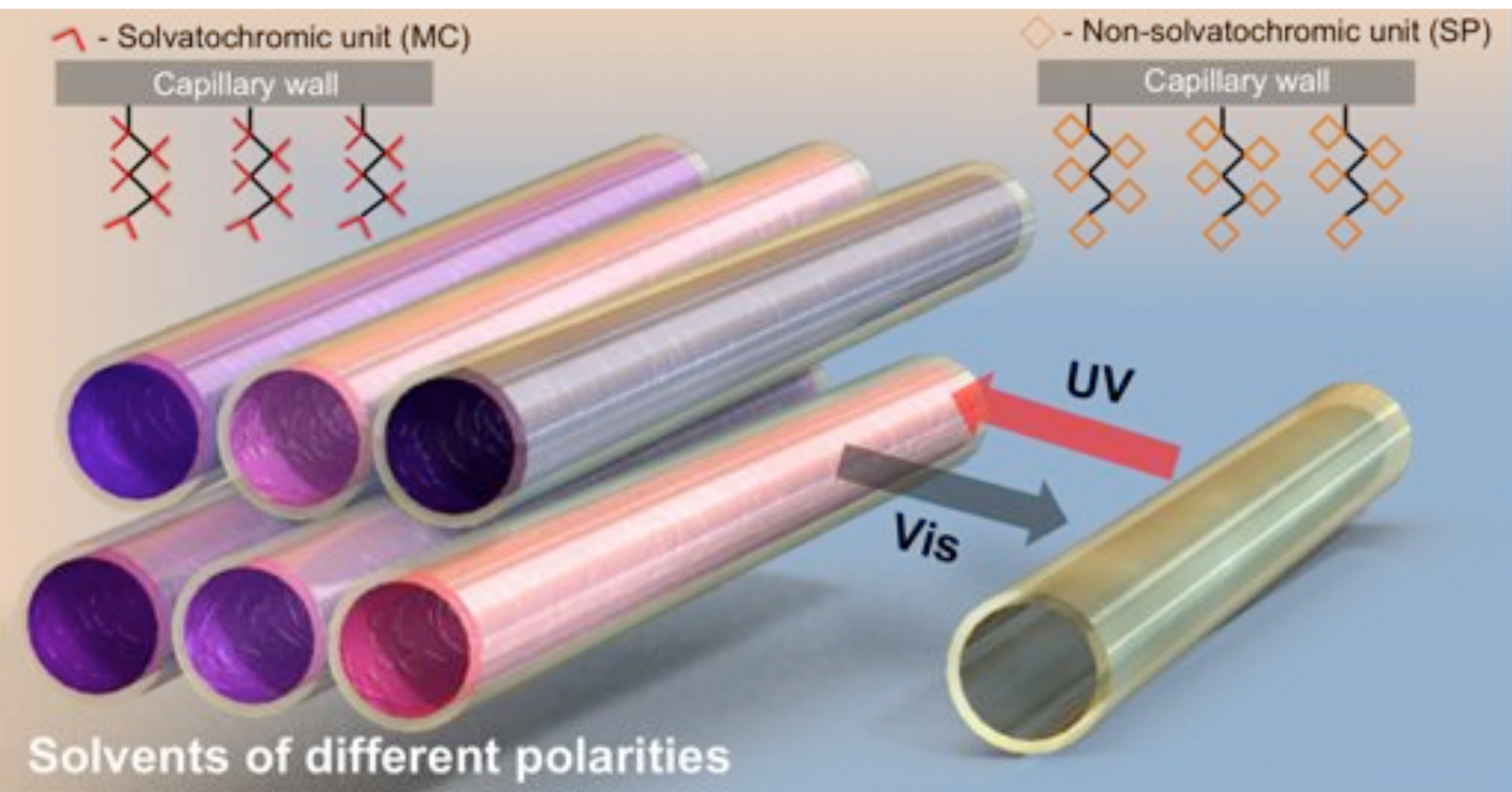


## ➤ In solution



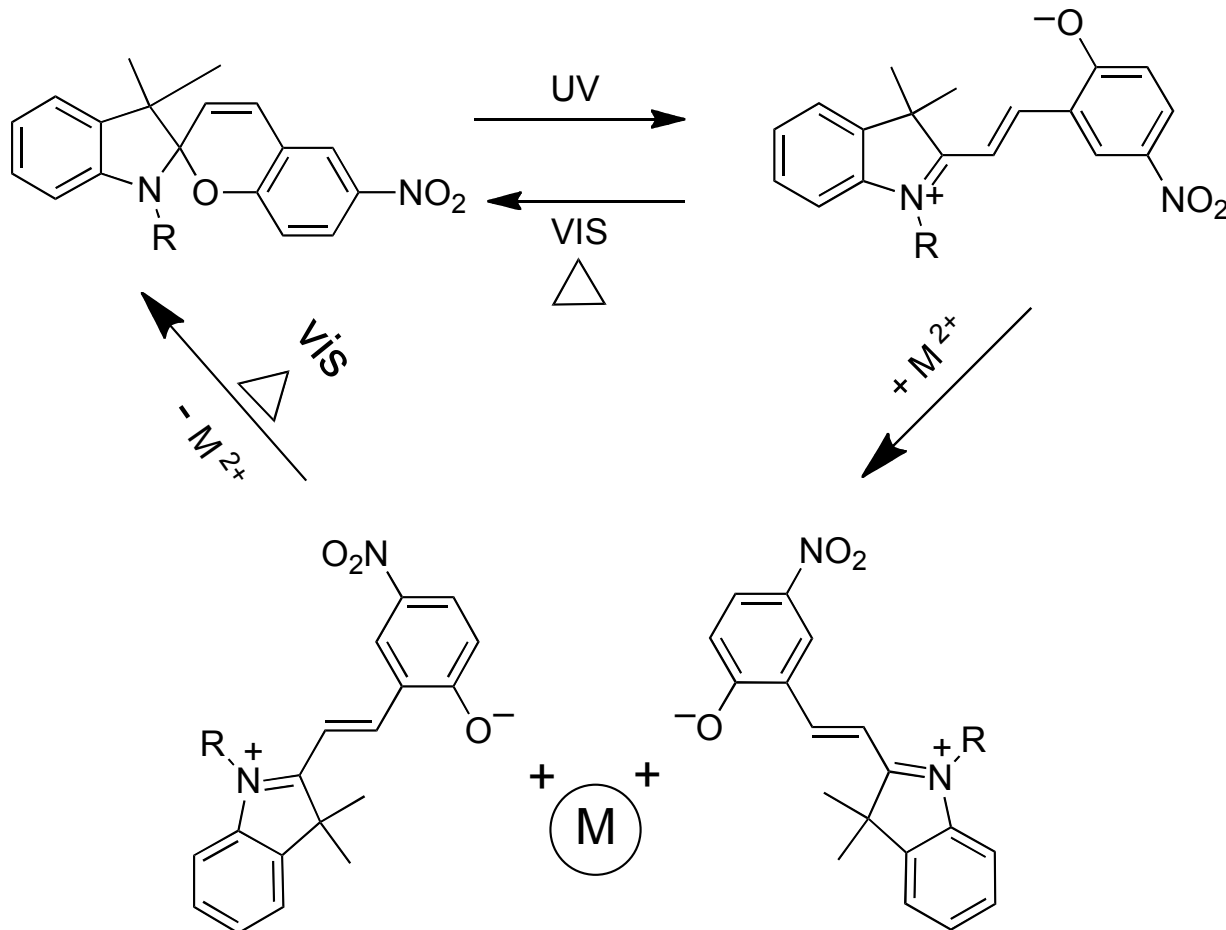
## ➤ Polymeric brushes





L. Florea, A. McKeon, D. Diamond and F. Benito-Lopez, *Langmuir*, 2013, 29, 2790-2797.

# Metal ions sensing, binding and releasing





# Metal ions sensing binding and releasing

## I. Solution studies

**SP-M sol in ACN**

**20 s UV light**

**+ Ni<sup>2+</sup>**

**+ Cd<sup>2+</sup>**

**+ Co<sup>2+</sup>**

**+ Cu<sup>2+</sup>**

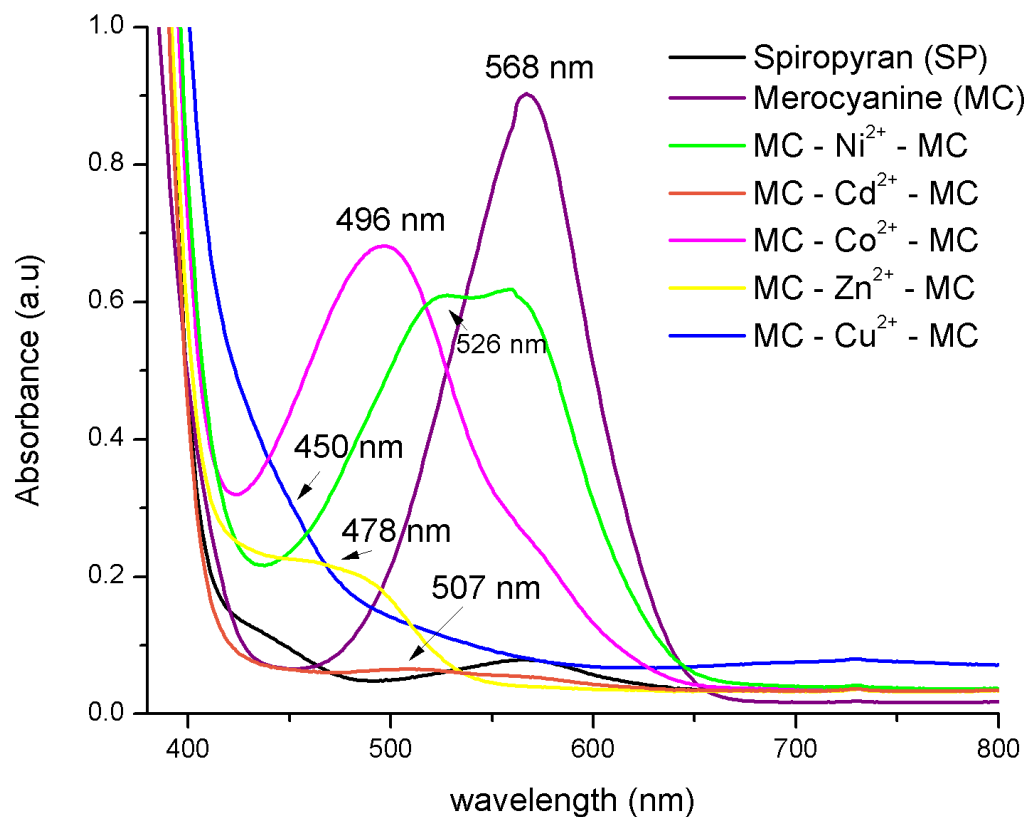
**+ Zn<sup>2+</sup>**





# Metal ions sensing binding and releasing

## I. Solution studies



	Wavelength (nm)
MC	568
MC-Ni <sup>2+</sup>	526
MC-Cd <sup>2+</sup>	507
MC-Co <sup>2+</sup>	496
MC-Zn <sup>2+</sup>	478
MC-Cu <sup>2+</sup>	450

Cu<sup>2+</sup>

Zn<sup>2+</sup>

Co<sup>2+</sup>

Cd<sup>2+</sup>

Ni<sup>2+</sup>

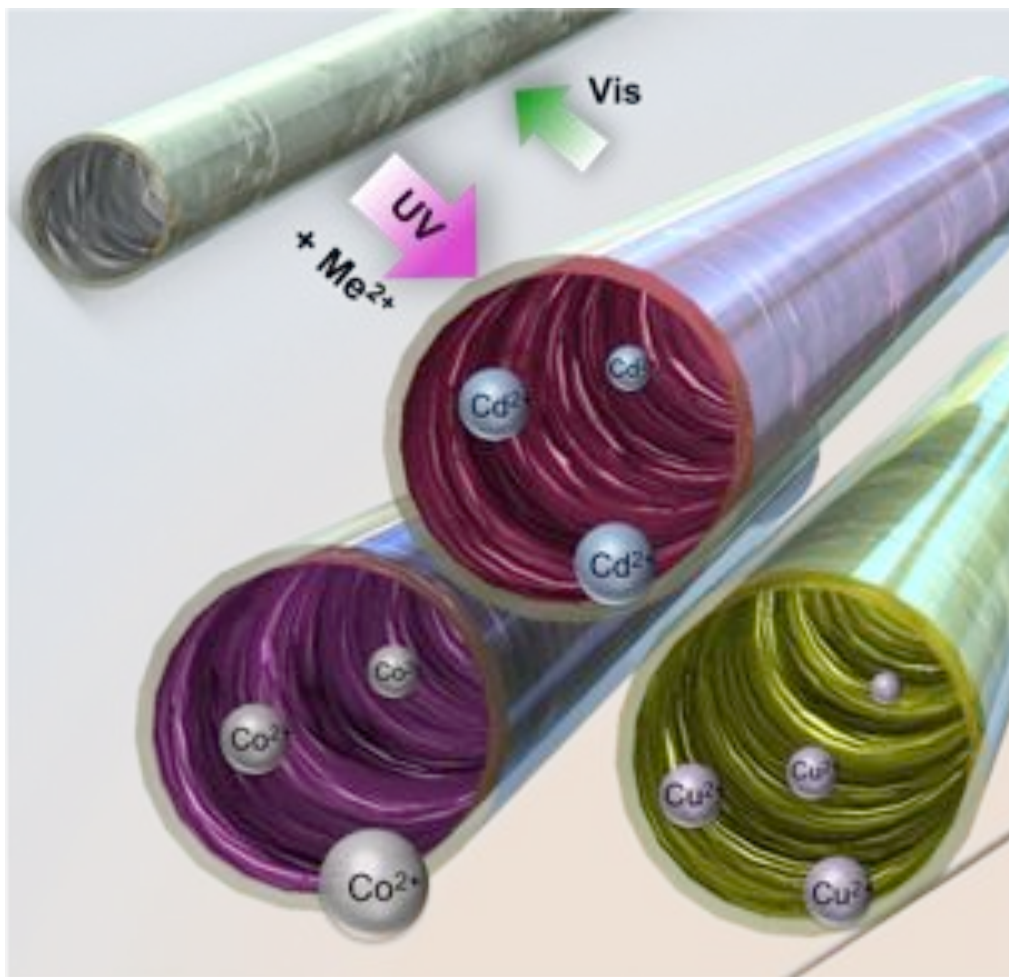
MC





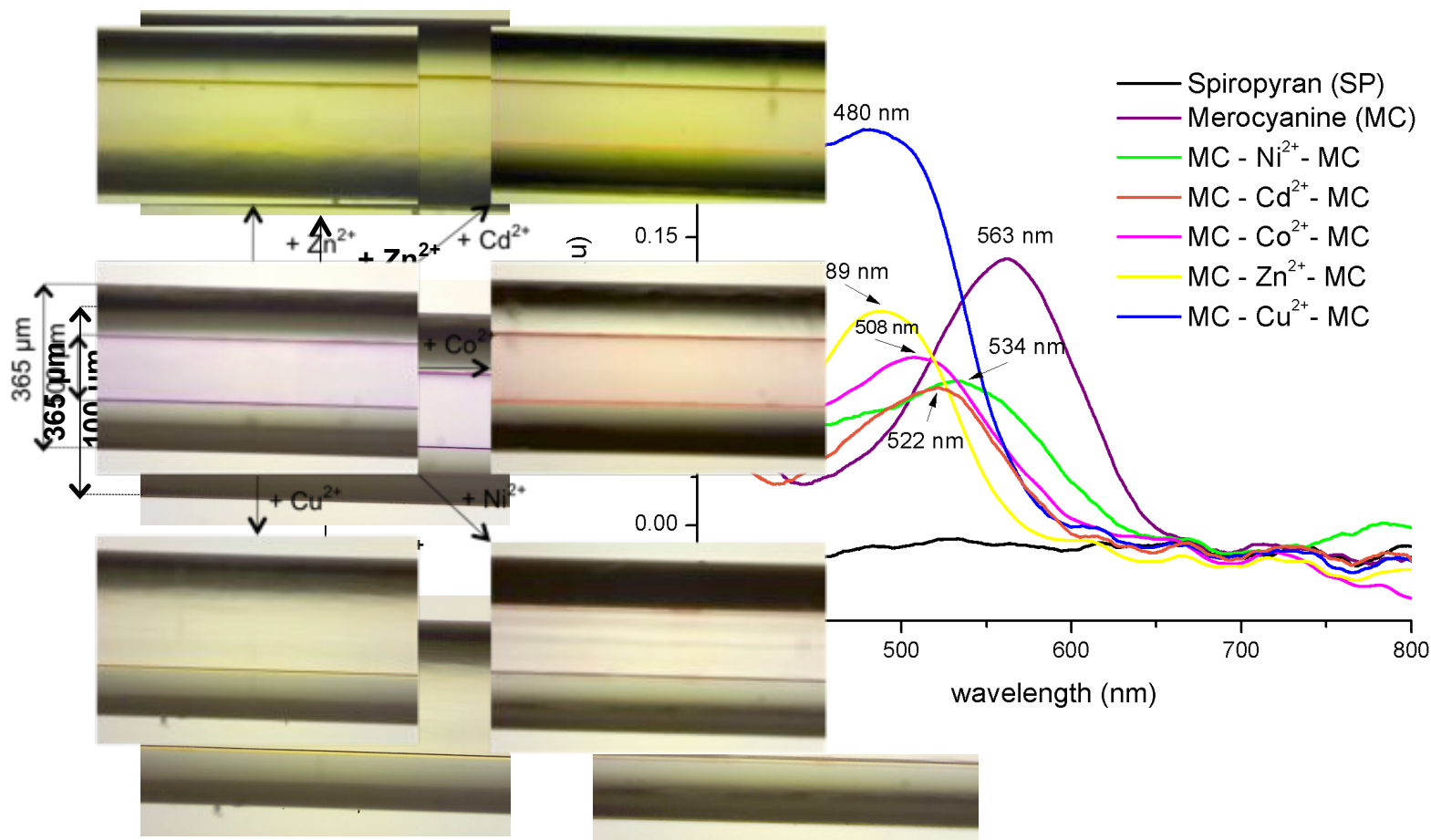


# Metal ions sensing, binding and releasing



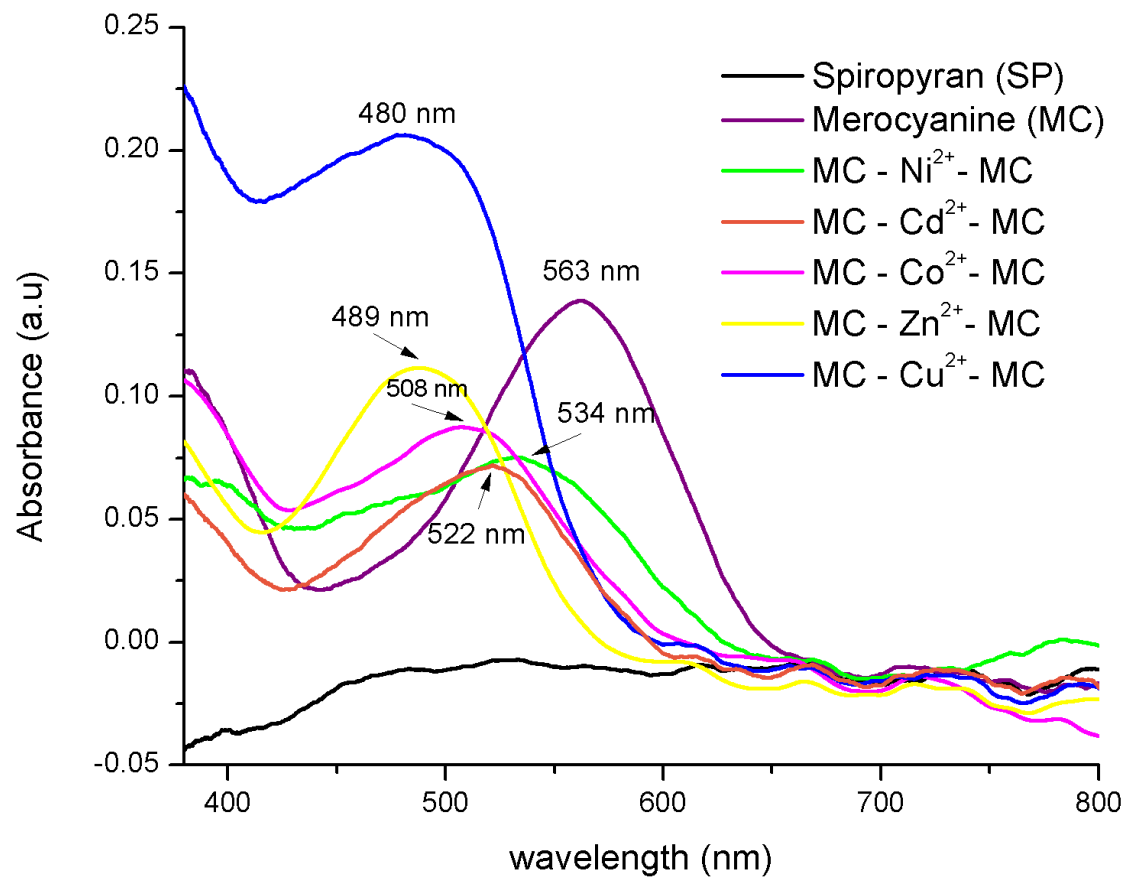
# Metal ions sensing binding and releasing

## II. Capillary coatings



# Metal ions sensing binding and releasing

## II. Capillary coatings



	Wavelength (nm)
MC	563
MC- $\text{Ni}^{2+}$	534
MC- $\text{Cd}^{2+}$	522
MC- $\text{Co}^{2+}$	508
MC- $\text{Zn}^{2+}$	489
MC- $\text{Cu}^{2+}$	480

$\text{Cu}^{2+}$

$\text{Zn}^{2+}$

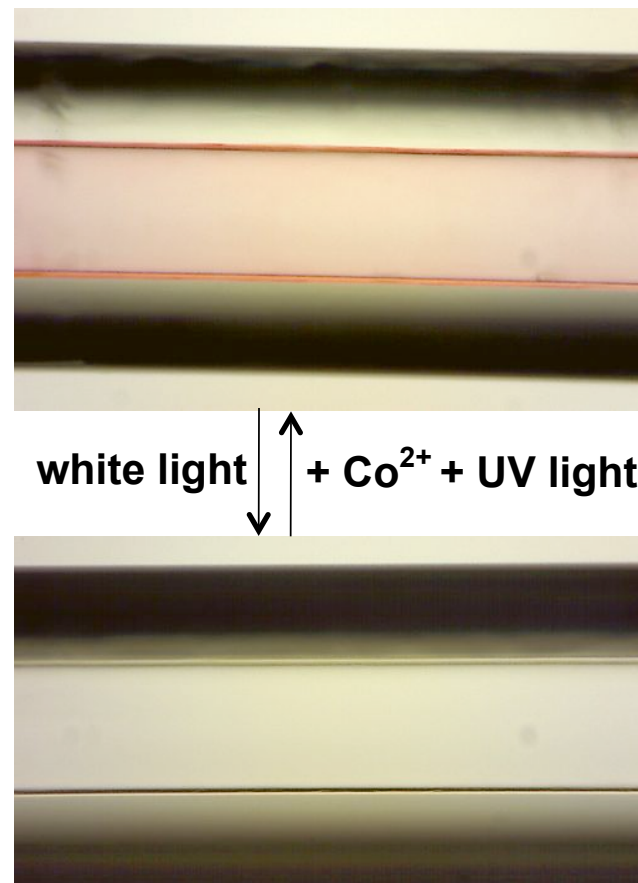
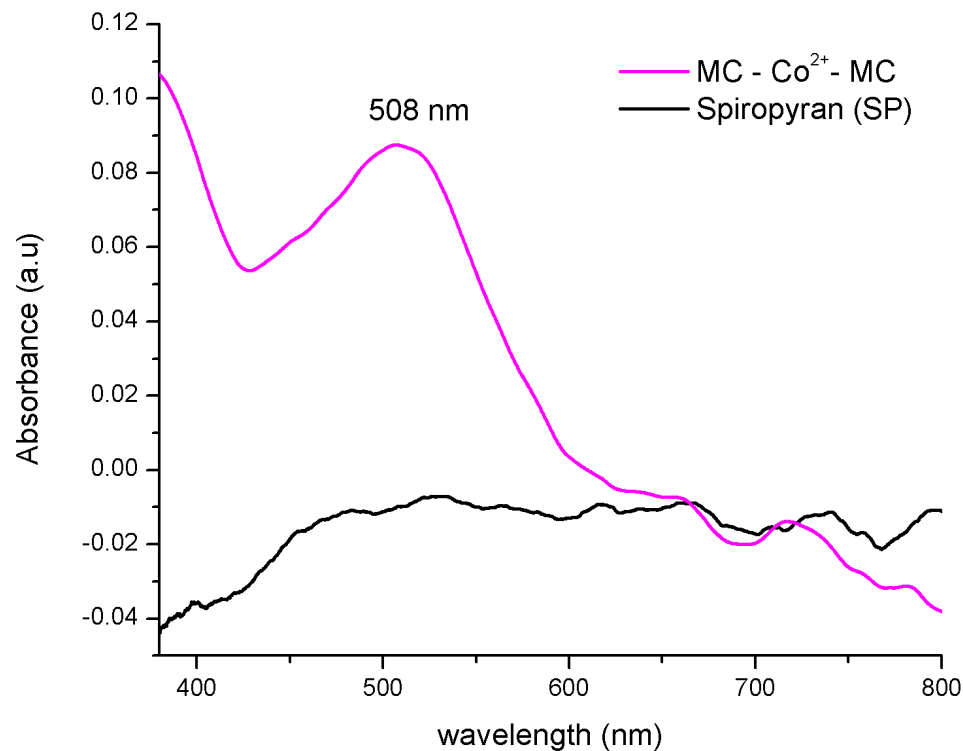
$\text{Co}^{2+}$

$\text{Cd}^{2+}$

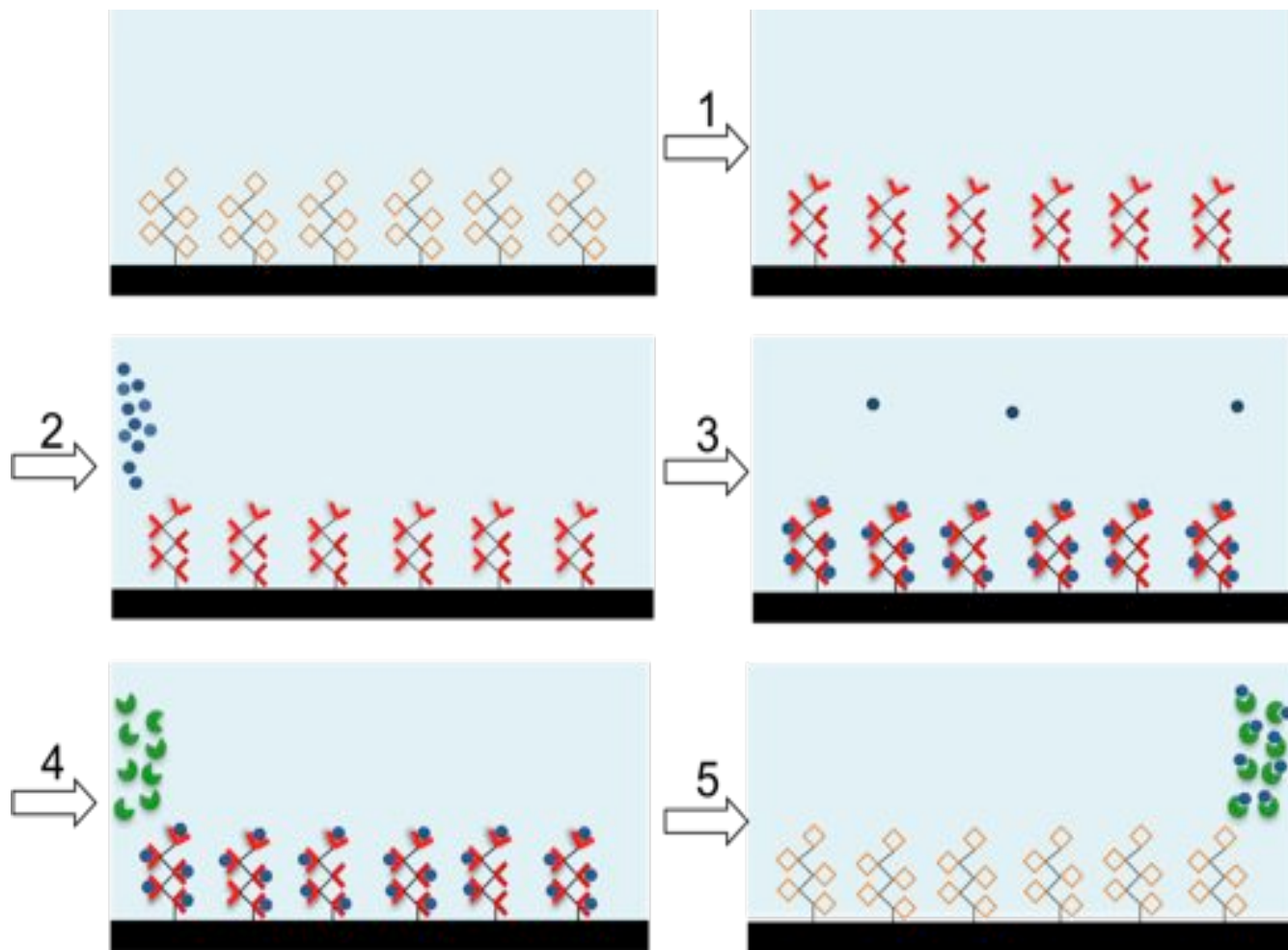
$\text{Ni}^{2+}$





MC

# Metal ions binding and releasing



# Quantitative binding



-  Non binding form
-  Binding form
-   $\text{Cu}^{2+}$
-  PADAP

**Binding capacity:**  
 $\sim 7 \times 10^{-8} \text{ Cu}^{2+} \text{ mole/}$   
 $\text{mm}^3 \text{ coating.}$



- **Self-diagnostic for continuous flow device**
- **Solvent detection and divalent metal ion detection in micro-capillaries**
- **Sensing behaviour can be switched on/off remotely using light**







# Acknowledgments



- **Dr. Cormac Fay**
- **Wayne Francis**
- **Aishling Dunne**
- **Dr. Fernando Benito-Lopez**
- **Prof. Gordon Wallace**
- **Prof. Dermot Diamond**
- **Adaptive Sensors Group**
- **Insight SFI award**





# Thank you!